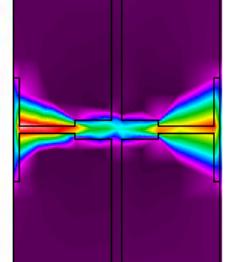
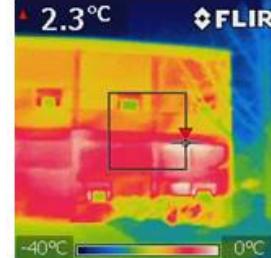


CAE 331/513

Building Science

Fall 2019



October 10, 2019

Psychrometric processes (part 2)

Built
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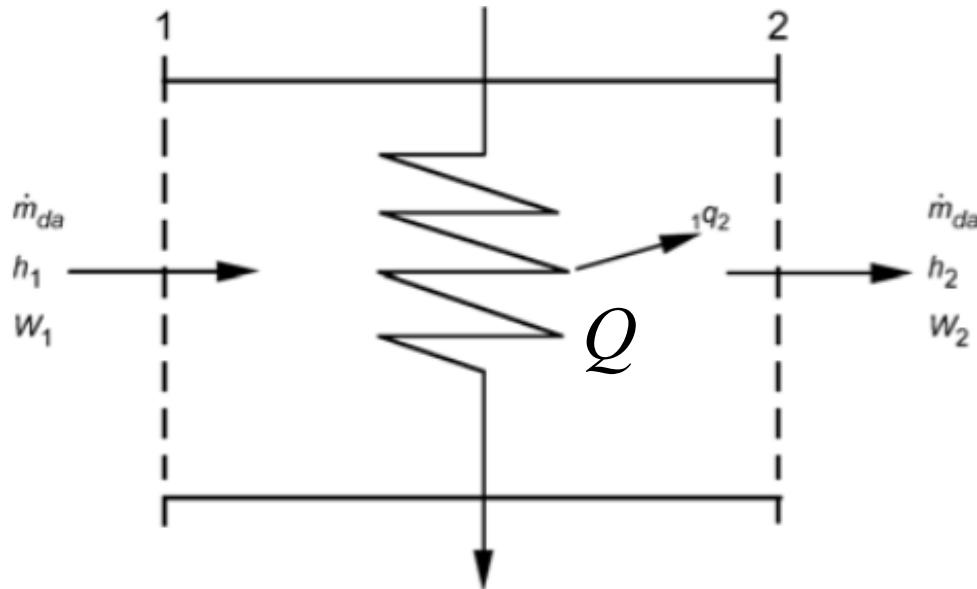
brent@iit.edu

PSYCHROMETRIC PROCESSES

Using energy and mass balance equations

Energy/mass balances for psychrometric processes

Heating air



$$\text{Energy balance: } \dot{m}_{da,1} h_1 + Q_{1 \rightarrow 2} = \dot{m}_{da,2} h_2$$

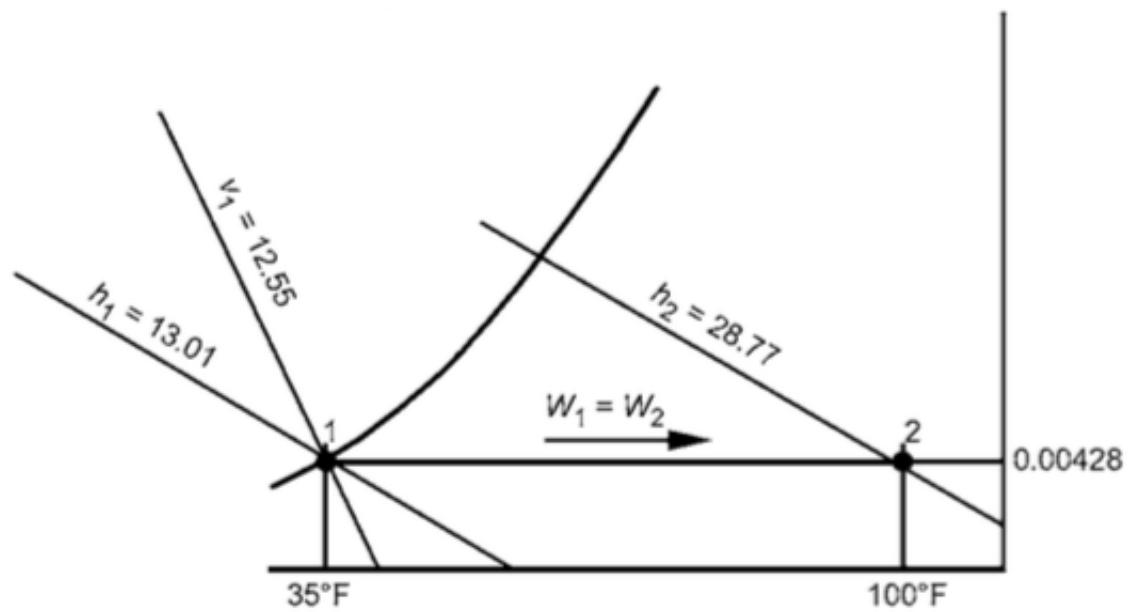
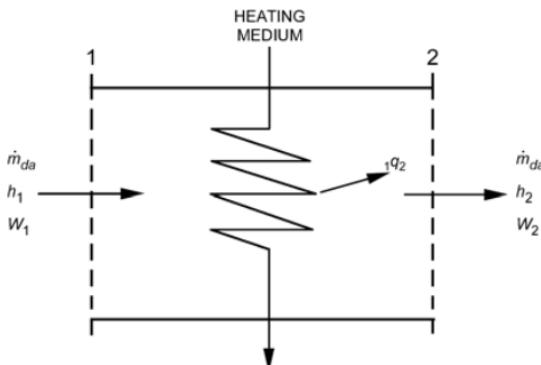
$$\text{Mass balance on air: } \dot{m}_{da,1} = \dot{m}_{da,2} = \dot{m}_{da}$$

$$\text{Mass balance on water vapor: } \dot{m}_{da,1} W_1 = \dot{m}_{da,2} W_2$$

$$\text{Therefore: } Q_{1 \rightarrow 2} = \dot{m}_{da} (h_2 - h_1)$$

Example 1: Heating (IP)

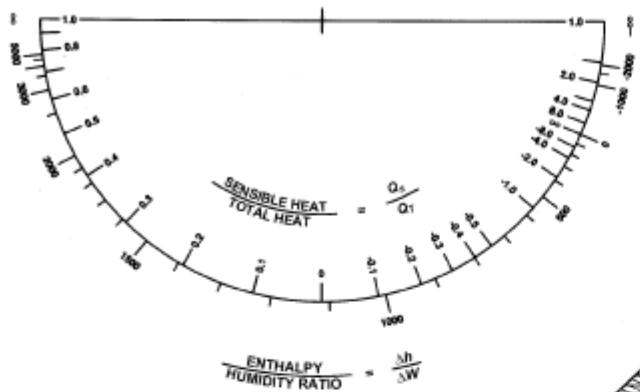
- Moist air, saturated at 35°F, enters a heating coil at a rate of 20,000 CFM and air leaves the coil at 100°F
 - Find the required rate of heat addition



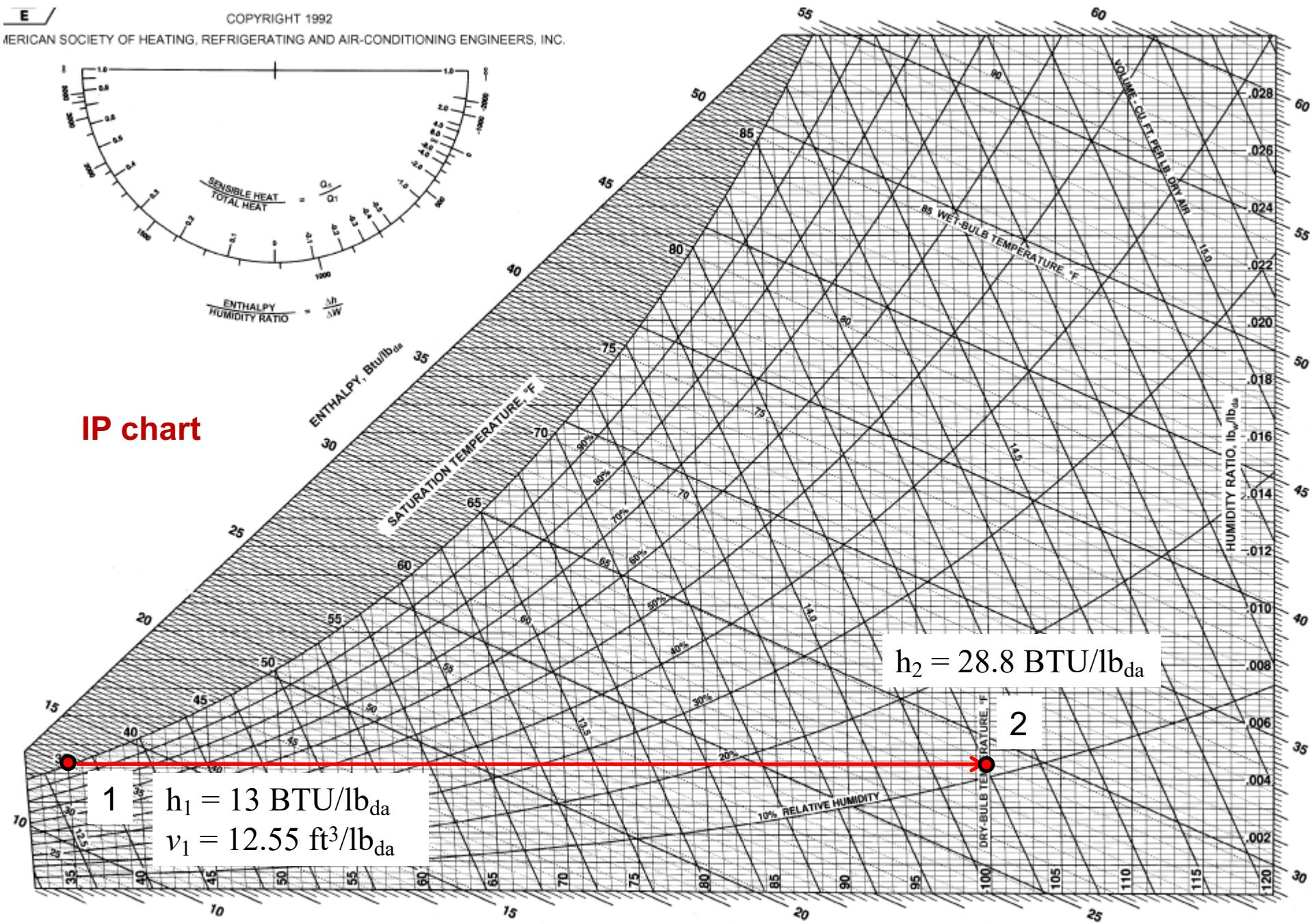
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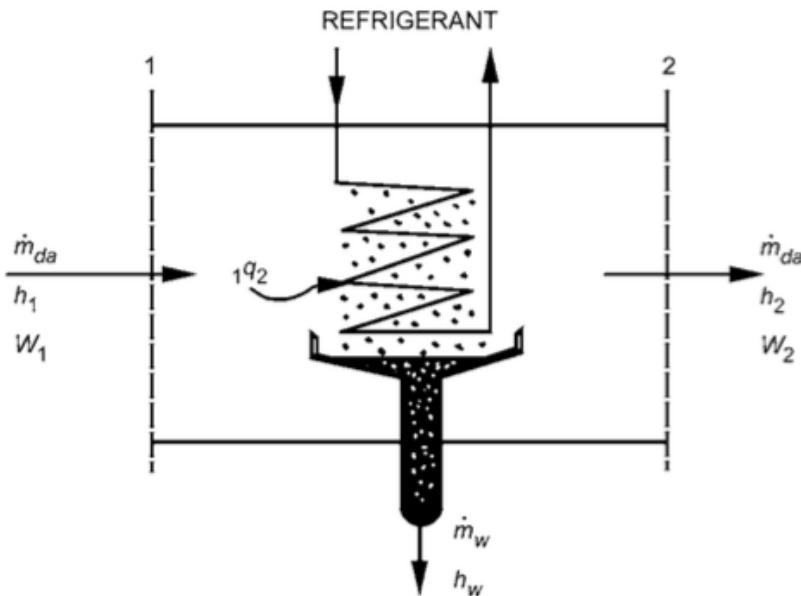
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IP chart



Energy/mass balances for psychrometric processes



Cooling and dehumidifying

*Note that $h_w = h_g$ for steam/vapor
and $h_w = h_f$ for water

$$\text{Energy balance: } \dot{m}_{da,1}h_1 + Q_{1 \rightarrow 2} = \dot{m}_{da,2}h_2 + \dot{m}_w h_{w,2}$$

$$\text{Mass balance on air: } \dot{m}_{da,1} = \dot{m}_{da,2} = \dot{m}_{da}$$

$$\text{Mass balance on water vapor: } \dot{m}_{da,1}W_1 = \dot{m}_{da,2}W_2 + \dot{m}_w$$

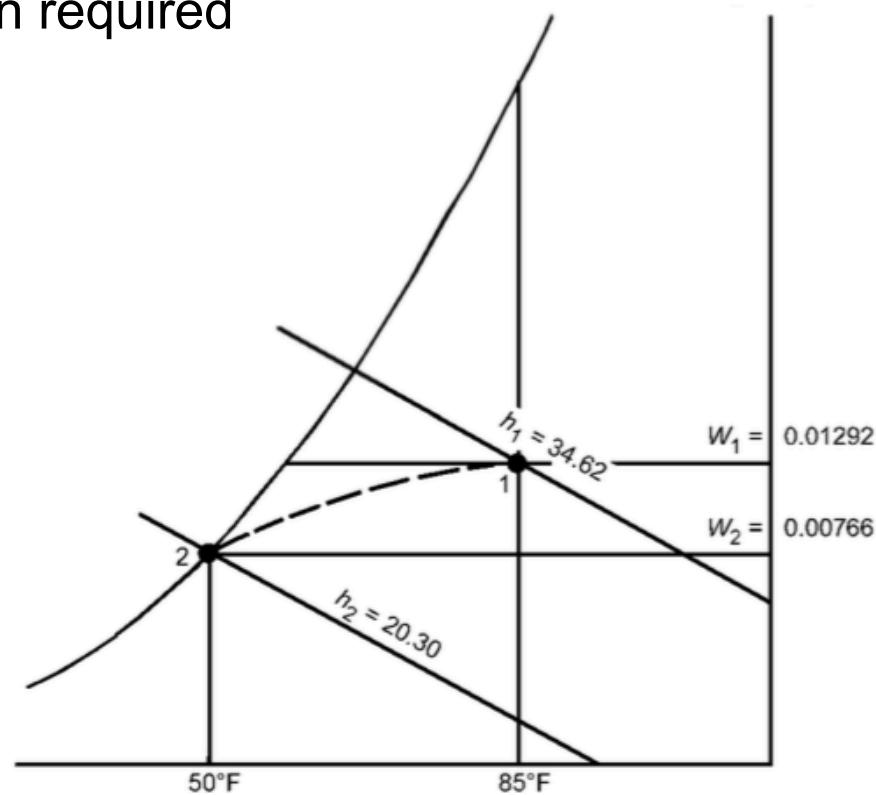
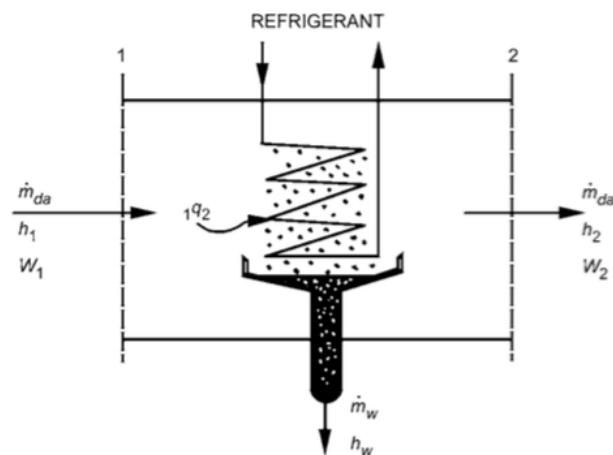
$$\text{Therefore: } \dot{m}_w = \dot{m}_{da}(W_1 - W_2)$$

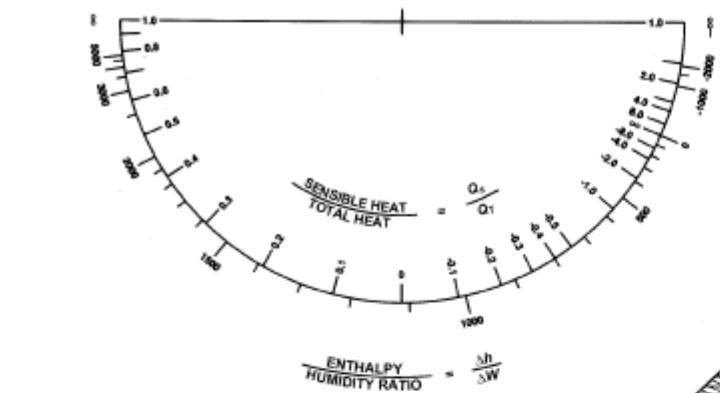
$$\text{And: } Q_{1 \rightarrow 2} = \dot{m}_{da}[(h_2 - h_1) - (W_2 - W_1)h_{w,2}]$$

(Q is negative for cooling)

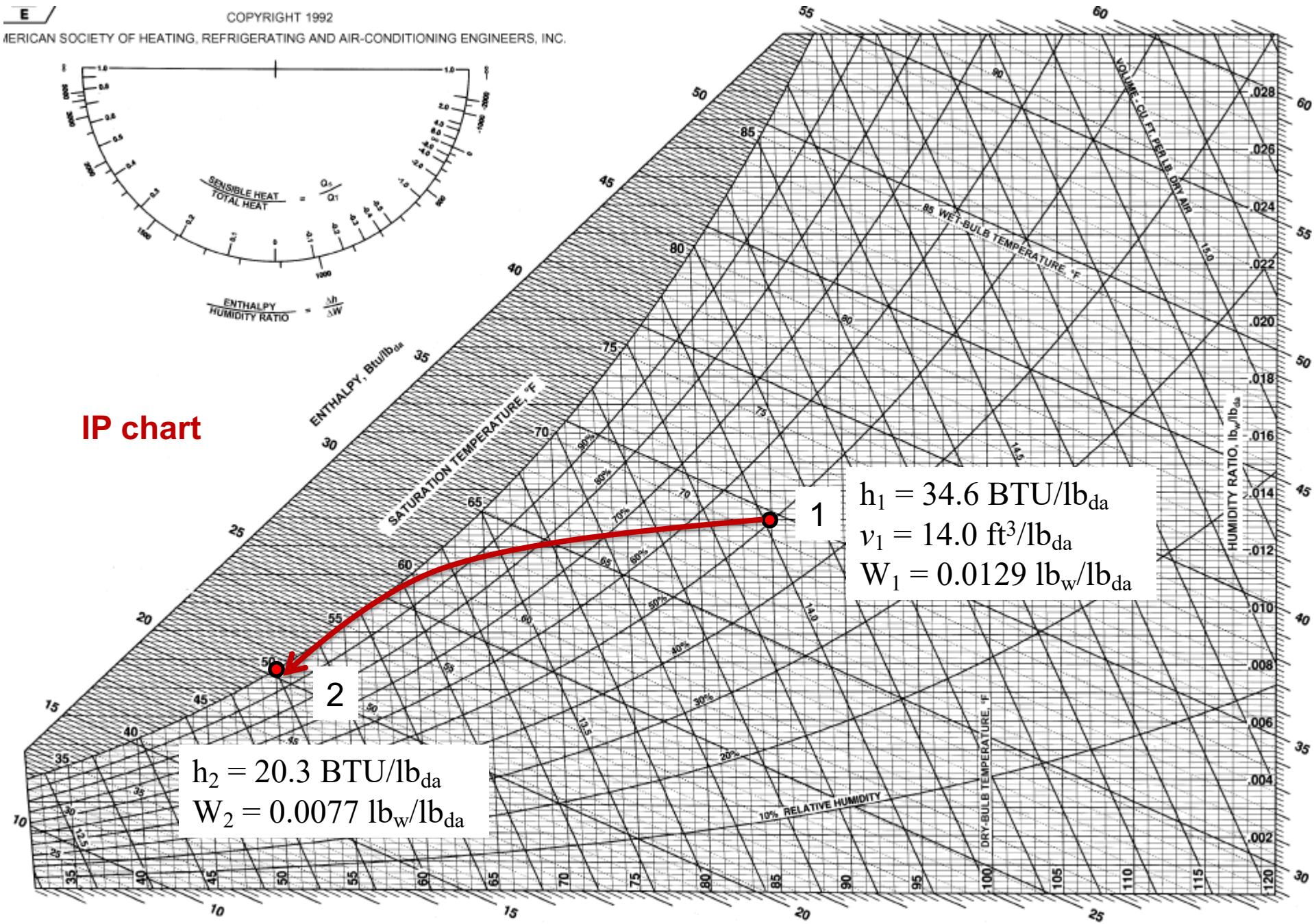
Example 2: Cooling and dehumidification (IP)

- Moist air at 85°F dry-bulb temperature and 50% RH enters a cooling coil at 10,000 CFM and is processed to final saturation conditions at 50°F
 - Find the tons of refrigeration required



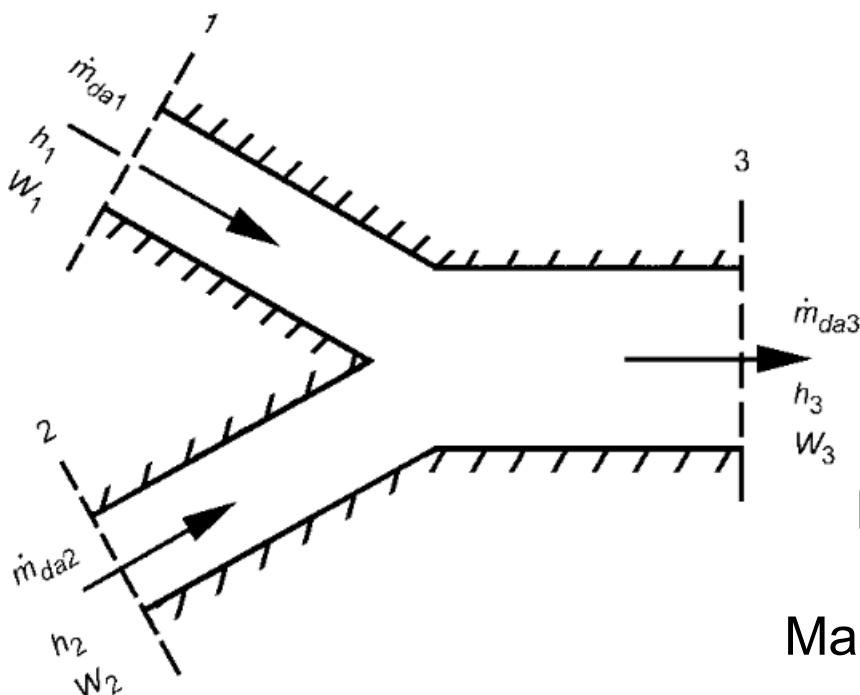


IP chart



Energy/mass balances for psychrometric processes

- **Mixing:** Often in HVAC systems we mix airstreams adiabatically
 - **Adiabatically** = Without the addition or extraction of heat
 - e.g. outdoor air mixed with a portion of return/recirculated air



$$\text{Energy: } \dot{m}_{da,1}h_1 + \dot{m}_{da,2}h_2 = \dot{m}_{da,3}h_3$$

$$\text{Mass (air): } \dot{m}_{da,1} + \dot{m}_{da,2} = \dot{m}_{da,3}$$

$$\text{Mass (water): } \dot{m}_{da,1}W_1 + \dot{m}_{da,2}W_2 = \dot{m}_{da,3}W_3$$

Energy/mass balances for psychrometric processes

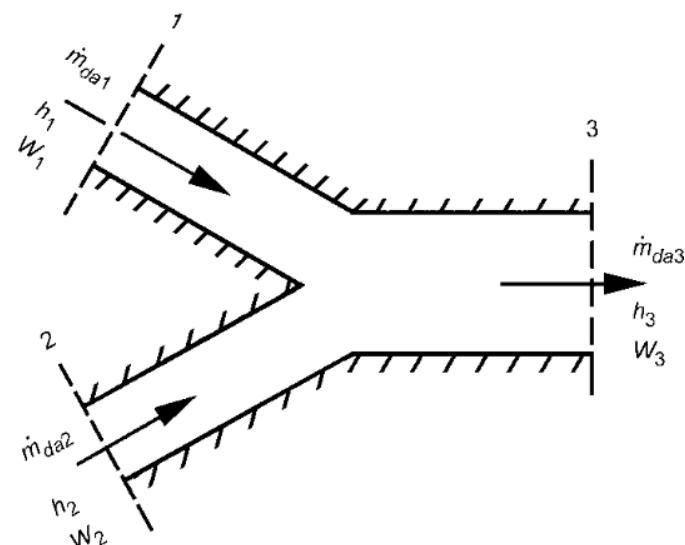
- **Mixing:** For most parameters, the outlet conditions end up being the weighted averages of the input conditions based on their mass flow rates

- Dry bulb temperature
- Humidity ratio
- Enthalpy
- (not RH!)

$$T_3 = \frac{\dot{m}_{da1}T_1 + \dot{m}_{da2}T_2}{\dot{m}_{da1} + \dot{m}_{da2}}$$

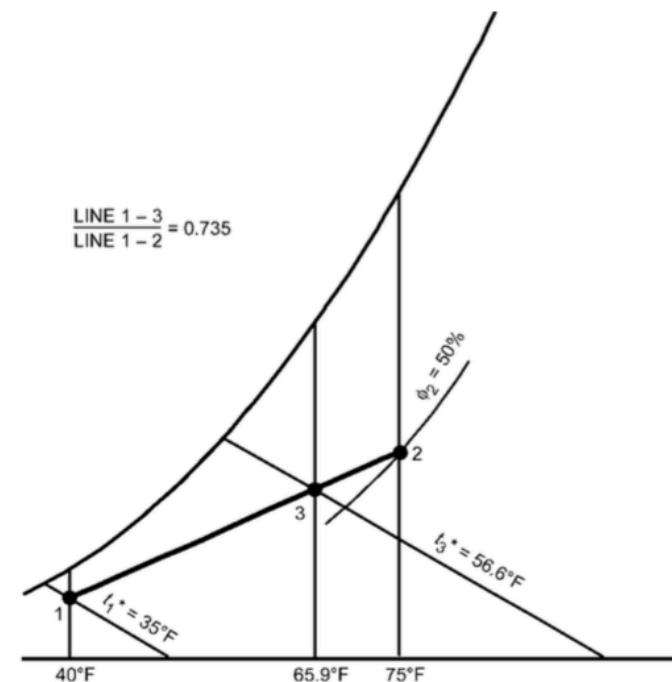
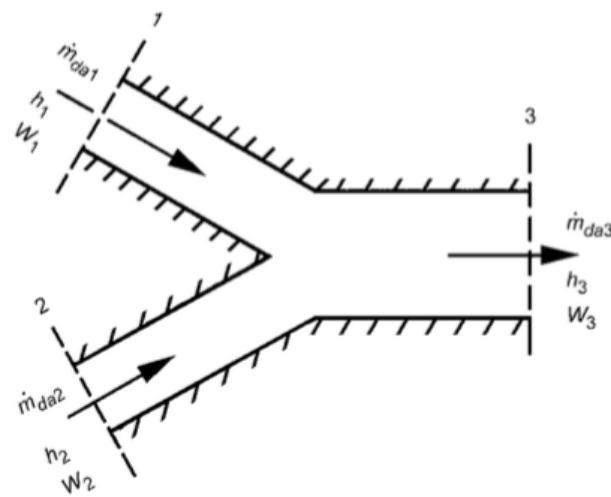
$$W_3 = \frac{\dot{m}_{da1}W_1 + \dot{m}_{da2}W_2}{\dot{m}_{da1} + \dot{m}_{da2}}$$

$$h_3 = \frac{\dot{m}_{da1}h_1 + \dot{m}_{da2}h_2}{\dot{m}_{da1} + \dot{m}_{da2}}$$



Example 3: Mixing (IP)

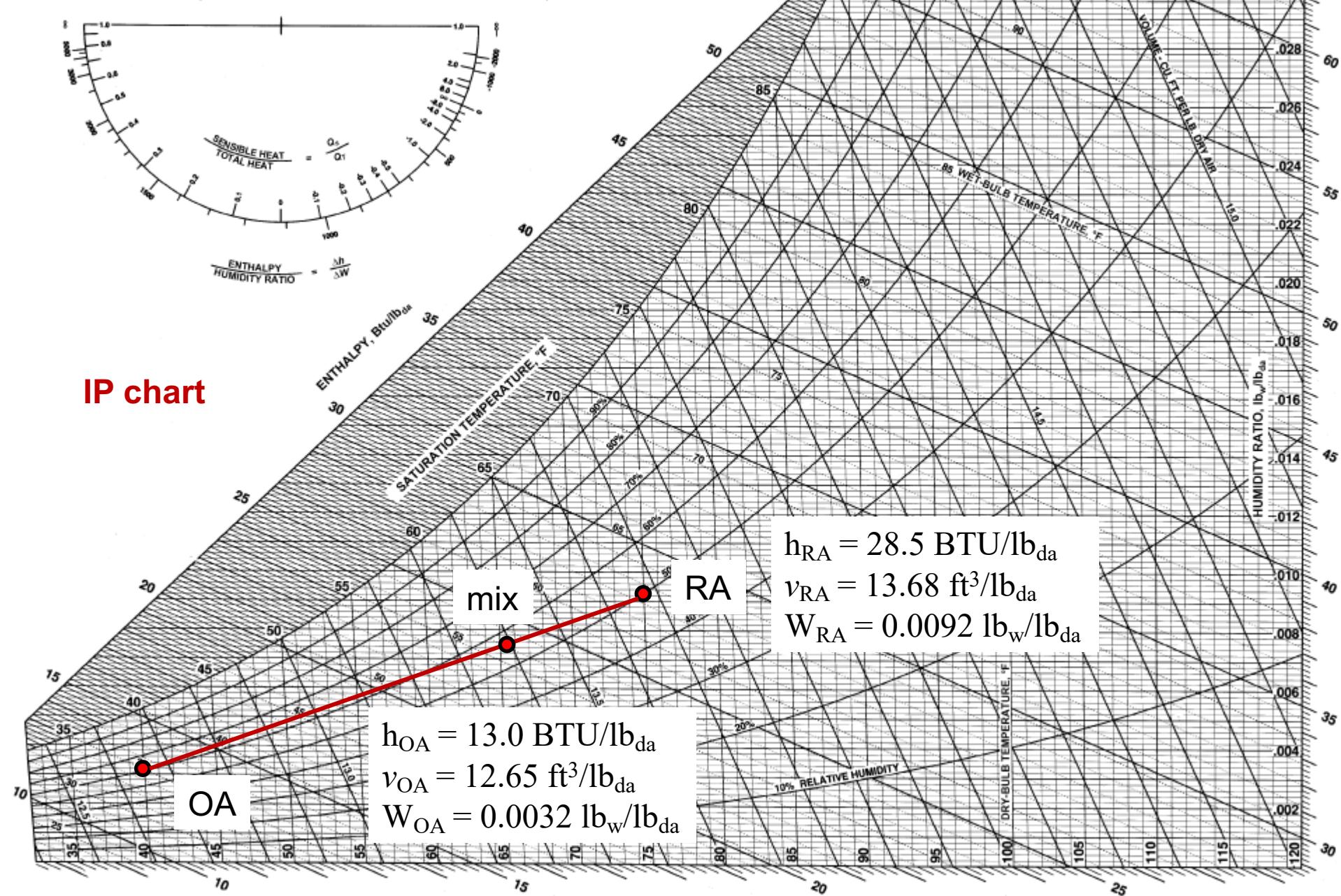
- A stream of 5,000 CFM of outdoor air at 40°F dry bulb temperature and 35°F thermodynamic wet bulb temperature is adiabatically mixed with 15,000 CFM of recirculated air at 75°F dry bulb temperature and 50% RH
 - Find the dry bulb temperature, humidity ratio, and enthalpy of the resulting mixture



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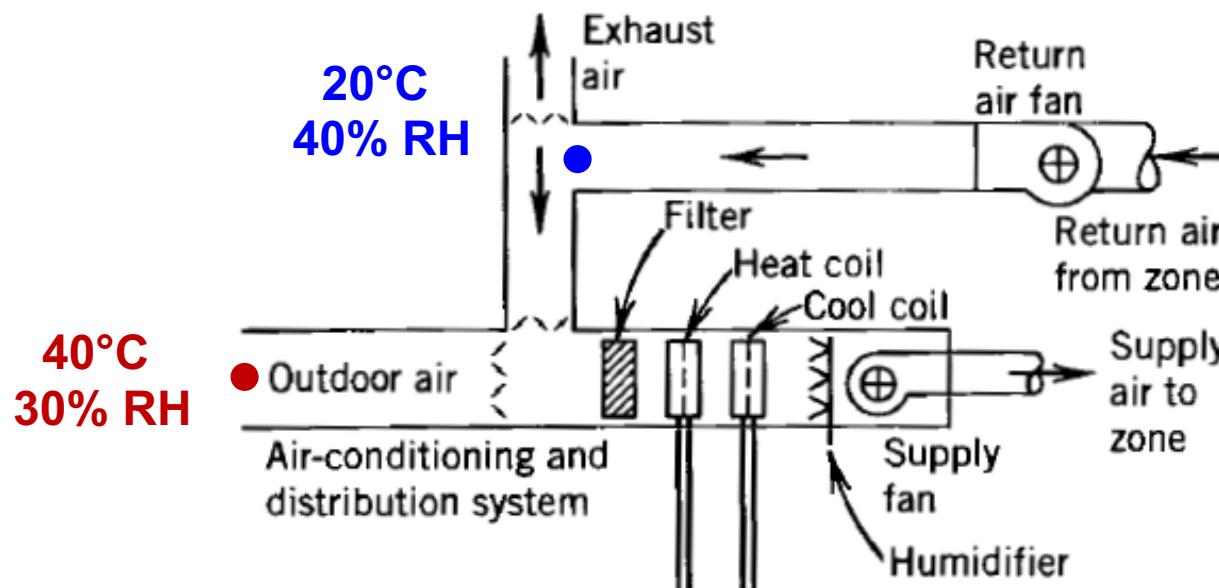
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Example 4: Mixing (SI)

- Hot, humid outdoor air is mixed with recirculated indoor air at an outdoor air mass flow fraction of 35%
 - Q1: What is T, W, RH, and h at the mixed condition?





ASHRAE PSYCHROMETRIC CHART NO.1

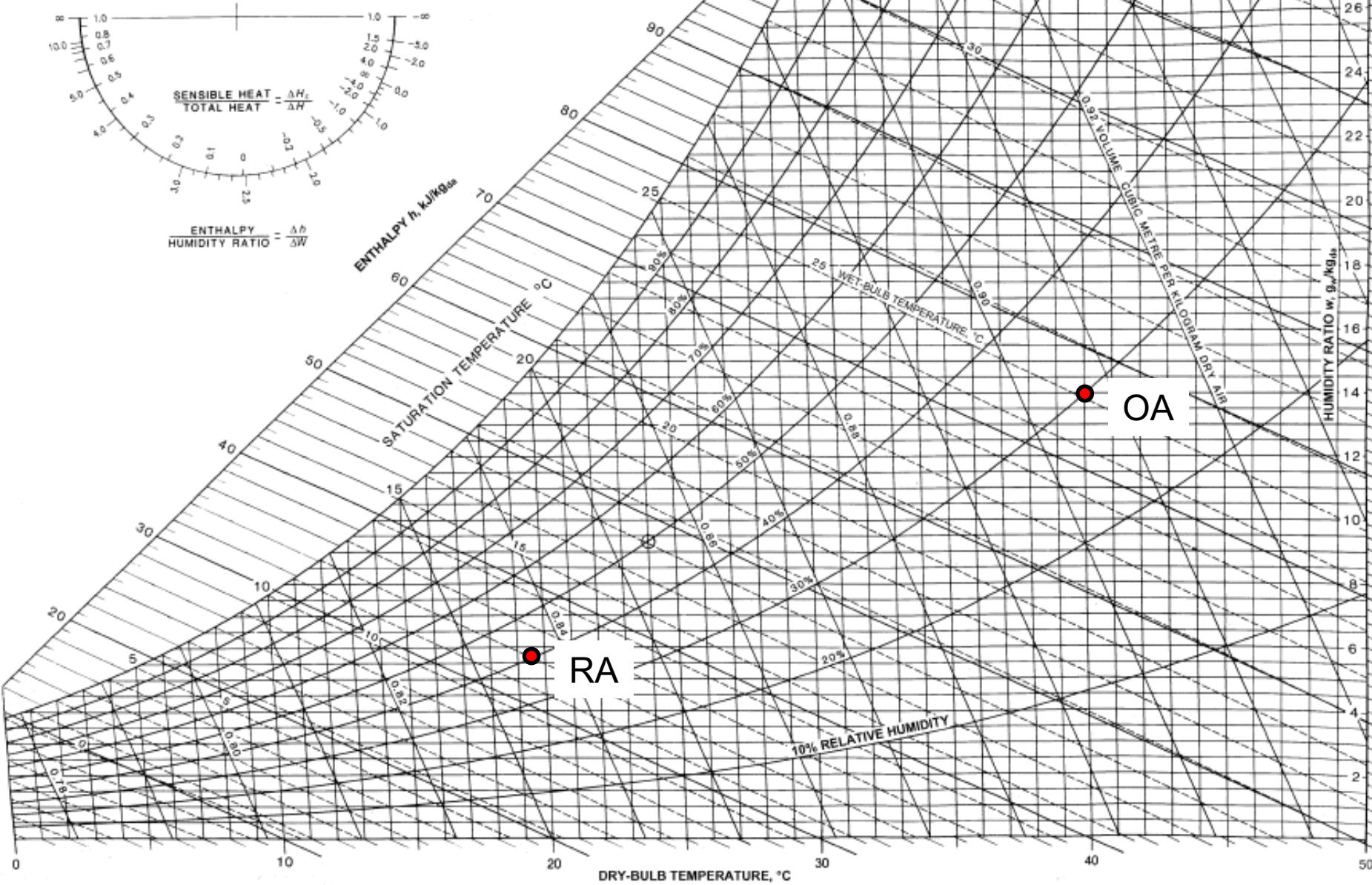
NORMAL TEMPERATURE

SEA LEVEL

BAROMETRIC PRESSURE: 101.325 kPa

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Mixing of airstreams



ASHRAE PSYCHROMETRIC CHART NO.1

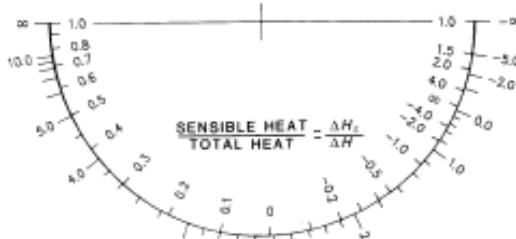
NORMAL TEMPERATURE

SEA LEVEL

BAROMETRIC PRESSURE: 101.325 kPa

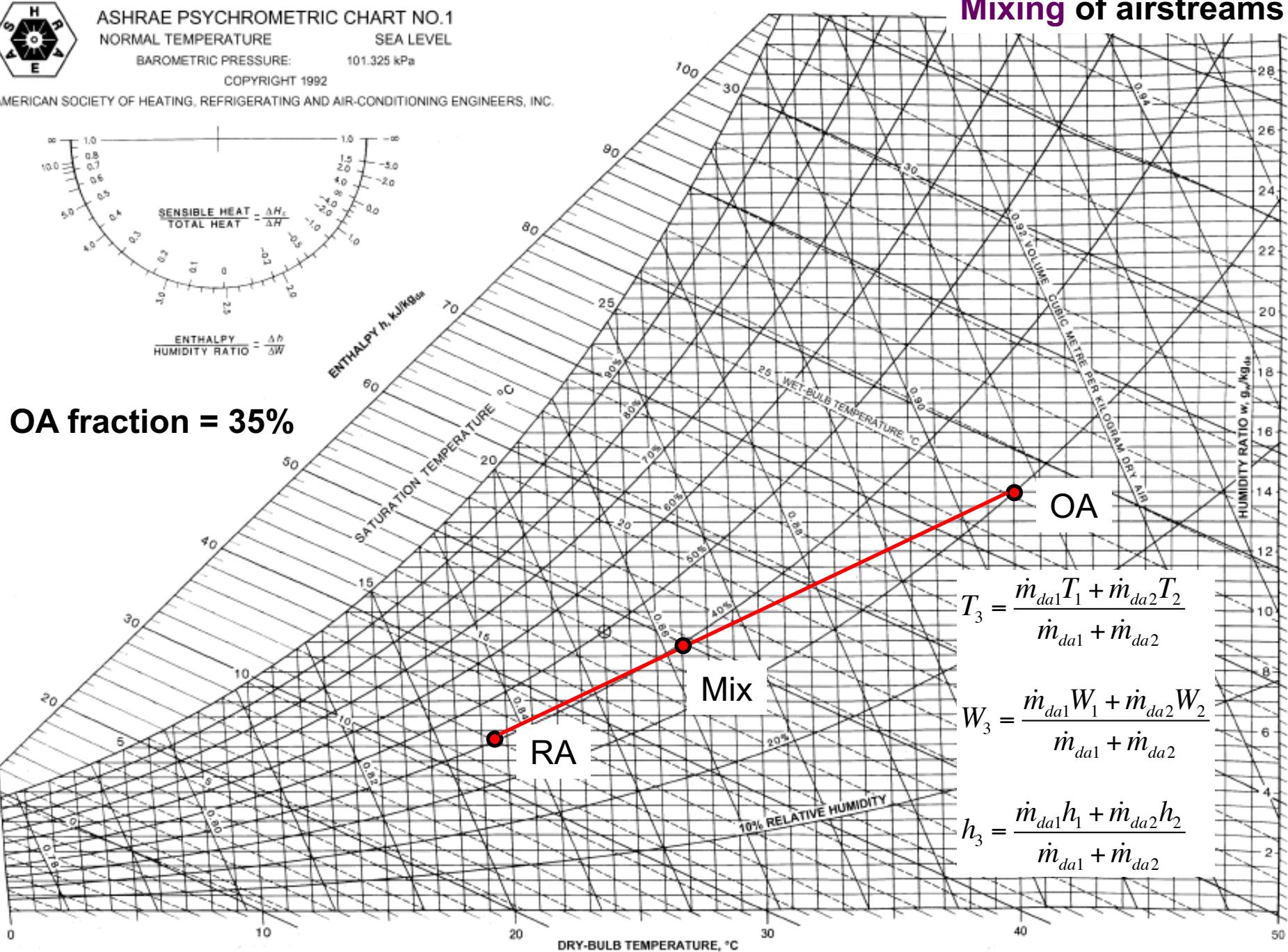
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$$\text{ENTHALPY } h, \text{ kJ/kg} = \frac{\Delta h}{\Delta W}$$

OA fraction = 35%



$$T_3 = \frac{\dot{m}_{da1}T_1 + \dot{m}_{da2}T_2}{\dot{m}_{da1} + \dot{m}_{da2}}$$

$$W_3 = \frac{\dot{m}_{da1}W_1 + \dot{m}_{da2}W_2}{\dot{m}_{da1} + \dot{m}_{da2}}$$

$$h_3 = \frac{\dot{m}_{da1}h_1 + \dot{m}_{da2}h_2}{\dot{m}_{da1} + \dot{m}_{da2}}$$



ASHRAE PSYCHROMETRIC CHART NO.1

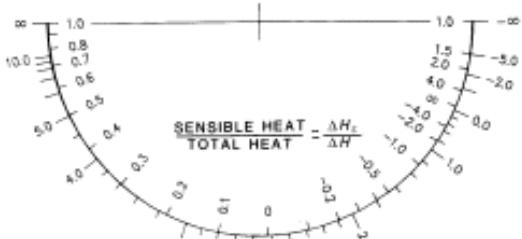
NORMAL TEMPERATURE

SEA LEVEL

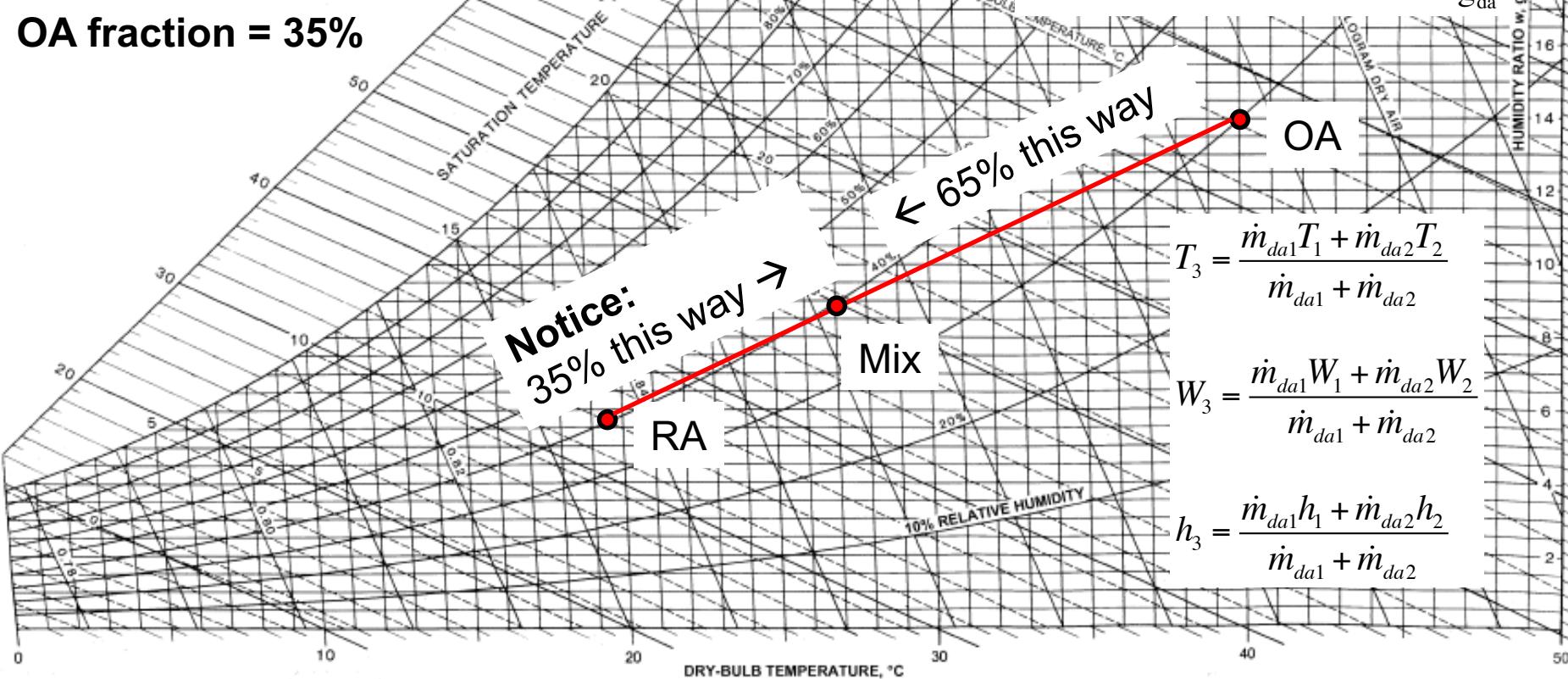
BAROMETRIC PRESSURE: 101.325 kPa

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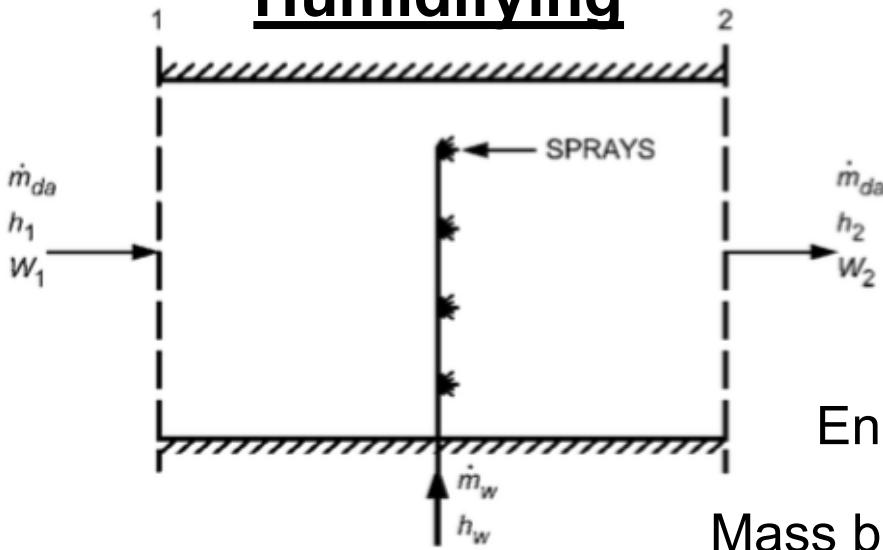


OA fraction = 35%



Energy/mass balances for psychrometric processes

Humidifying



Energy balance:

$$\dot{m}_{da,1}h_1 + \dot{m}_w h_w = \dot{m}_{da,2}h_2$$

Mass balance on air:

$$\dot{m}_{da,1} = \dot{m}_{da,2} = \dot{m}_{da}$$

Mass balance on water vapor:

$$\dot{m}_{da,1}W_1 + \dot{m}_w = \dot{m}_{da,2}W_2$$

Therefore:

$$\dot{m}_w = \dot{m}_{da}(W_2 - W_1)$$

And:

$$\dot{m}_w h_w = \dot{m}_{da}(h_2 - h_1)$$

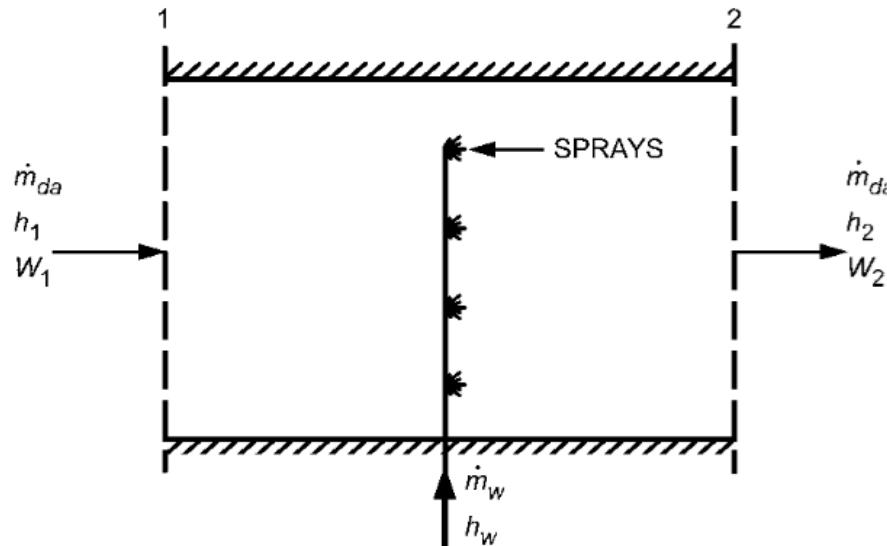
And:

$$\frac{h_2 - h_1}{W_2 - W_1} = \frac{\Delta h}{\Delta W} = h_w$$

*Note that $h_w = h_g$ for steam/vapor
and $h_w = h_f$ for water

Example 5: Humidification (IP)

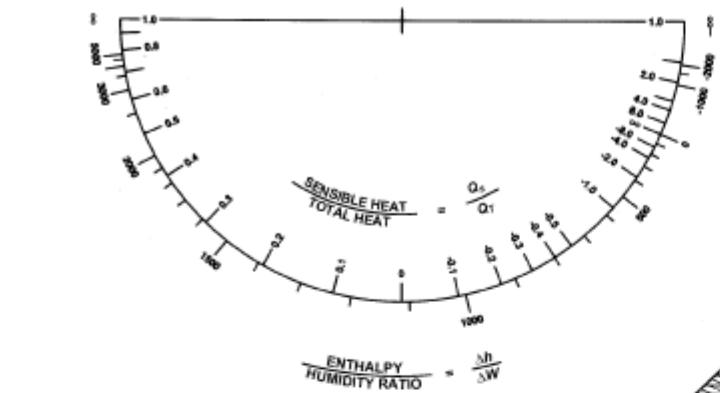
- Moist air at 70°F dry bulb temperature and 45°F wet bulb temperature is to be processed to a final dew point of 55°F by adiabatic injection of saturated steam at 230°F
- The air flow rate is 10,000 CFM
 - Find the rate of steam flow required



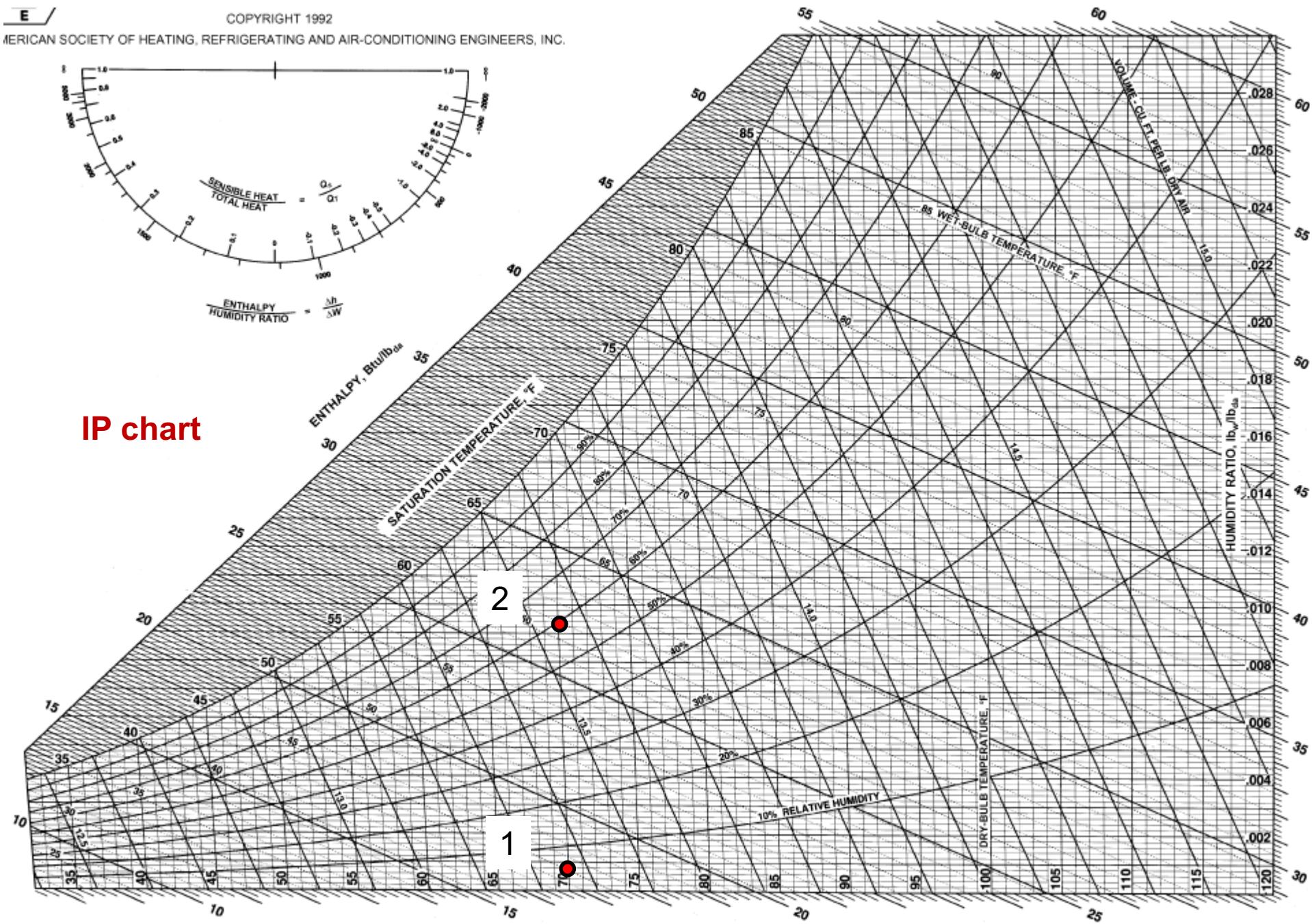
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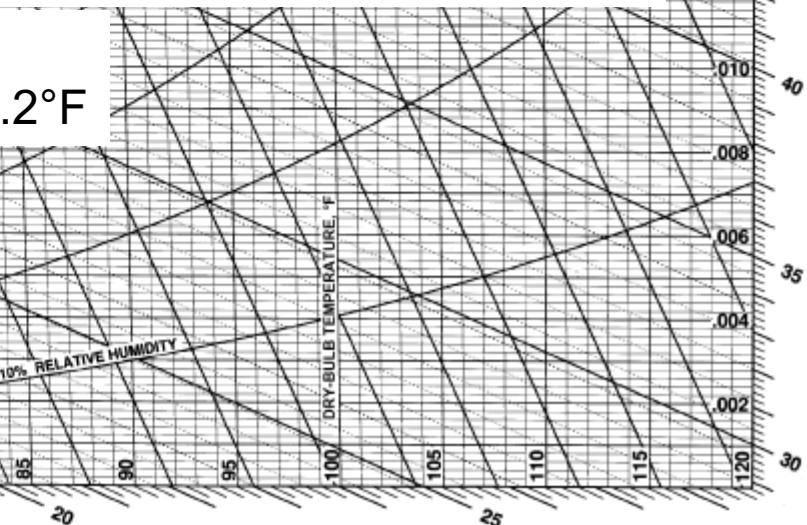
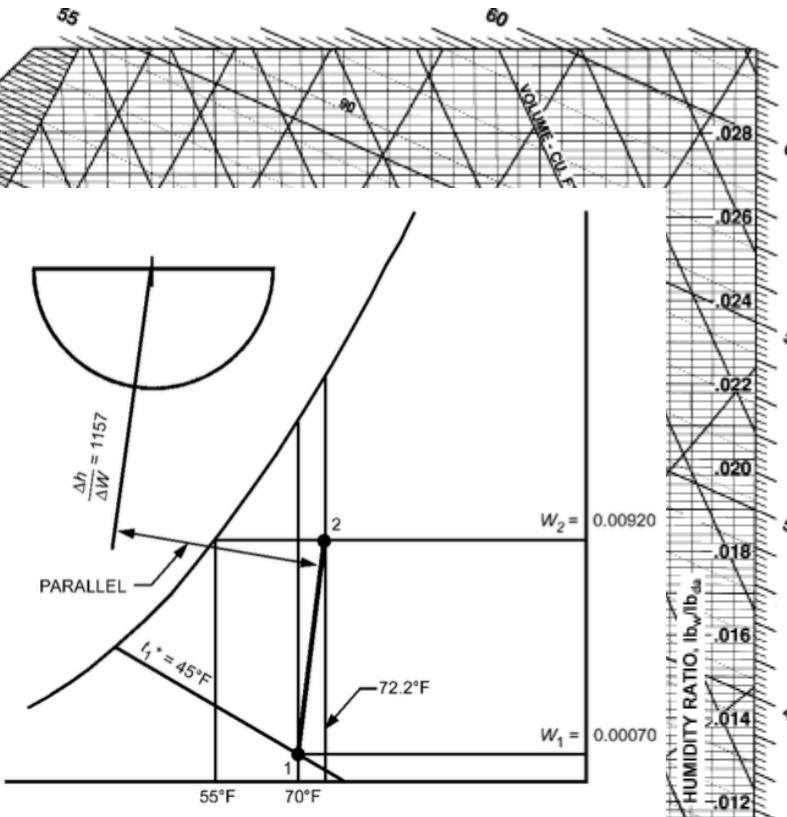
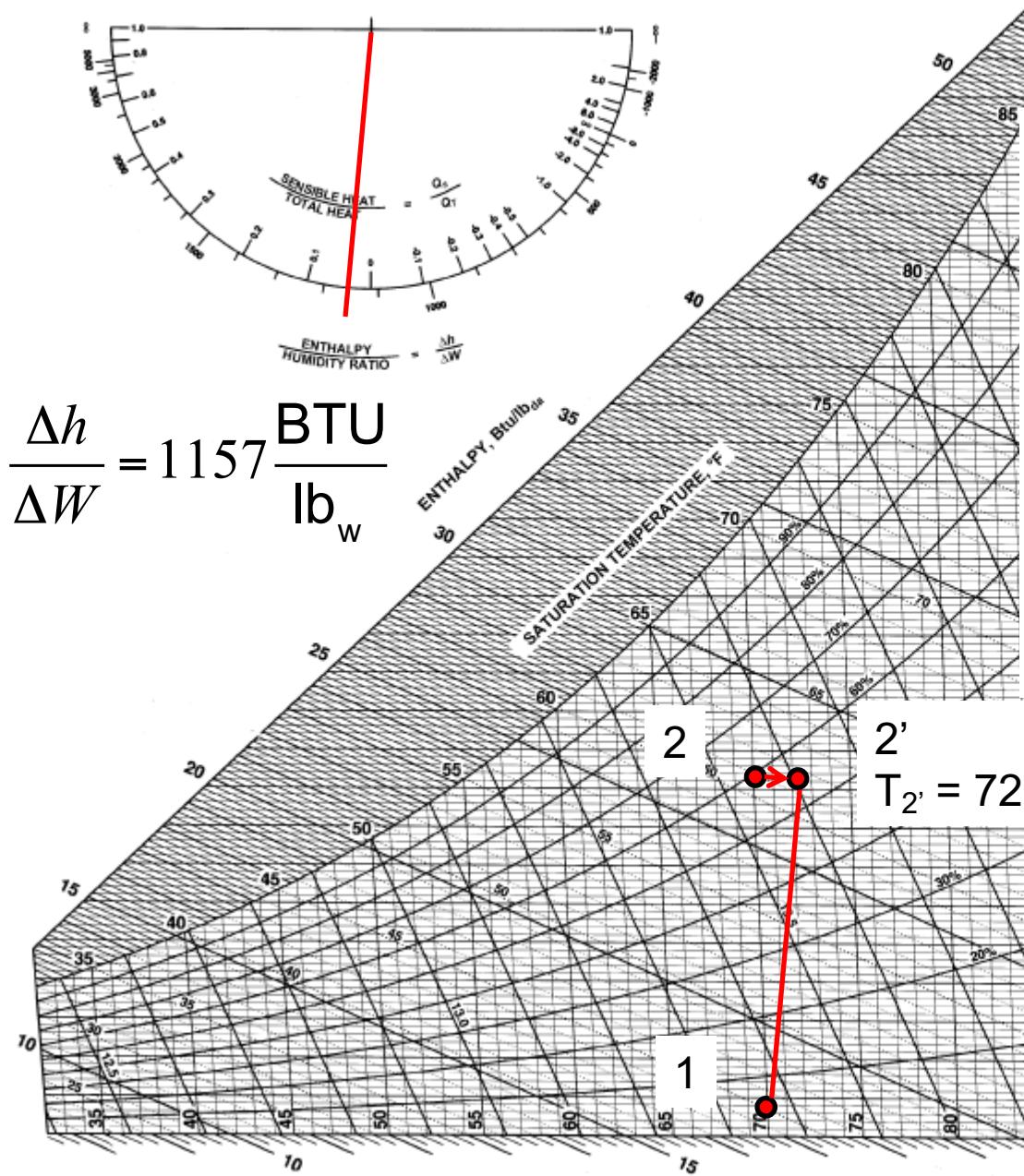
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IP chart



$$\frac{\Delta h}{\Delta W} = 1157 \frac{\text{BTU}}{\text{lb}_w}$$



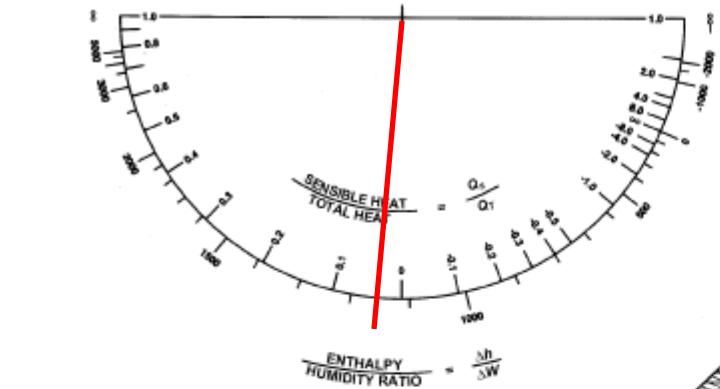
Example 6: Heating and humidification (IP)

- Moist air is heated and humidified by passing it first over a heating coil and then adding moisture. The moist air enters the system at 40°F dry bulb temperature and 36°F wet bulb temperature at a mass flow rate of $235 \text{ lb}_{\text{da}}/\text{min}$. The humidifier injects saturated steam at 230°F. The moist air exists the system at 90°F dry bulb temperature and 40% RH.
- Locate state 2 on a psychrometric chart and determine the rate of heat addition by the heating coil and the rate of mass addition by the humidifier.

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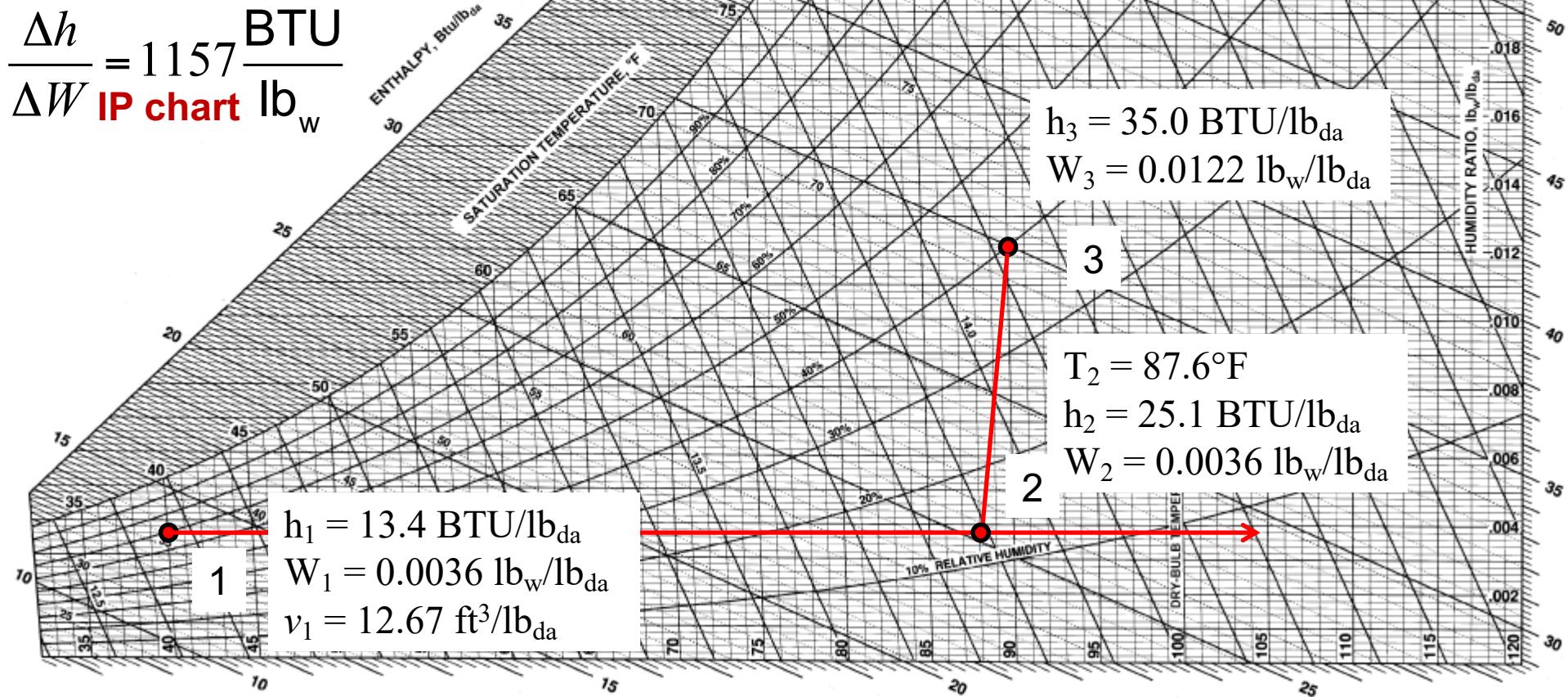
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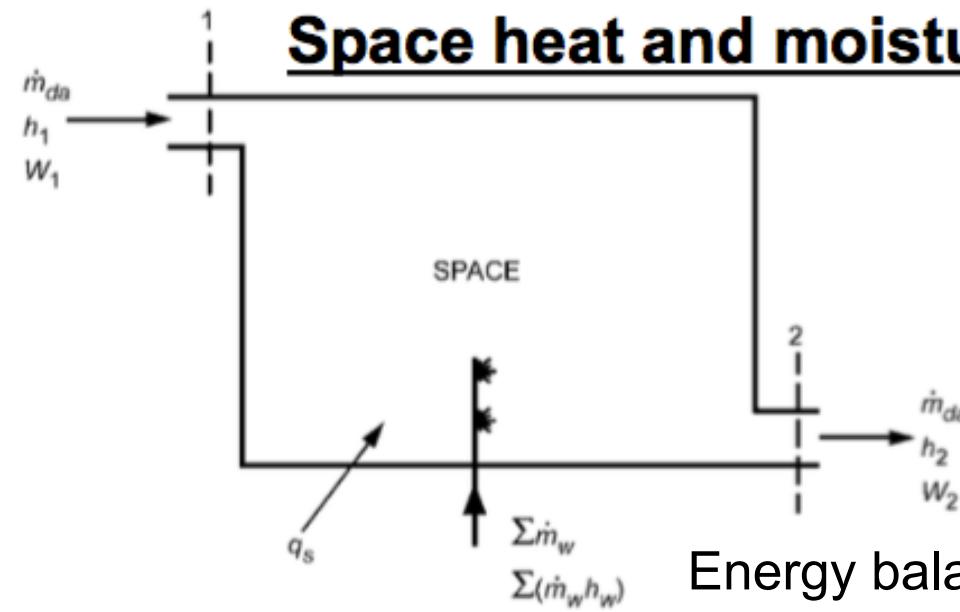
$$\frac{\Delta h}{\Delta W} = 1157 \frac{\text{BTU}}{\text{lb}_w}$$

IP chart



Energy/mass balances for psychrometric processes

Space heat and moisture gains



Energy balance: $\dot{m}_{da} h_1 + Q_{gains} + \sum \dot{m}_w h_w = \dot{m}_{da} h_2$

Mass balance on water vapor: $\dot{m}_{da} W_1 + \sum \dot{m}_w = \dot{m}_{da} W_2$

Therefore: $\sum \dot{m}_w = \dot{m}_{da} (W_2 - W_1)$

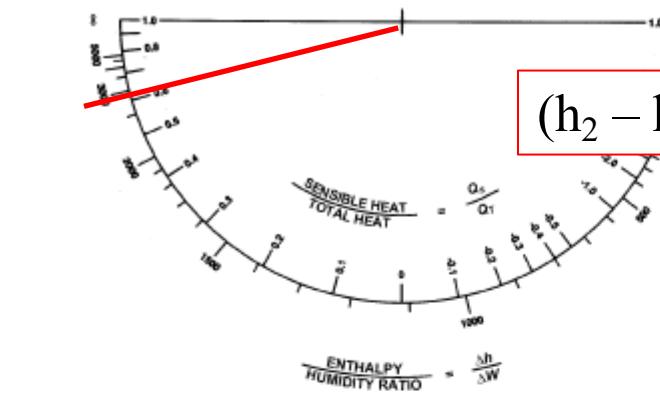
Therefore: $\sum \dot{m}_w h_w + Q_{gains} = \dot{m}_{da} (h_2 - h_1)$

*Note that $h_w = h_g$ for steam/vapor and $h_w = h_f$ for water

And:
$$\frac{\Delta h}{\Delta W} = \frac{\sum \dot{m}_w h_w + Q_{gains}}{\sum \dot{m}_w}$$

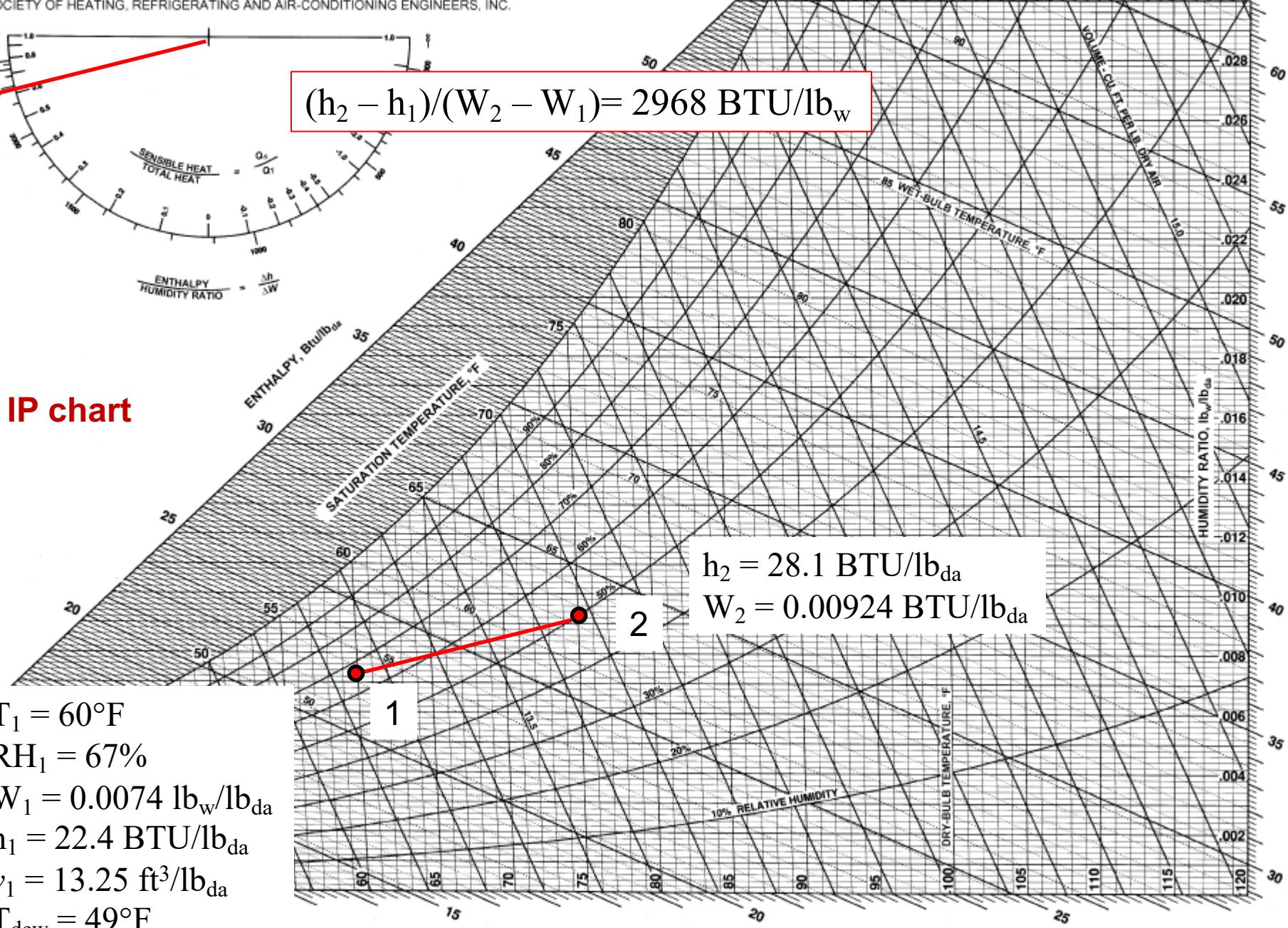
Example 7: Space conditioning – cooling (IP)

- The air in a restaurant is to be maintained at 75°F dry-bulb temperature and 50% RH. The load calculations for the restaurant estimate the rate of sensible heat gain to be 178,000 BTU/h. The rate of moisture gain is estimated to be 95 lb_w/h with an average enthalpy of moisture (h_w) of 1095 BTU/ lb_w . The supply air temperature is to be 60°F. Assume standard atmospheric pressure.
- Determine the following:
 - a) The required dew-point temperature of the supply air
 - b) The required volumetric flow rate of supply air (in CFM)



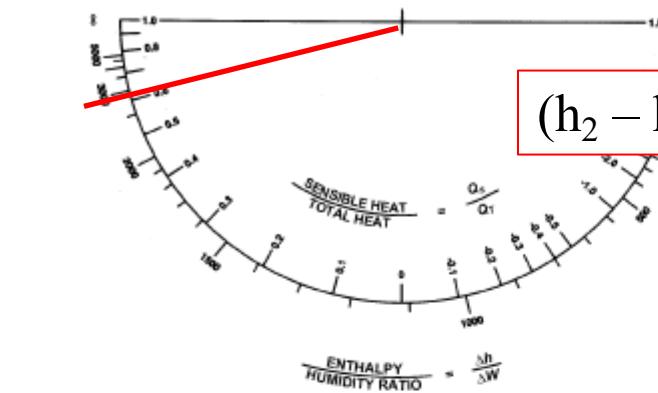
$$(h_2 - h_1)/(W_2 - W_1) = 2968 \text{ BTU/lb}_w$$

IP chart



Example 8: Single-zone space conditioning (IP)

- Assume the restaurant from the previous example is to be served by an air handling unit that mixes outdoor air with recirculated air, then passes the air over a cooling coil to the space. Outside air conditions are 92°F dry bulb and 77°F wet bulb temperatures. The rate of exhaust from the restaurant is 4500 CFM.
- Determine the following:
 - a) The mass flow rate of recirculated air
 - b) The thermodynamic state of the moist air entering the cooling coil
 - c) The refrigeration capacity required



$$(h_2 - h_1)/(W_2 - W_1) = 2968 \text{ BTU/lb}_w$$

IP chart

