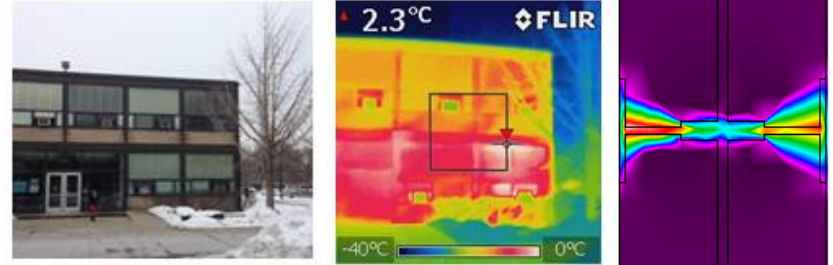


CAE 331/513

Building Science

Fall 2016



Week 6: September 27, 2016

Psychrometrics (chart and definitions)

Built
Environment
Research

@ IIT



*Advancing energy, environmental, and
sustainability research within the built environment*

www.built-envi.com

Twitter: [@built_envi](https://twitter.com/built_envi)

Dr. Brent Stephens, Ph.D.
Civil, Architectural and Environmental Engineering
Illinois Institute of Technology
brent@iit.edu

Exam 1 graded and returned

- Overall grade summary
 - Mean: 84%
 - Standard deviation: 11%
 - Minimum: 61%
 - Maximum: 98%
- Undergraduate average: 85%
- Graduate average: 83%

Graduate student projects

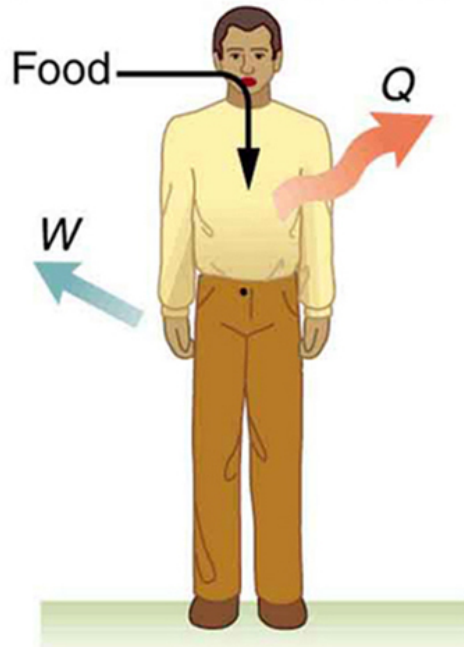
- Expectations document on BB now
 - Individual projects
 - Literature review
 - Some modeling and/or measurement
 - Conference paper and presentation
- Due dates for deliverables:
 - Tuesday, October 18: Project topic
 - Thursday, December 1: Final report and presentation in class

Graduate student projects: Topic suggestions

- Energy questions
 - Efficiency of radiant vs. central forced air heating/cooling?
 - Efficiency of different air distribution systems (e.g., overhead/UFAD)
 - Net zero energy/carbon design/operation
 - Electrical metering and power draw signatures
- HVAC systems
 - Heat pumps, geothermal, energy recovery, absorption chillers, cogeneration
- Green building rating systems
 - LEED, Green Globes, EnergyStar, Living Building, BREEAM, 189.1
- Moisture
 - Dampness, fungal growth, remediation, buffering capacity
- IAQ/IEQ
 - Thermal comfort, aerosols, ventilation, VOCs
- Other
 - Electrical, lighting, plumbing, acoustics
- Tools you can use:
 - Energy simulation, MATLAB modeling, measurements in BERG Lab

Last time (September 15)

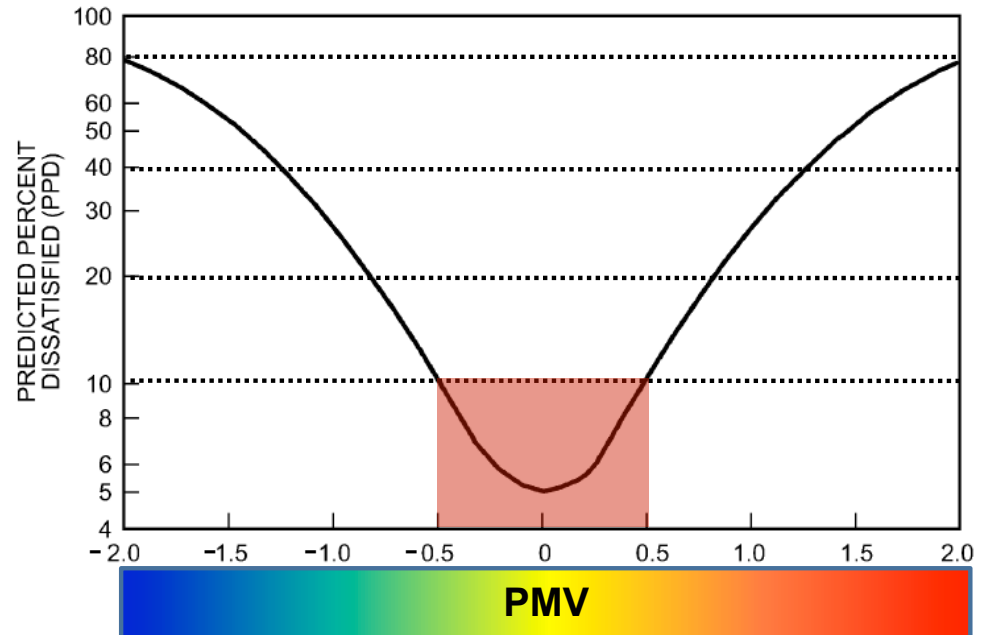
- Human thermal comfort



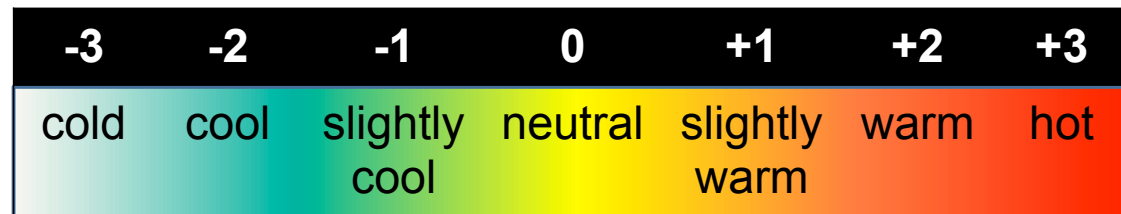
$$\dot{Q} = MA_{skin}$$

$$A_D = 0.202m^{0.425}l^{0.725}$$

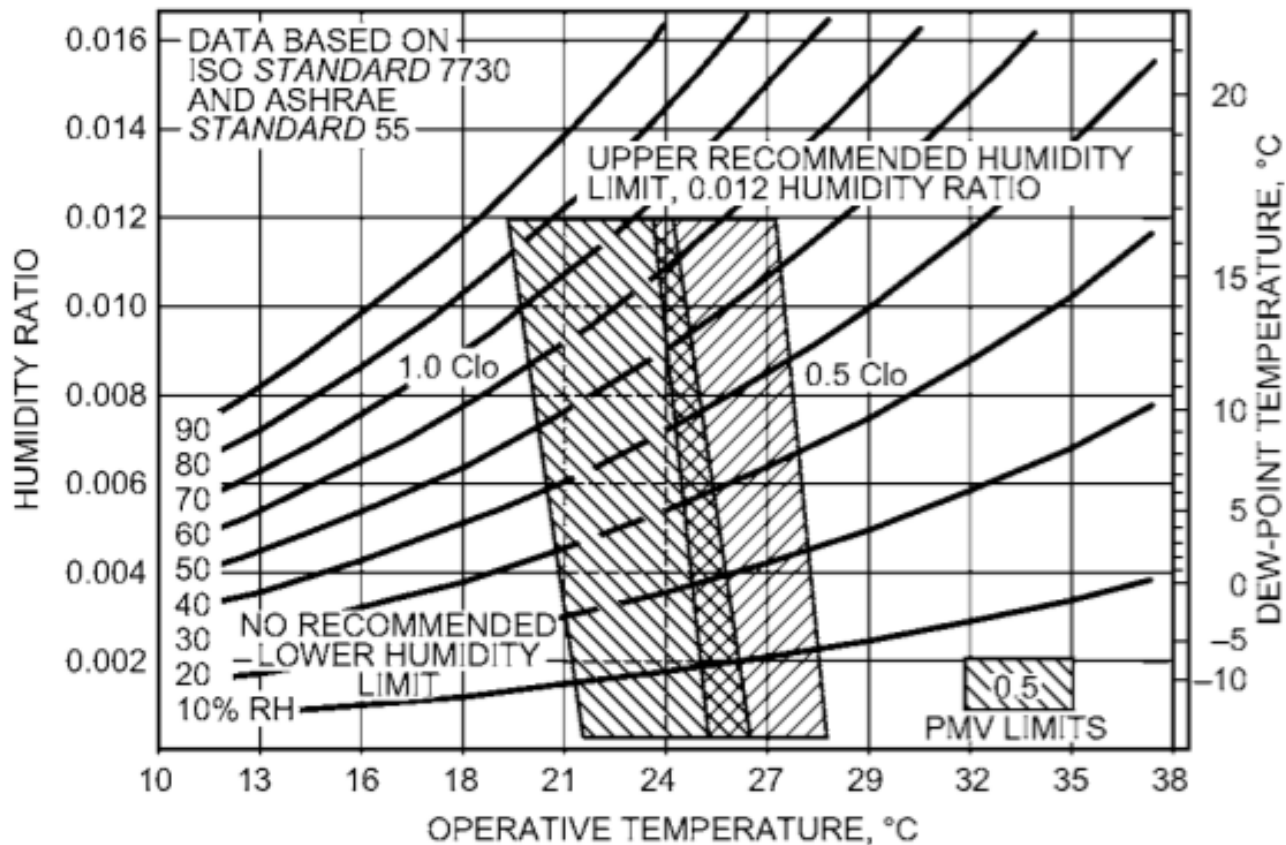
A_D = DuBois surface area, m^2
 m = mass, kg
 l = height, m



$$1 \text{ met} = 18.4 \frac{\text{Btu}}{\text{h} \cdot \text{ft}^2} = 58 \frac{\text{W}}{\text{m}^2}$$



ASHRAE comfort zone



3 equations for operative temperatures:

$$t_o = \frac{(h_r t_{mr} + h_c t_a)}{h_r + h_c}$$

$$t_o = \frac{(t_{mr} + (t_a \times \sqrt{10v}))}{1 + \sqrt{10v}}$$

$$t_o = \frac{(t_a + t_{mr})}{2}$$

Fig. 5 ASHRAE Summer and Winter Comfort Zones
 [Acceptable ranges of operative temperature and humidity with air speed ≤ 0.2 m/s for people wearing 1.0 and 0.5 clo clothing during primarily sedentary activity (≤ 1.1 met)].

ASHRAE comfort zone: CBE Thermal Comfort Tool

CBE Thermal Comfort Tool

ASHRAE-55

Compare

Ranges

Select method:

PMV method

Air temperature

24.6 °C

Use operative temperature

Mean radiant temperature

26 °C

Air speed

0.07

Humidity

50

Metabolic

1.3

Clothing

0.55

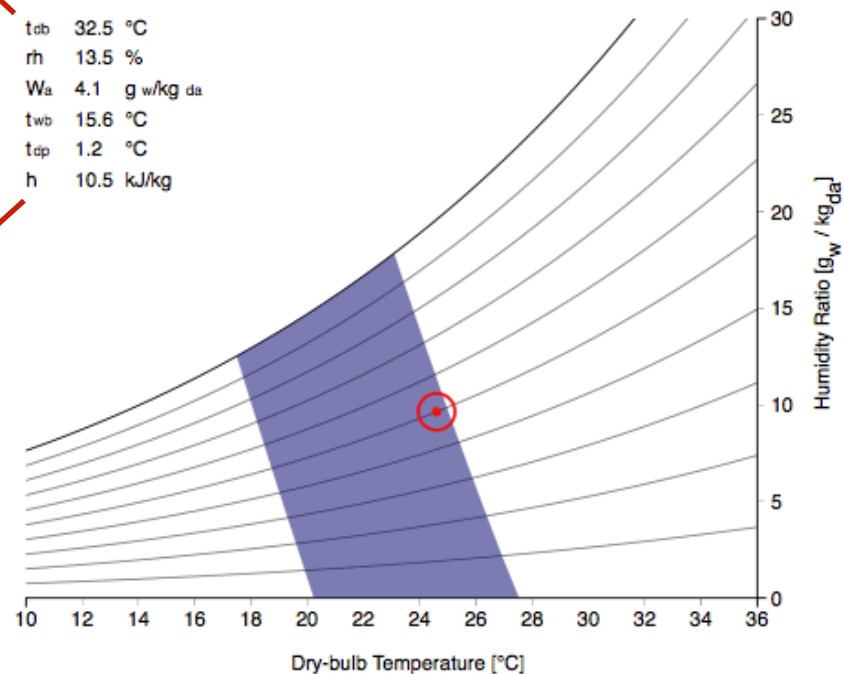
t_{db} 32.5 °C
 rh 13.5 %
 W_a 4.1 g w/kg da
 t_{wb} 15.6 °C
 t_{dp} 1.2 °C
 h 10.5 kJ/kg

✓ Complies with ASHRAE Standard 55-2010

PMV 0.44
PPD 9%
Sensation Neutral
SET 26.7°C

Psychrometric chart (air temperature)

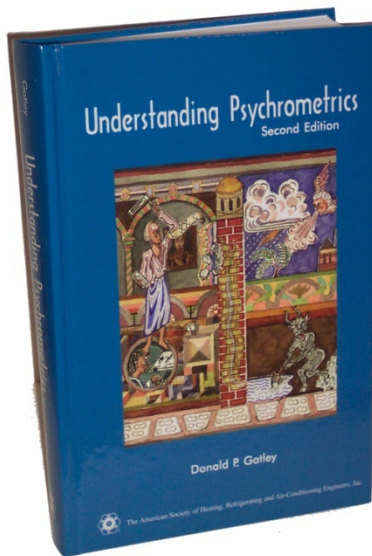
t_{db} 32.5 °C
 rh 13.5 %
 W_a 4.1 g w/kg da
 t_{wb} 15.6 °C
 t_{dp} 1.2 °C
 h 10.5 kJ/kg



PSYCHROMETRICS

Psychrometrics

Psychrometrics is the science and engineering of air/vapor mixtures



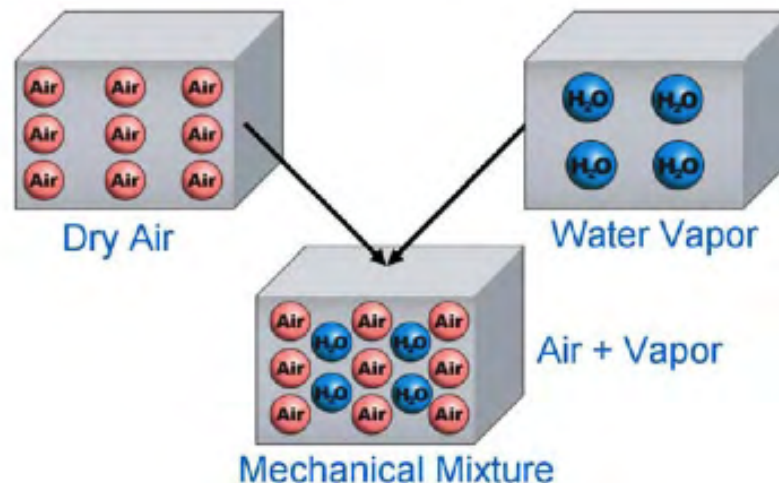
- For architectural engineers and building scientists, the vapor is usually **water vapor**
- We use psychrometrics to relate the thermodynamic and physical properties of moist air

Applying psychrometrics

- We need to understand **air temperature** and **moisture content** to understand **human thermal comfort**
 - In **hot, humid** weather we design HVAC systems to **remove** moisture by dehumidification/cooling
 - In **dry, cold** weather, we **add** moisture by humidifiers
- We are also concerned about moisture for energy use, structural, aesthetic, and indoor air quality reasons
- Psychrometrics also involves learning how to use and combine a variety of moist air parameters

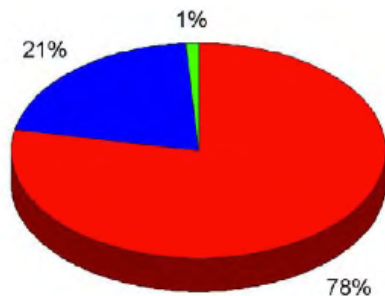
Some definitions for psychrometrics

- **Atmospheric air** contains:
 - Many gaseous components
 - Water vapor
 - Contaminants (particulate matter and gaseous pollutants)
- **Dry air** is atmospheric air with all of the water vapor removed
- **Moist air** is a two-component mixture of dry air and water vapor



Standard composition of dry air

Gas	Molecular weight (g/mol)	Volume %
Nitrogen (N ₂)	32.000	78.084
Oxygen (O ₂)	28.016	20.946
Argon (Ar)	39.444	0.9340
Carbon Dioxide (CO ₂)	44.010	0.03697
Neon (Ne)	20.179	0.00182
Helium (He)	4.002	0.00052
Methane (CH ₄)	16.042	0.00014
Krypton	83.800	0.00010



■ Nitrogen ■ Oxygen ■ Other Gases

Where does water fit in?

Standard composition of **moist** air

Gas	Molecular weight (g/mol)	Volume %
Nitrogen (N ₂)	32.000	78.084%
Oxygen (O ₂)	28.016	20.946%
Water (H₂O)	18.015	0 to 4%
Argon (Ar)	39.444	0.9340%
Carbon Dioxide (CO ₂)	44.010	0.03697%
Neon (Ne)	20.179	0.00182%
Helium (He)	4.002	0.00052%
Methane (CH ₄)	16.042	0.00014%
Krypton	83.800	0.00010%

Key terms for describing moist air

- To describe and deal with moist air, we need to be able to describe the relative portions of dry air and water vapor
- There are several different equivalent measures...
- Which one you use depends on what data you have to start with and what quantity you are trying to find
- If you know two properties, you can usually get all the others

Key terms for describing moist air

Key terms to learn today:

1. Dry bulb temperature
2. Vapor pressure
3. Saturation
4. Relative humidity
5. Absolute humidity (or humidity ratio)
6. Dew point temperature
7. Wet bulb temperature
8. Enthalpy
9. Density
10. Specific volume

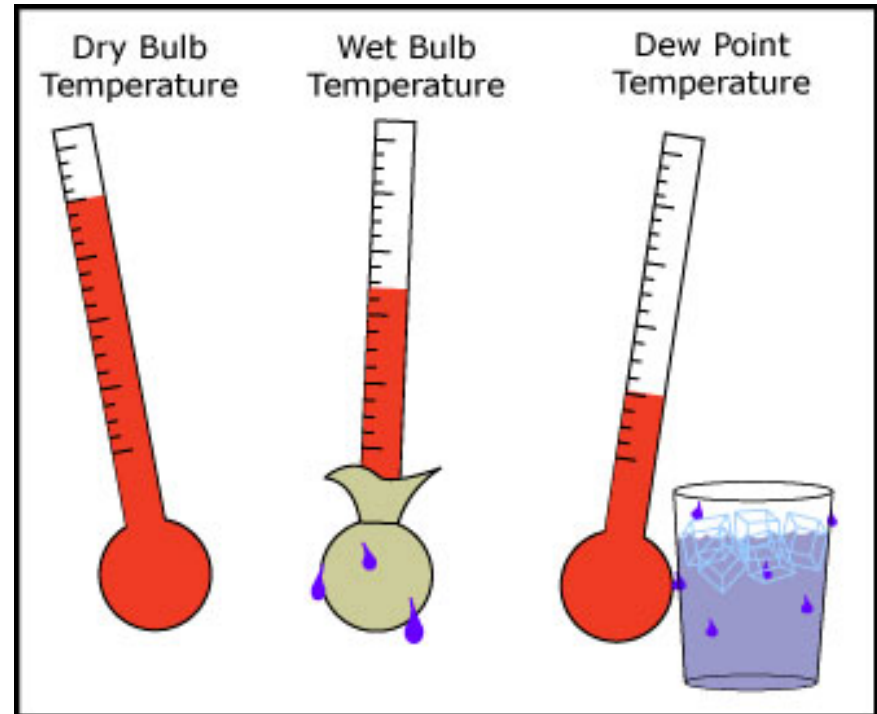
Three different temperatures: T , T_{dew} , and T_{wb}

The standard temperature, T , we are all familiar with is called the **dry-bulb** temperature, or T_d

- It is a measure of internal energy

We can also define:

- **Dew-point** temperature, T_{dew}
 - Temperature at which water vapor changes into liquid (condensation)
 - Air is maximally **saturated** with water vapor
 - **Wet-bulb** temperature, T_{wb}
 - The temperature that a parcel of air would have if it were cooled to saturation (100% **relative humidity**) by the evaporation of water into it
- ✓ The energy needed to evaporate liquid water (heat of vaporization) is taken from the air in the form of sensible heat and converted to latent heat, which lowers the temperature at constant enthalpy



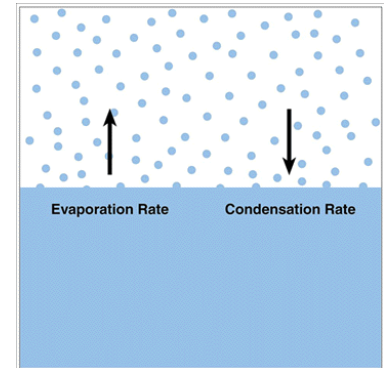
*Units of Celsius, Fahrenheit, or Kelvin

Key concepts: Vapor pressure and Saturation

- Air can hold moisture (i.e., **water vapor**)
- **Vapor pressure** is the pressure exerted by a vapor in thermodynamic equilibrium with its condensed phases

$$p_w$$

*Units of pressure, Pa or kPa
(aka “**partial pressure**”)



- The amount of moisture air can hold in vapor form before condensation occurs is dependent on temperature
 - We call the limit **saturation**

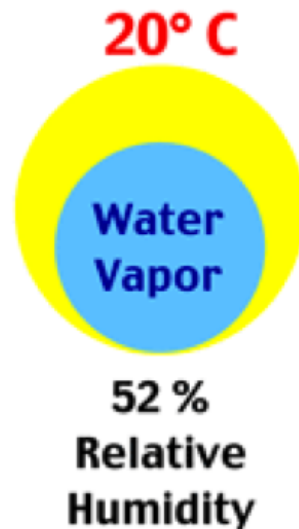
$$p_{ws}$$

*Units of pressure, Pa or kPa
(aka “**saturation vapor pressure**”)



Key concept: Relative humidity, ϕ

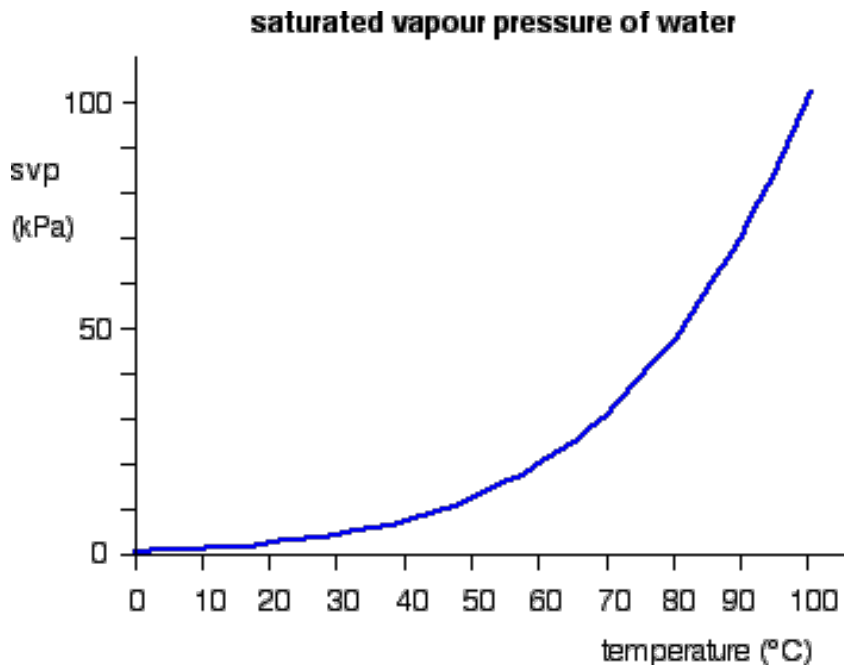
- **Relative humidity** (RH, or ϕ) is the ratio of the vapor pressure of water vapor in a sample of air to the **saturation** vapor pressure at the dry bulb temperature of the sample
- Relative humidity \neq absolute humidity!



$$\phi = \frac{p_w}{p_{ws}}$$

Key concept: Saturation vapor pressure, p_{ws}

- The **saturation vapor pressure** is the partial pressure of water vapor at saturation (p_{ws}) *Units of pressure, Pa or kPa
 - Cannot absorb any more moisture at that temperature
- We can look up p_{ws} in tables (as a function of T)
 - Table 3 in Ch.1 of 2013 ASHRAE Fundamentals
- We can also use empirical equations:



Equation for p_{ws} :

$$\ln p_{ws} = \frac{C_8}{T} + C_9 + C_{10}T + C_{11}T^2 + C_{12}T^3 + C_{13} \ln T$$

where

$$C_8 = -5.800\ 220\ 6\ E+03$$

$$C_9 = 1.391\ 499\ 3\ E+00$$

$$C_{10} = -4.864\ 023\ 9\ E-02$$

$$C_{11} = 4.176\ 476\ 8\ E-05$$

$$C_{12} = -1.445\ 209\ 3\ E-08$$

$$C_{13} = 6.545\ 967\ 3\ E+00$$

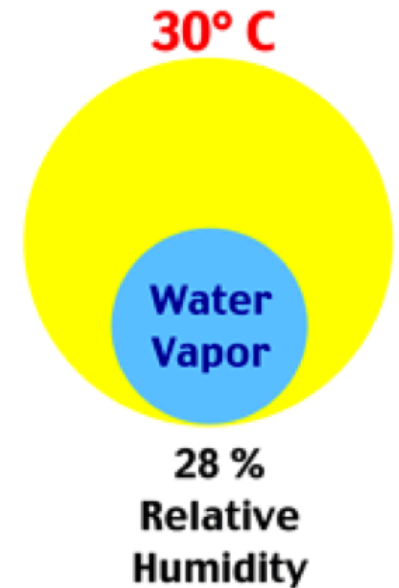
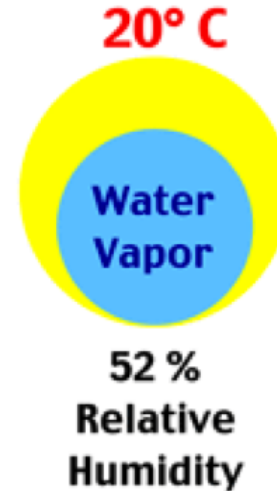
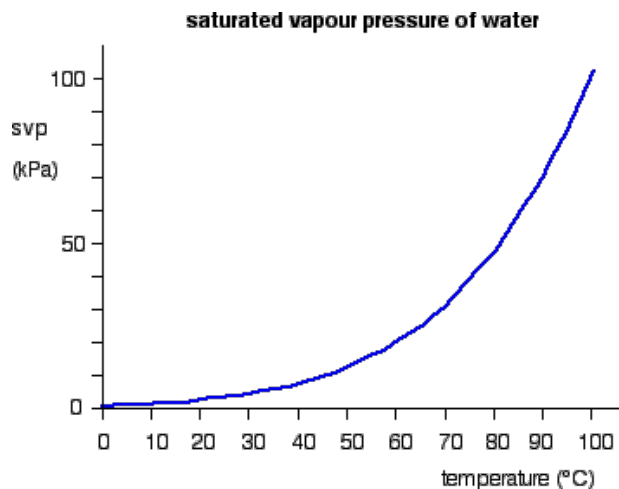
p_{ws} = saturation pressure, Pa

T = absolute temperature, K = °C + 273.15

Relative humidity and temperature

- Relative humidity (RH, or ϕ) is a function of temperature

$$\phi = \frac{p_w}{p_{ws}}$$



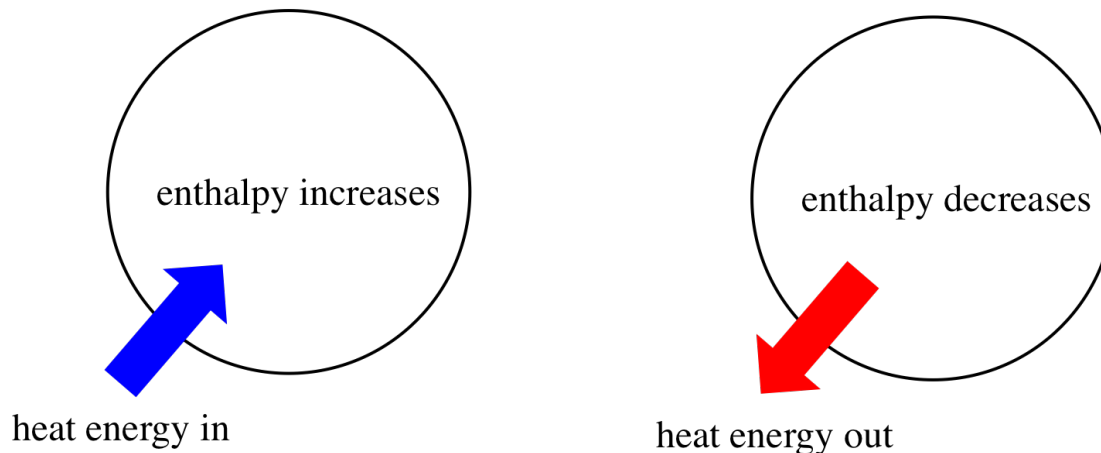
Key concept: Humidity ratio, W

- The **humidity ratio** is a measure of the **mass of water vapor** present in a parcel of air (a measure of *absolute humidity*)
- The humidity ratio is simply the mass of water vapor that exists in a parcel of mass of *dry air*
 - Units of mass of water vapor per mass of dry air
 - kg/kg ($\text{kg}_w/\text{kg}_{da}$)
 - g/kg (g_w/g_{da})

$$W = \frac{\text{mass of water vapor}}{\text{mass of dry air}} \left[\frac{\text{kg}_w}{\text{kg}_{da}} \right]$$

Key concept: **Enthalpy**

- **Enthalpy** is a measure of the amount of energy in a system
 - Units of Joules or BTU (or J/kg or BTU/lb)
- The enthalpy of moist air is the total enthalpy of the dry air plus the water vapor mixture per mass of moist air
- Includes:
 - Enthalpy of dry air, or **sensible** heat
 - Enthalpy of evaporated water, or **latent** heat



Key concept: Density and specific volume

Air density

- Density is a measure of the mass of moist air per unit volume of air
- Includes mass of dry air + water vapor

$$\rho = \frac{\text{mass of moist air}}{\text{volume of moist air}} \quad \left[\frac{\text{kg}}{\text{m}^3} \right]$$

Specific volume

- Specific volume is the volume of unit mass of dry air at a given temperature, expressed as m³/kg (inverse of dry air density)

$$v = \frac{\text{volume of dry air}}{\text{mass of dry air}} \quad \left[\frac{\text{m}^3}{\text{kg}_{\text{da}}} \right]$$

The Psychrometric Chart

- There are both simple and complex ways to estimate these properties
 - Equations and tables (more complex, save for next lecture)
 - Graphically using ...
- **The Psychrometric Chart**
 - Plots dry bulb temperature (T) on the x-axis and humidity ratio (W) on the y-axis
 - Shows relationships between T and W and relative humidity, wet-bulb temperature, vapor pressure, specific volume, and enthalpy
 - Charts are unique at each value of atmospheric pressure (p)
- Both SI and IP versions are on BB in the materials folder



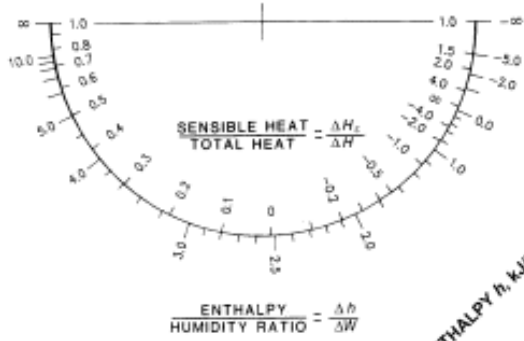
ASHRAE PSYCHROMETRIC CHART NO.1

NORMAL TEMPERATURE SEA LEVEL

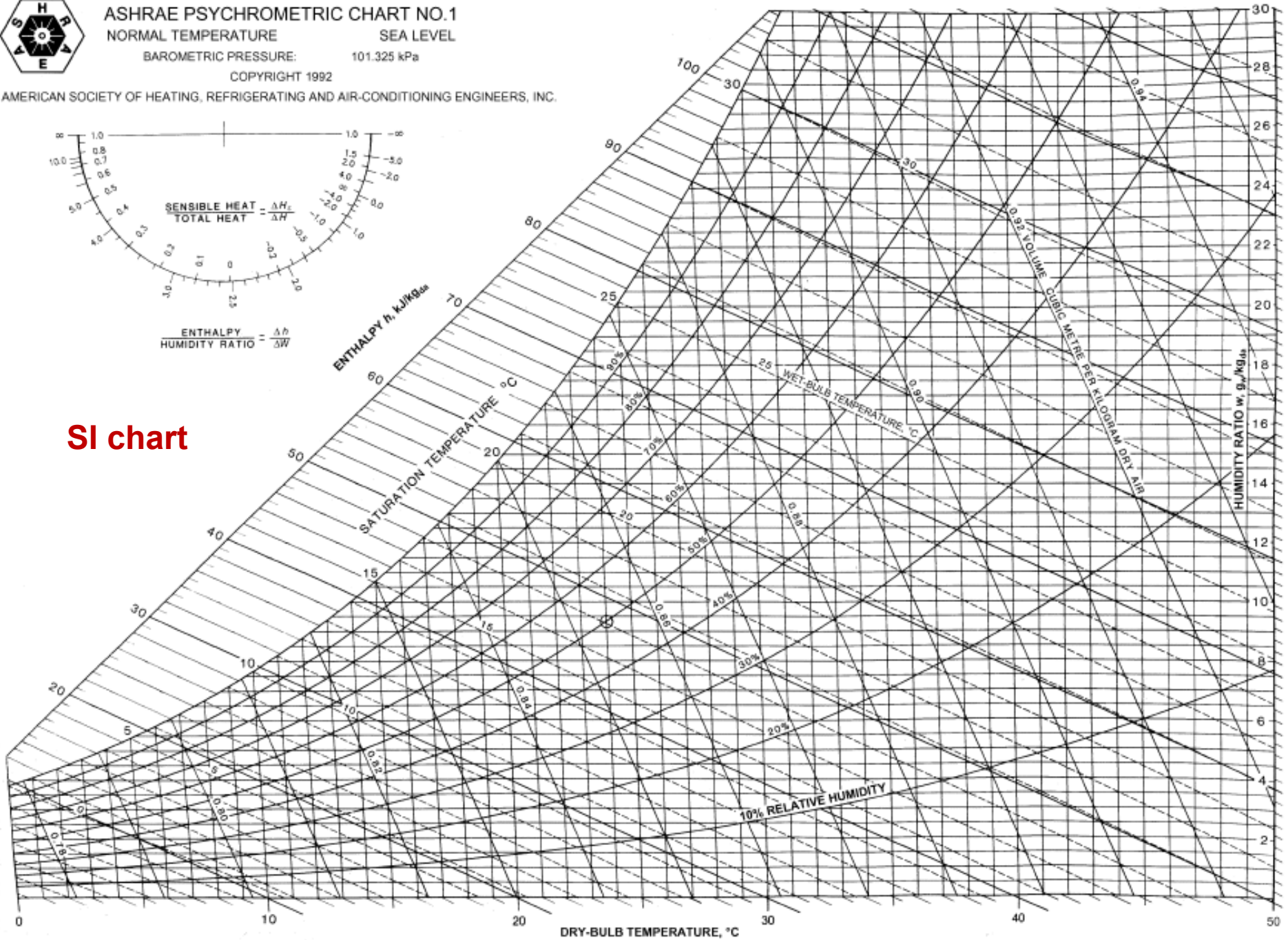
BAROMETRIC PRESSURE: 101.325 kPa

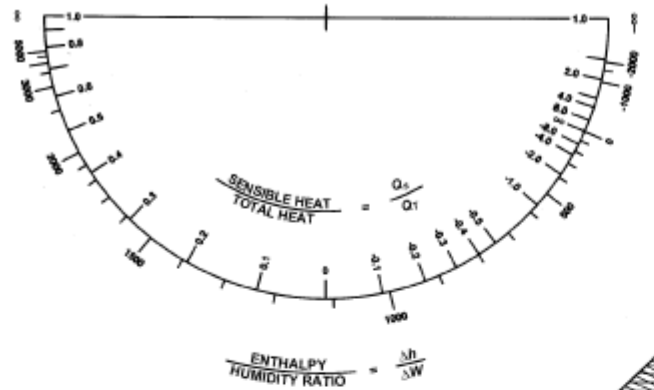
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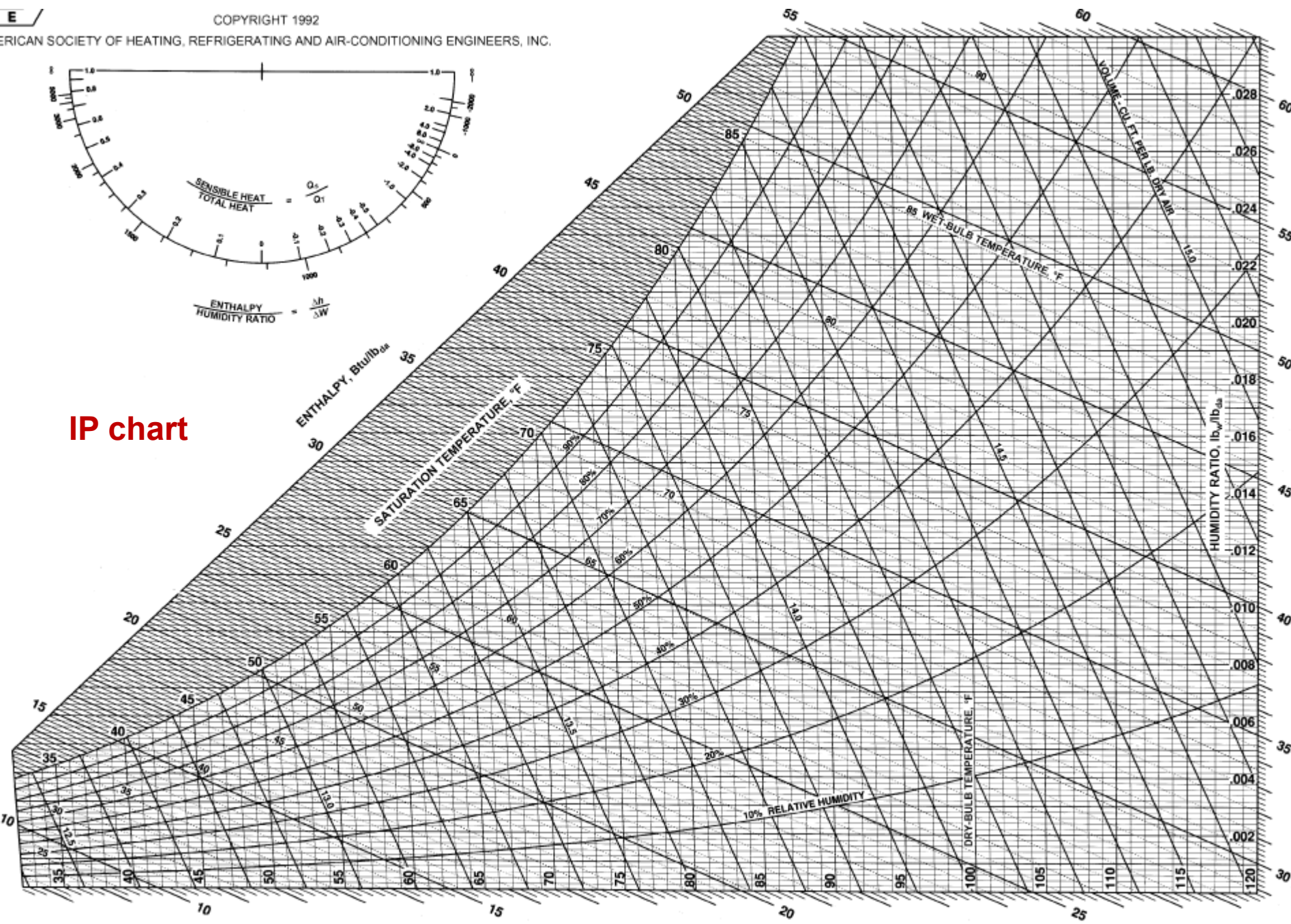


SI chart





IP chart



ASHRAE Psychrometric Chart No. 1

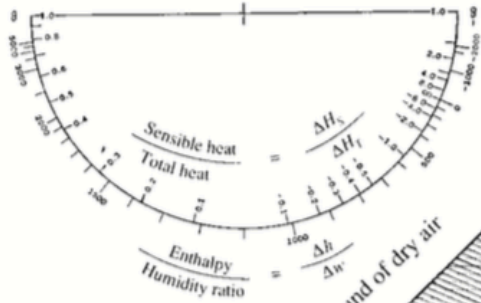
Normal temperature

Barometric pressure 29.921 inches of mercury

Copyright 1963

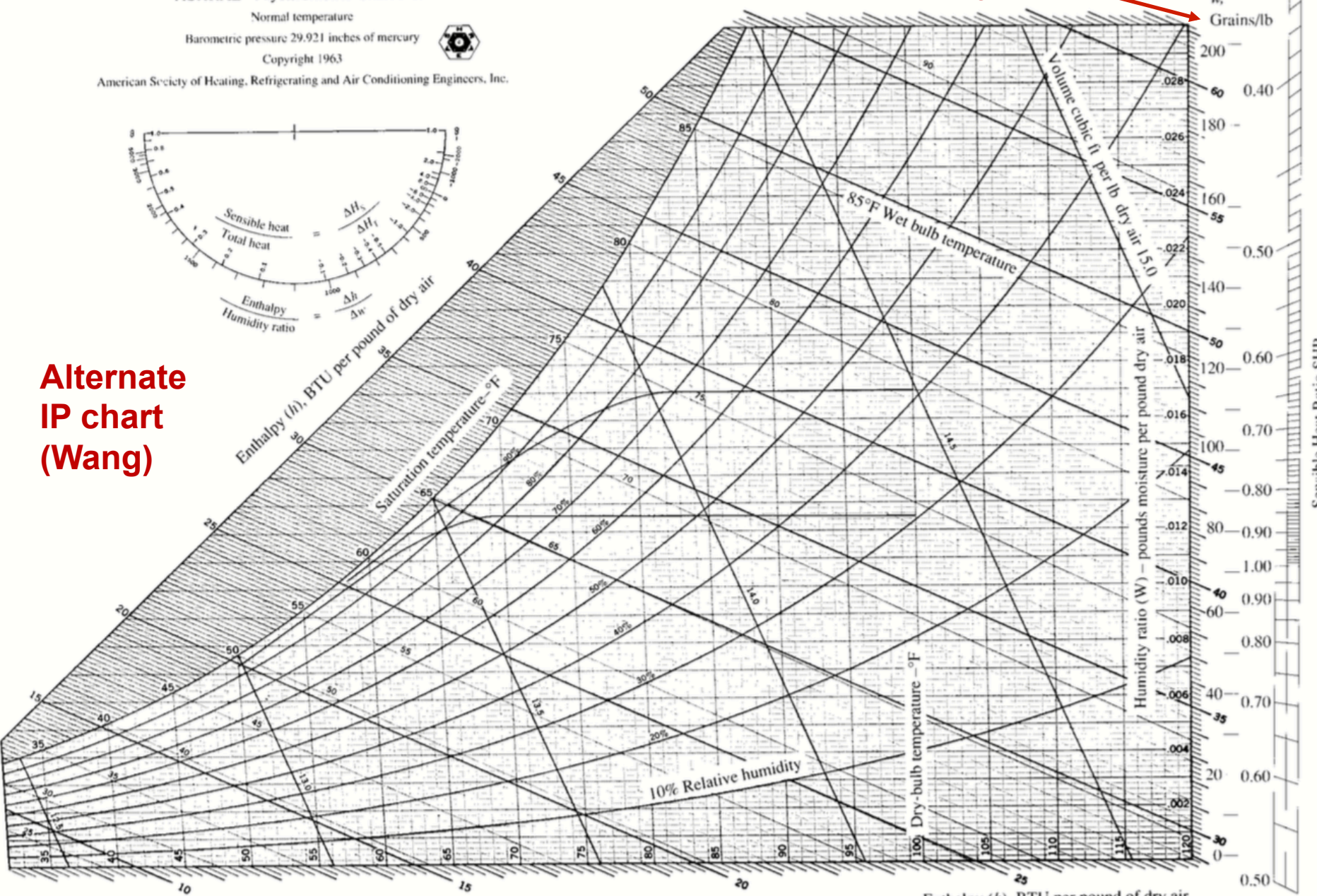
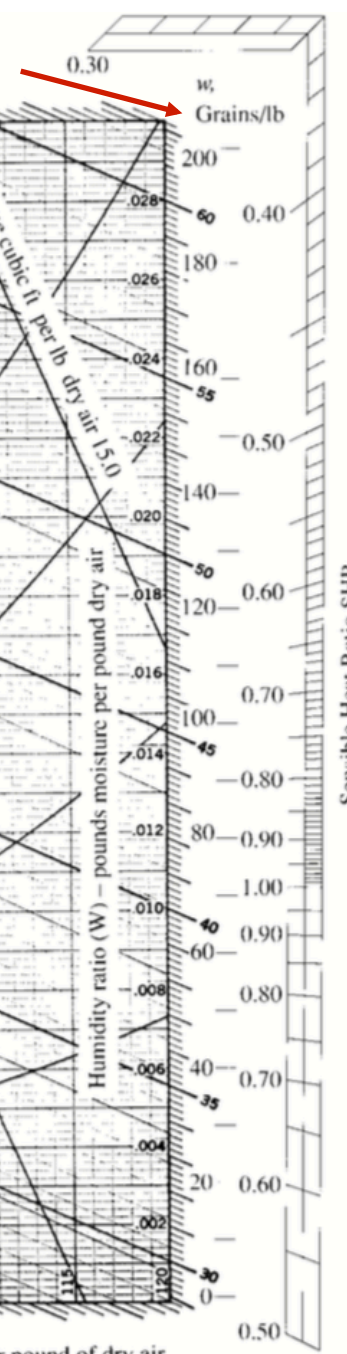


American Society of Heating, Refrigerating and Air Conditioning Engineers, Inc.



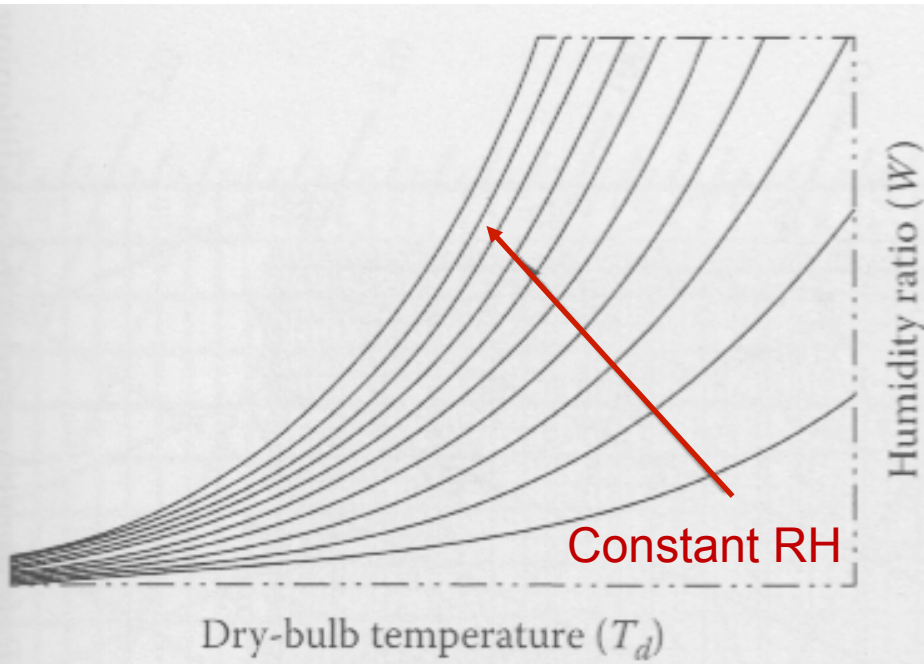
Alternate IP chart (Wang)

grains/lb:
1 lb = 7000 grains

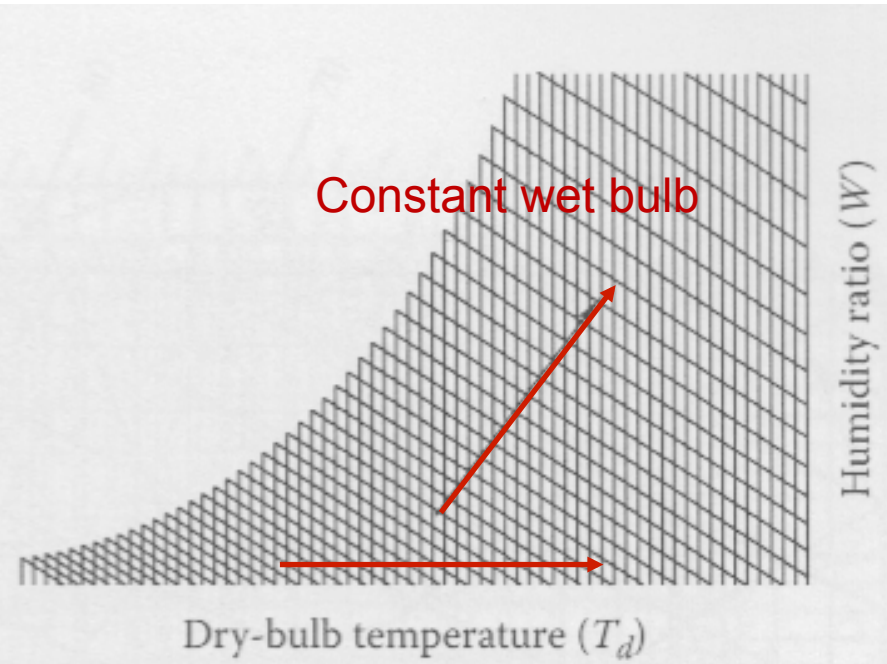


Deciphering the psychrometric chart

Lines of constant RH



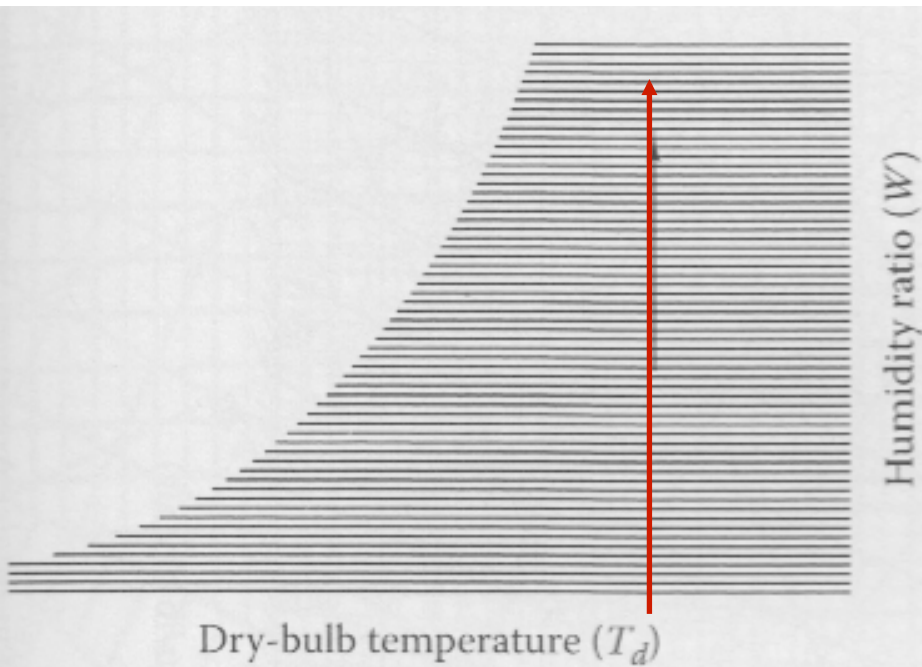
Lines of constant wet-bulb and dry-bulb



Constant dry bulb temperature

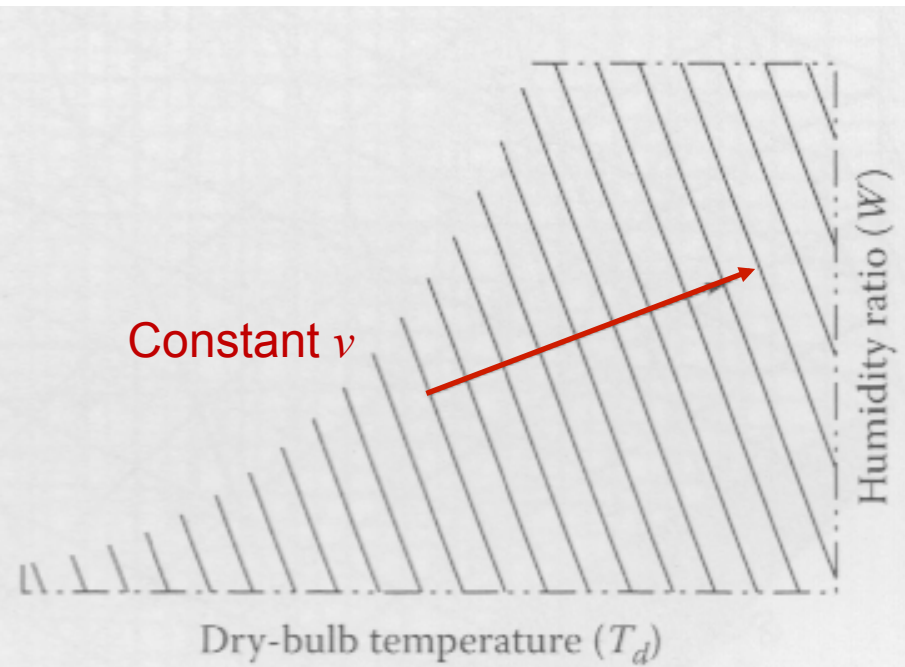
Deciphering the psychrometric chart

Lines of constant humidity ratio



Constant W

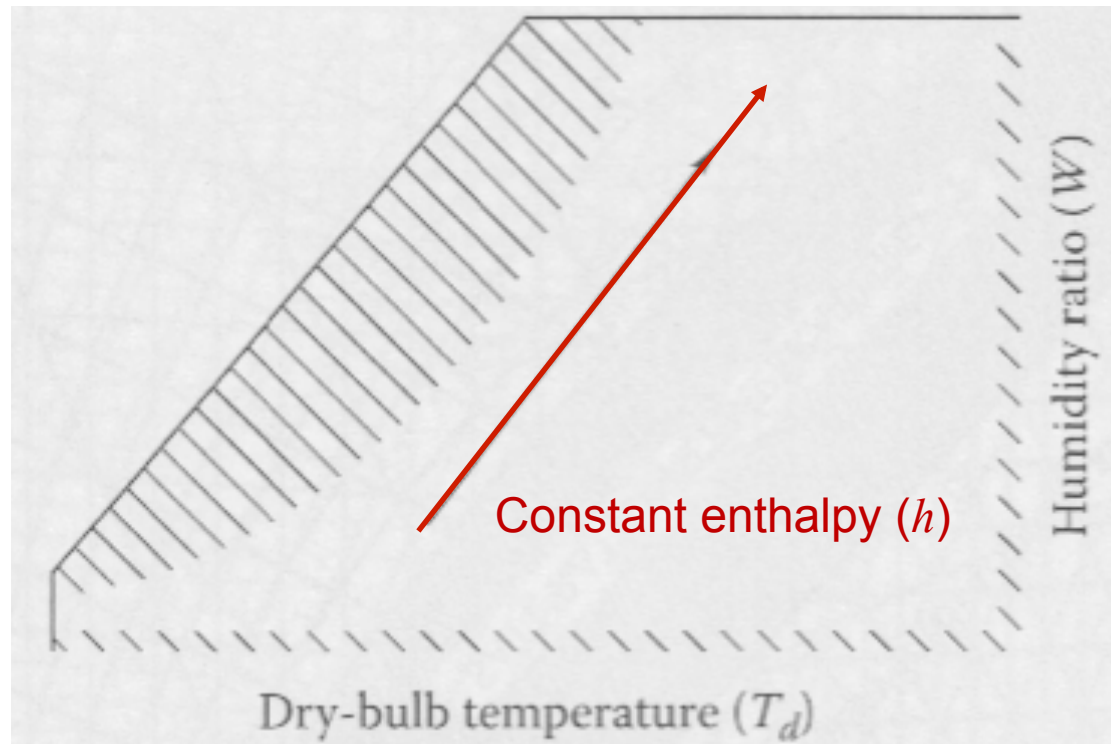
Lines of constant specific volume



Constant v

Deciphering the psychrometric chart

Lines of constant enthalpy





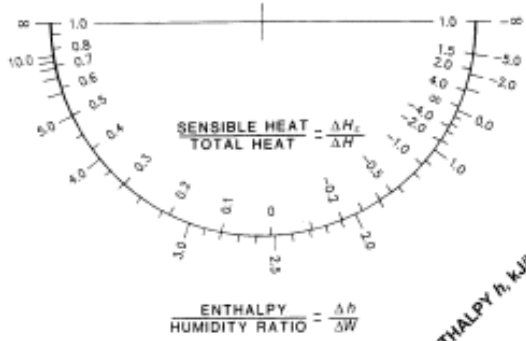
ASHRAE PSYCHROMETRIC CHART NO.1

NORMAL TEMPERATURE SEA LEVEL

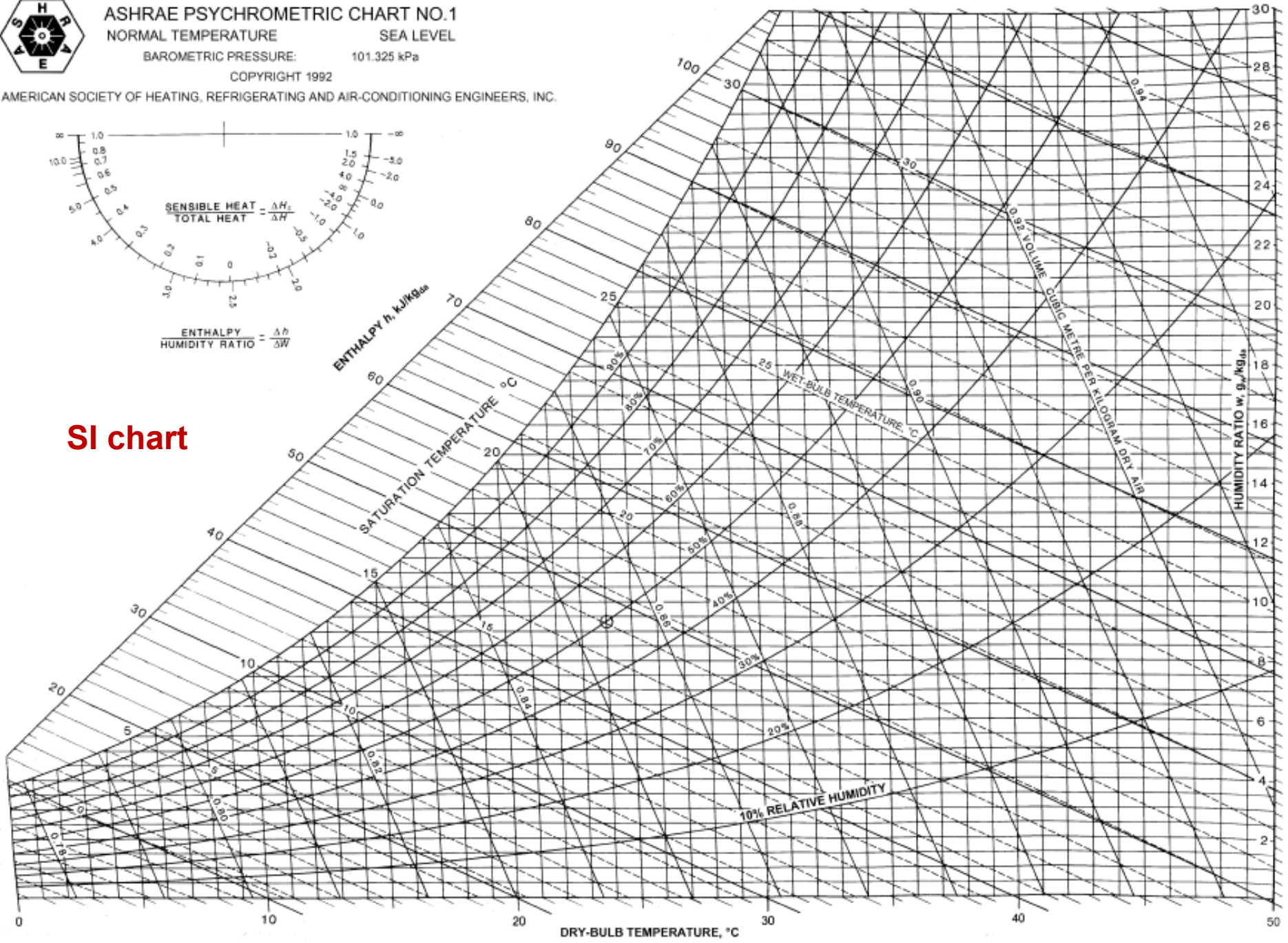
BAROMETRIC PRESSURE: 101.325 kPa

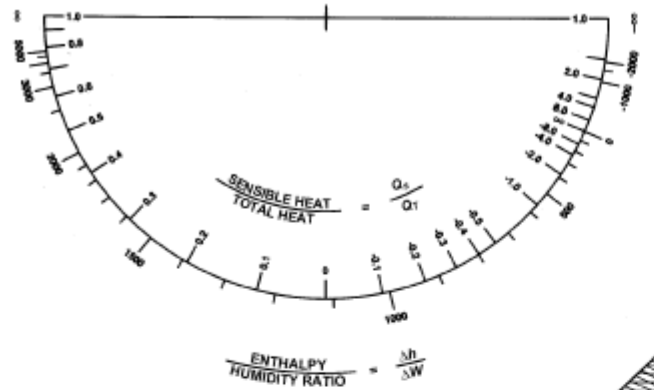
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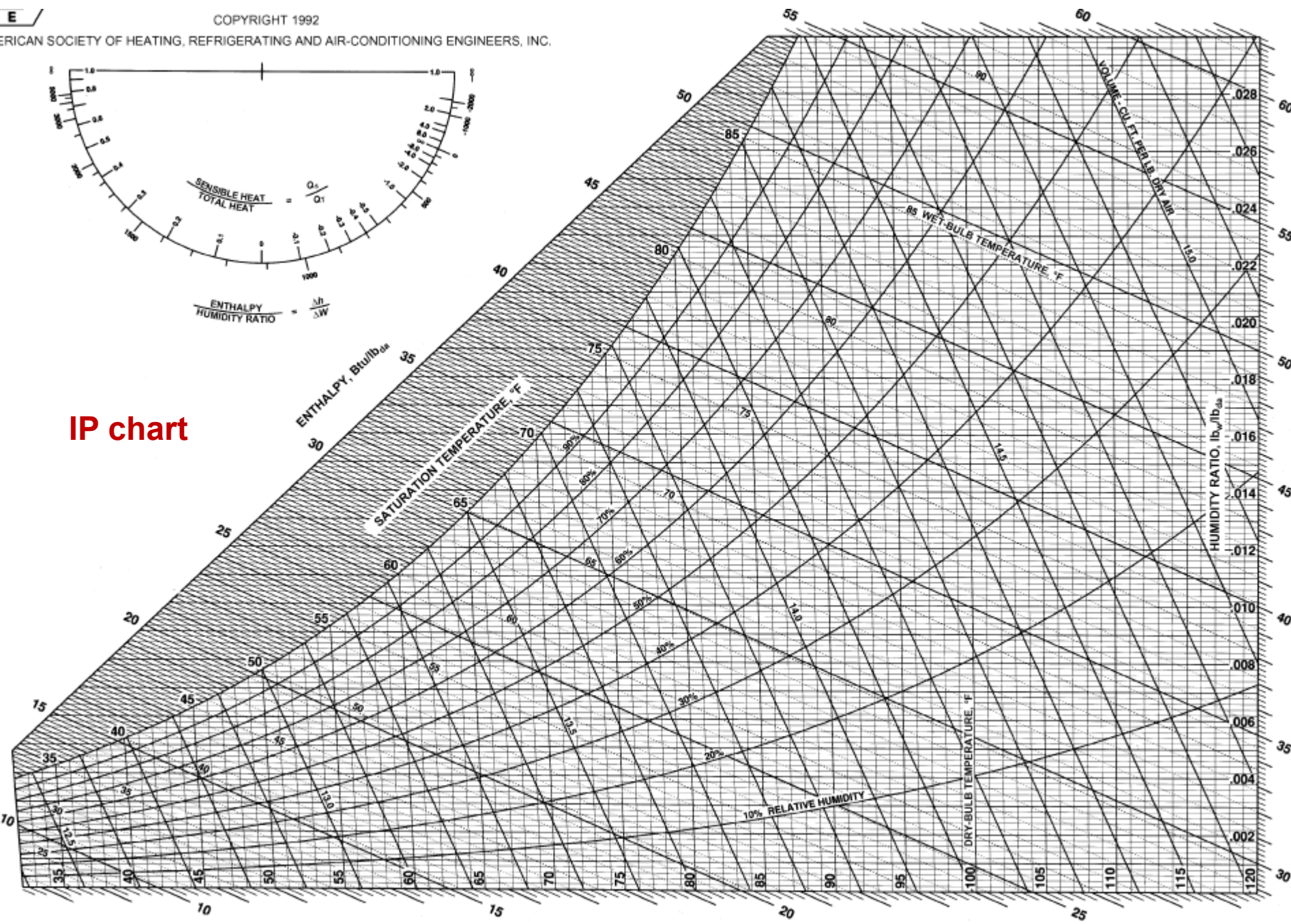


SI chart





IP chart



Some psychrometric examples

Moist air exists at 22°C dry-bulb temperature with 50% RH

Find the following:

- (a) the humidity ratio, W
- (b) dew point temperature, T_{dew}
- (c) wet-bulb temperature, T_{wb}
- (d) enthalpy, h
- (e) specific volume, v
- (f) dry air density, ρ



ASHRAE PSYCHROMETRIC CHART NO.1

NORMAL TEMPERATURE

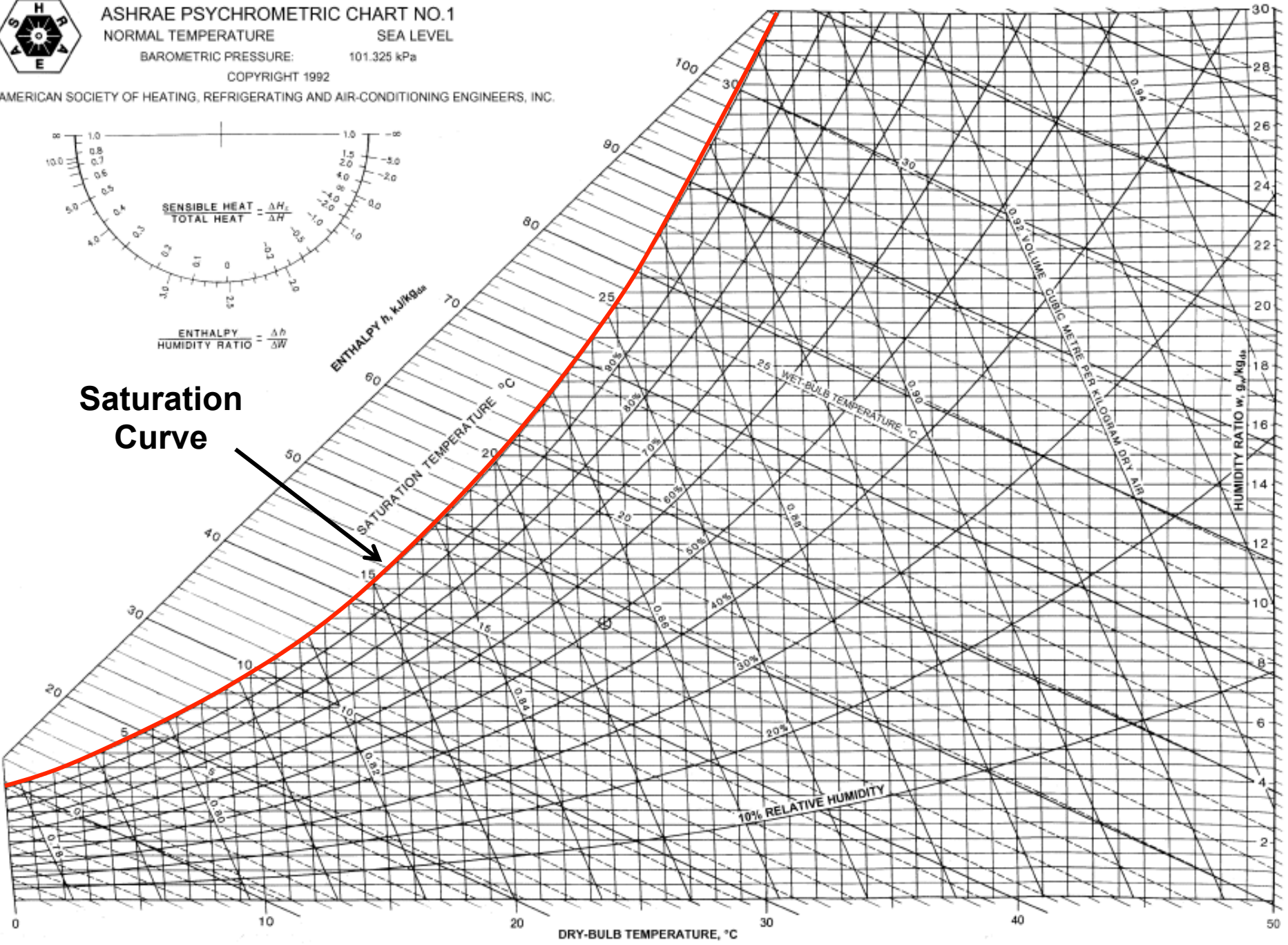
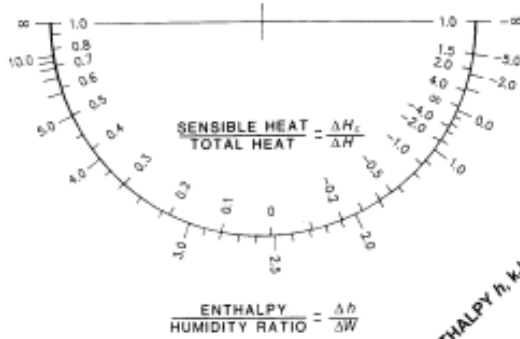
SEA LEVEL

BAROMETRIC PRESSURE:

101.325 kPa

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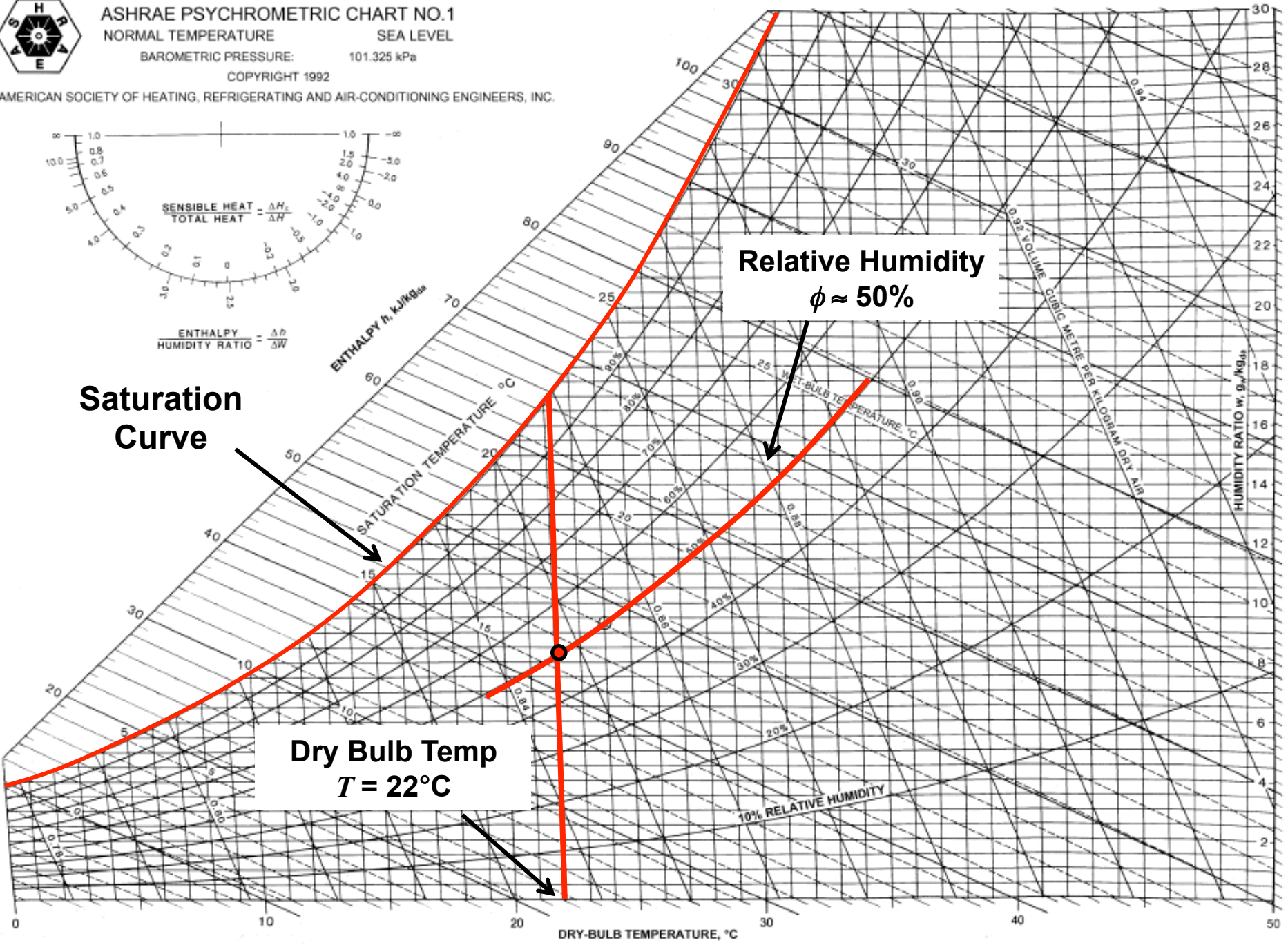
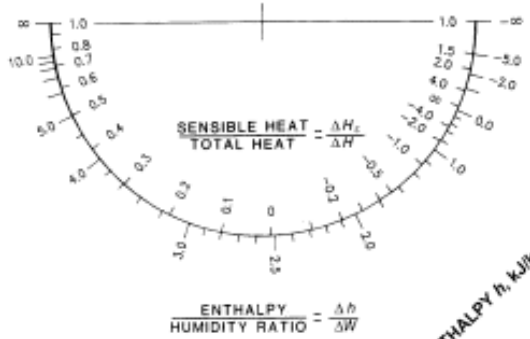
Saturation Curve





ASHRAE PSYCHROMETRIC CHART NO.1
NORMAL TEMPERATURE SEA LEVEL
BAROMETRIC PRESSURE: 101.325 kPa
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Saturation Curve

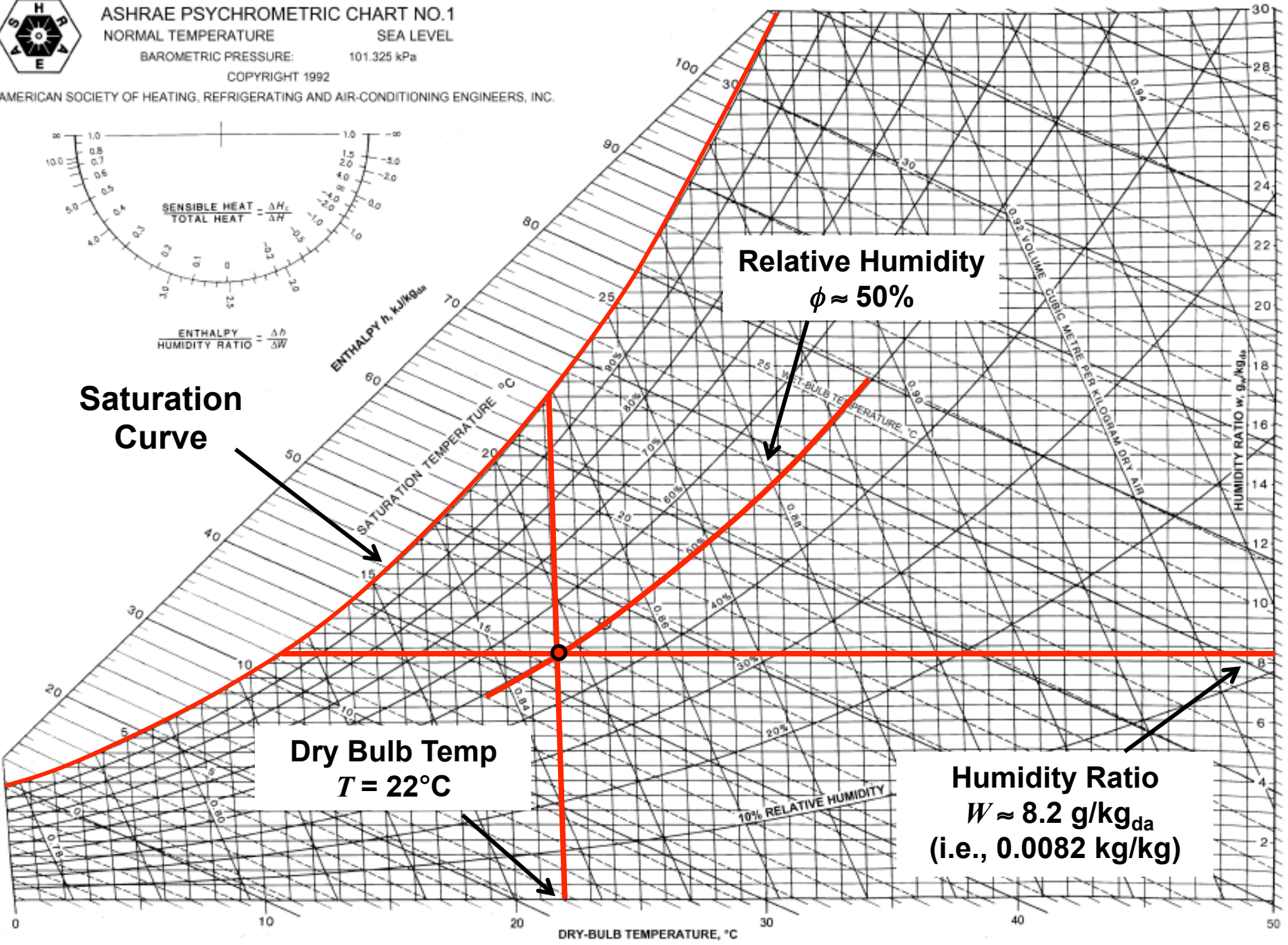
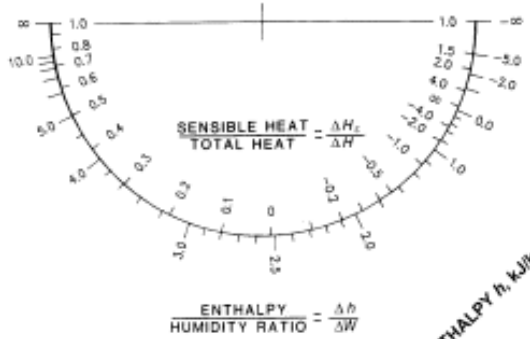
Dry Bulb Temp
 $T = 22^\circ\text{C}$

Relative Humidity
 $\phi \approx 50\%$



ASHRAE PSYCHROMETRIC CHART NO.1
 NORMAL TEMPERATURE SEA LEVEL
 BAROMETRIC PRESSURE: 101.325 kPa
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Saturation Curve

Dry Bulb Temp
 $T = 22^\circ\text{C}$

Relative Humidity
 $\phi \approx 50\%$

Humidity Ratio
 $W \approx 8.2 \text{ g/kg}_{da}$
 (i.e., 0.0082 kg/kg)



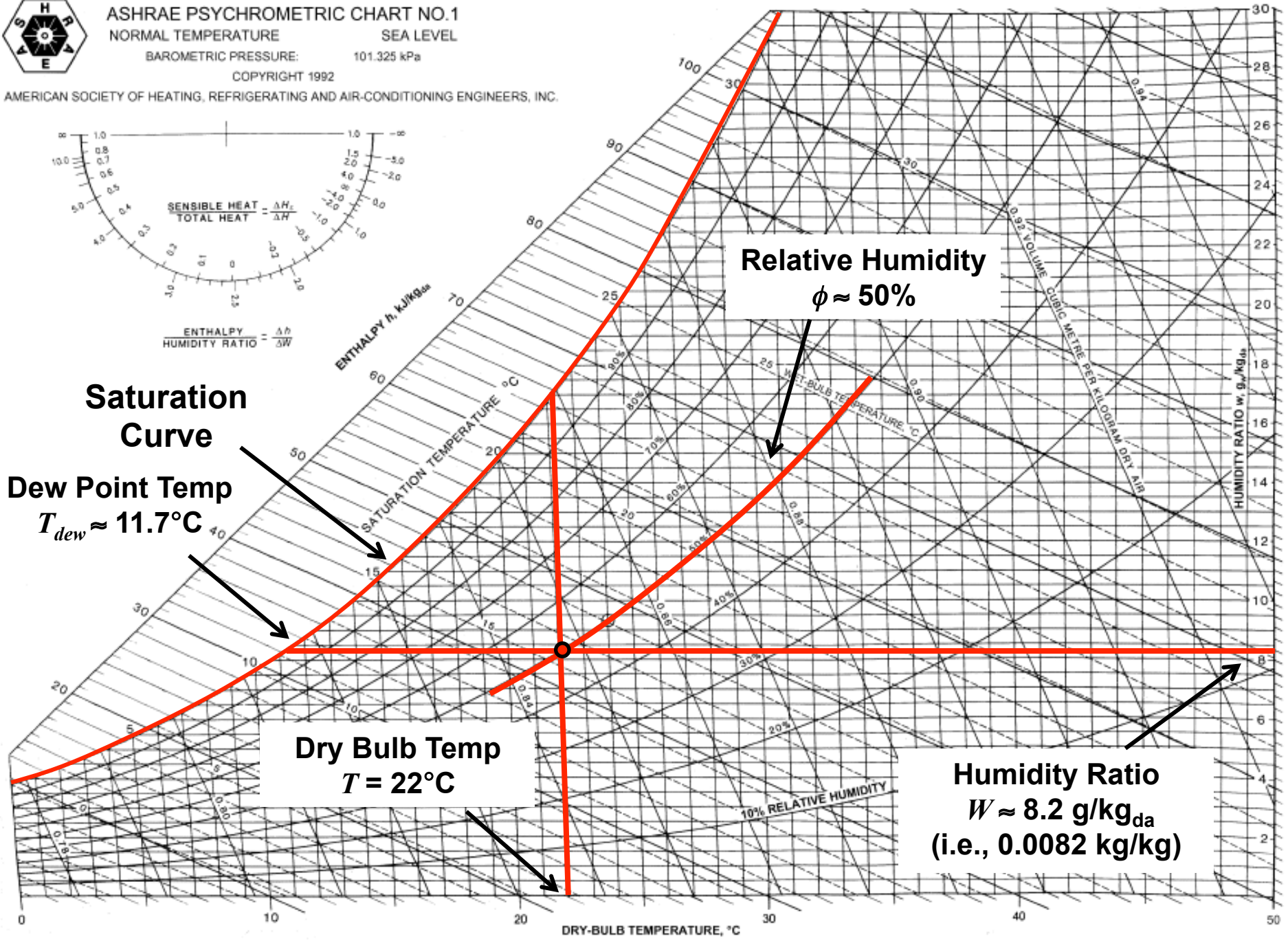
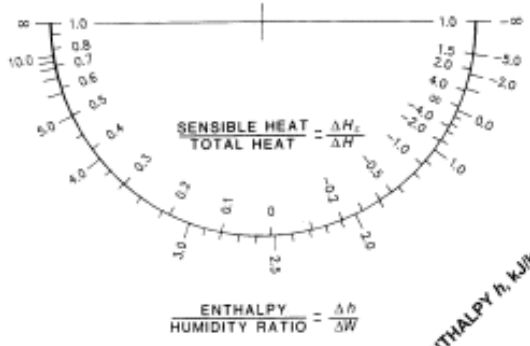
ASHRAE PSYCHROMETRIC CHART NO.1

NORMAL TEMPERATURE SEA LEVEL

BAROMETRIC PRESSURE: 101.325 kPa

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Saturation Curve
Dew Point Temp
 $T_{dew} \approx 11.7^\circ\text{C}$

Dry Bulb Temp
 $T = 22^\circ\text{C}$

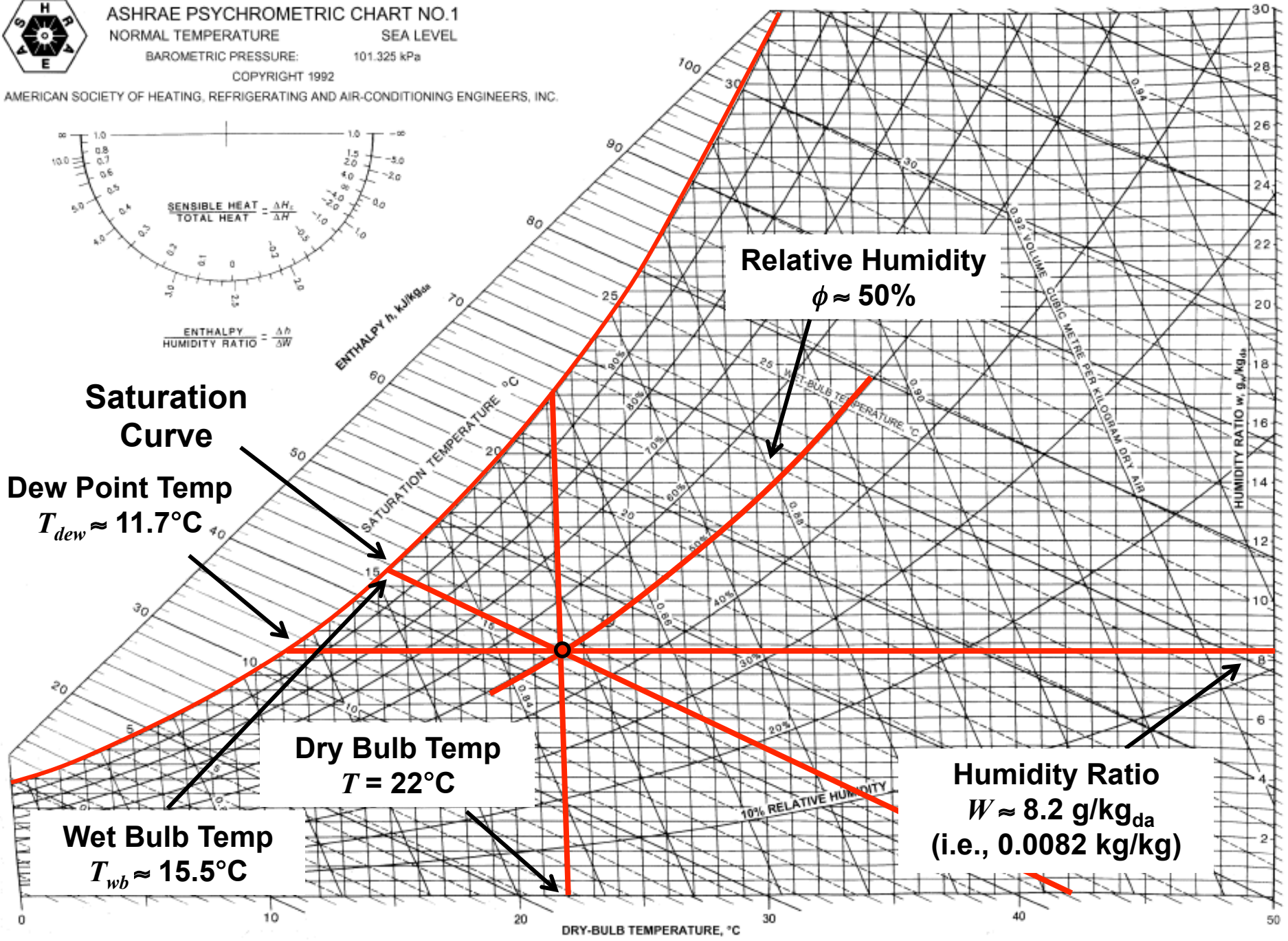
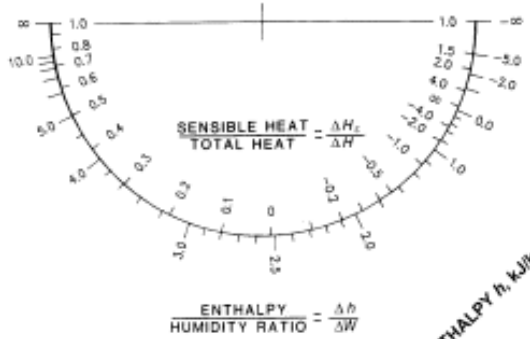
Relative Humidity
 $\phi \approx 50\%$

Humidity Ratio
 $W \approx 8.2 \text{ g/kg}_{da}$
 (i.e., 0.0082 kg/kg)



ASHRAE PSYCHROMETRIC CHART NO.1
 NORMAL TEMPERATURE SEA LEVEL
 BAROMETRIC PRESSURE: 101.325 kPa
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Relative Humidity
 $\phi \approx 50\%$

Saturation Curve
 Dew Point Temp
 $T_{dew} \approx 11.7^\circ\text{C}$

Dry Bulb Temp
 $T = 22^\circ\text{C}$

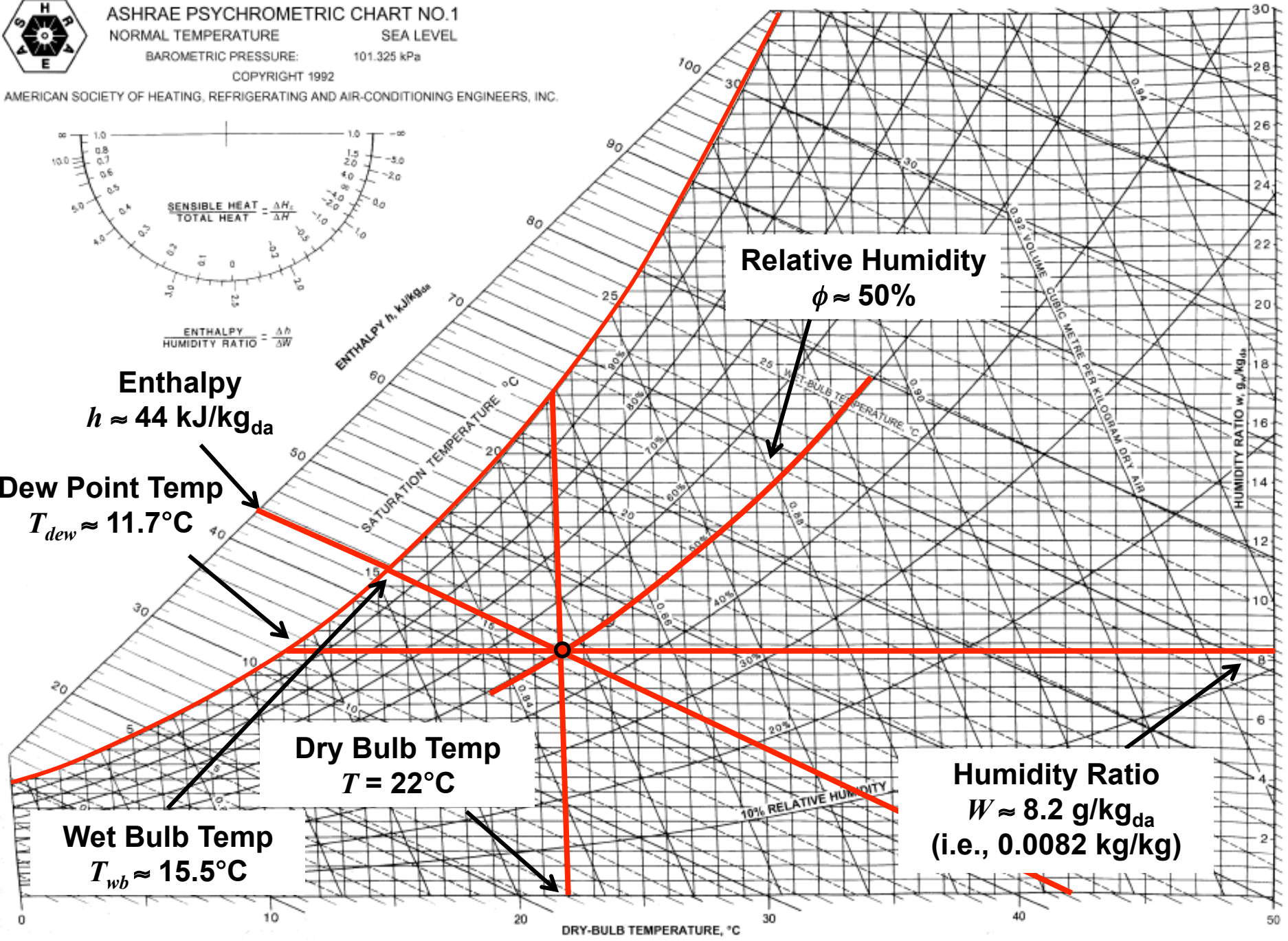
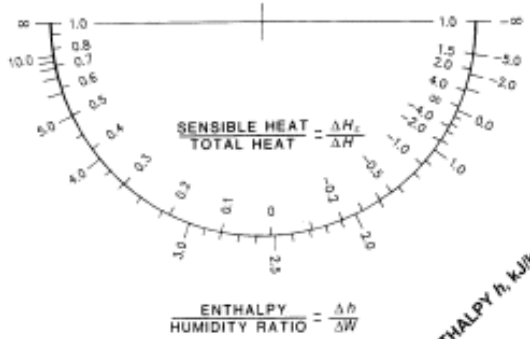
Wet Bulb Temp
 $T_{wb} \approx 15.5^\circ\text{C}$

Humidity Ratio
 $W \approx 8.2 \text{ g/kg}_{da}$
 (i.e., 0.0082 kg/kg)



ASHRAE PSYCHROMETRIC CHART NO.1
NORMAL TEMPERATURE SEA LEVEL
BAROMETRIC PRESSURE: 101.325 kPa
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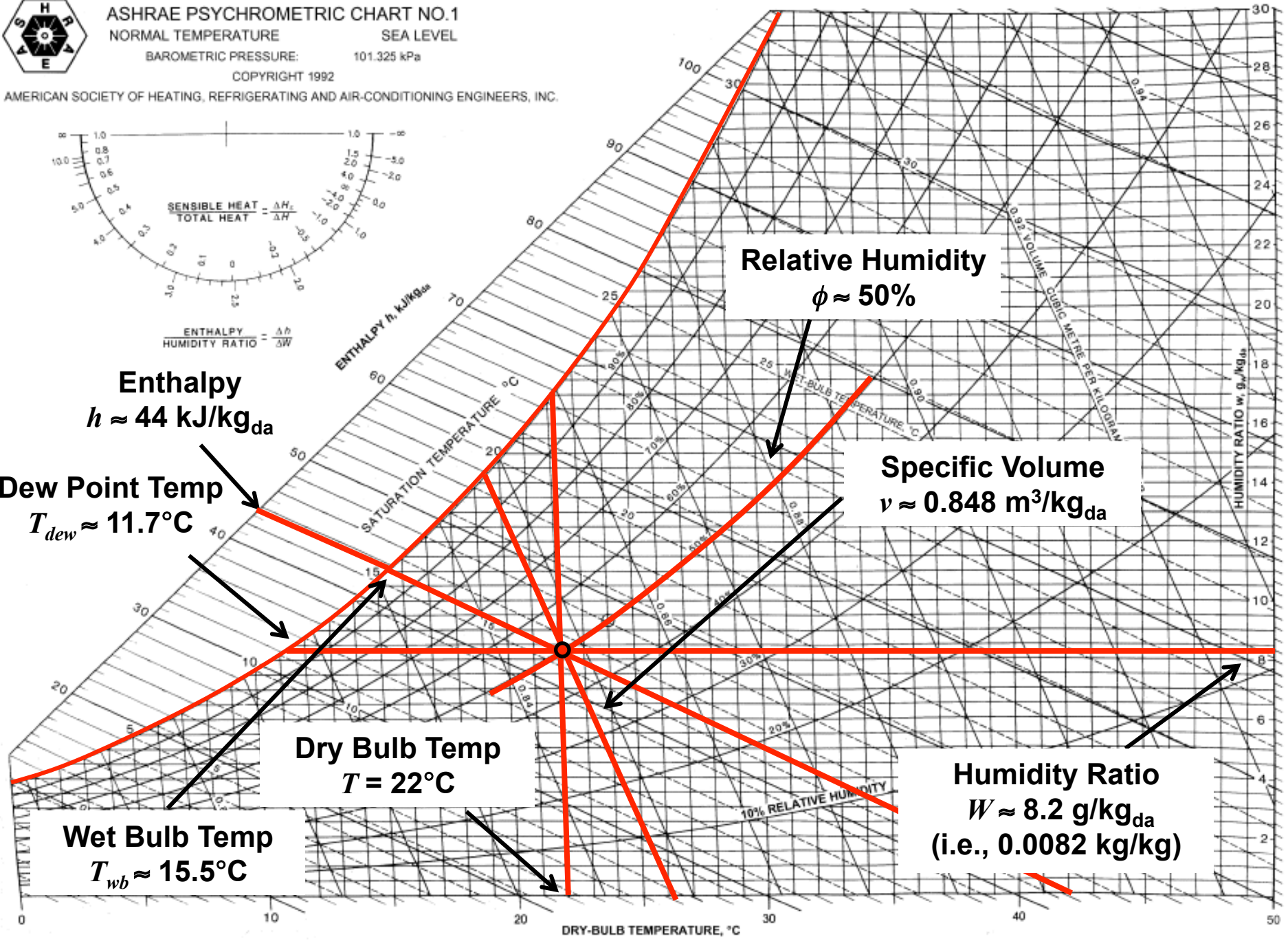
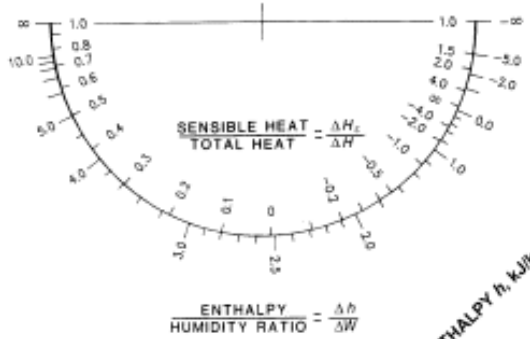
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Relative Humidity
 $\phi \approx 50\%$

Specific Volume
 $v \approx 0.848 \text{ m}^3/\text{kg}_{da}$

Humidity Ratio
 $W \approx 8.2 \text{ g}/\text{kg}_{da}$
 (i.e., 0.0082 kg/kg)

Dry Bulb Temp
 $T = 22^\circ\text{C}$

Wet Bulb Temp
 $T_{wb} \approx 15.5^\circ\text{C}$

Dew Point Temp
 $T_{dew} \approx 11.7^\circ\text{C}$

Enthalpy
 $h \approx 44 \text{ kJ}/\text{kg}_{da}$

DRY-BULB TEMPERATURE, °C

HUMIDITY RATIO w , g/kg_{da}

ENTHALPY h , kJ/kg_{da}

SATURATION TEMPERATURE °C

WET-BULB TEMPERATURE, °C

10% RELATIVE HUMIDITY

0.92 VOLUME CUBIC METRE PER KILOGRAM



ASHRAE PSYCHROMETRIC CHART NO.1

NORMAL TEMPERATURE

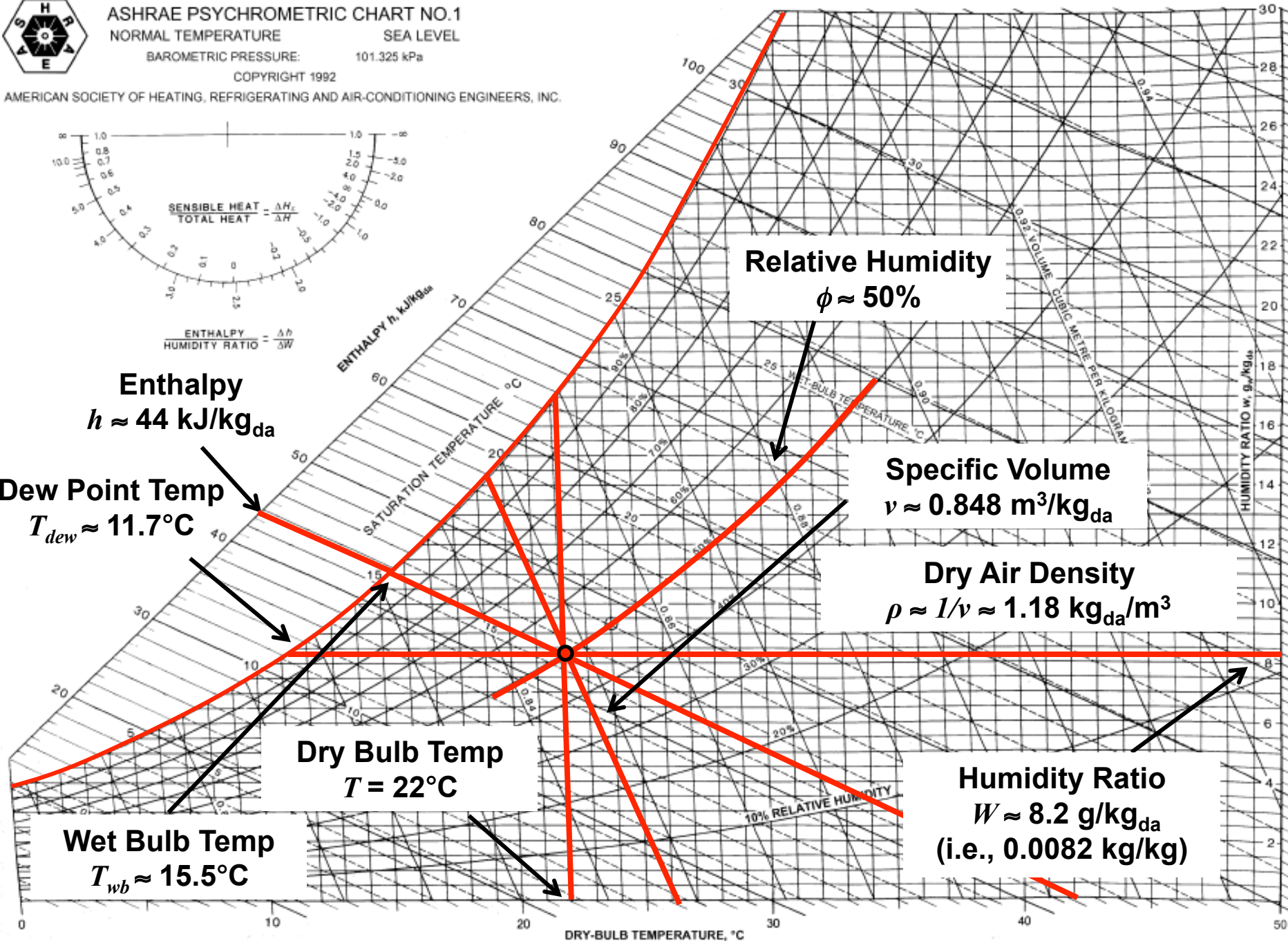
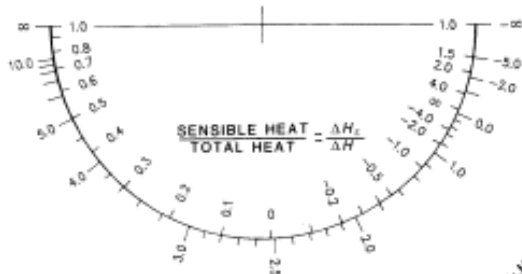
SEA LEVEL

BAROMETRIC PRESSURE:

101.325 kPa

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Relative Humidity
 $\phi \approx 50\%$

Specific Volume
 $v \approx 0.848 \text{ m}^3/\text{kg}_{da}$

Dry Air Density
 $\rho \approx 1/v \approx 1.18 \text{ kg}_{da}/\text{m}^3$

Humidity Ratio
 $W \approx 8.2 \text{ g}/\text{kg}_{da}$
(i.e., 0.0082 kg/kg)

Enthalpy
 $h \approx 44 \text{ kJ}/\text{kg}_{da}$

Dew Point Temp
 $T_{dew} \approx 11.7^\circ\text{C}$

Dry Bulb Temp
 $T = 22^\circ\text{C}$

Wet Bulb Temp
 $T_{wb} \approx 15.5^\circ\text{C}$

Some psychrometric examples

Moist air exists at 30°C dry-bulb temperature with a 15°C dew point temperature

Find the following:

- (a) the humidity ratio, W
- (b) wet-bulb temperature, T_{wb}
- (c) enthalpy, h
- (d) specific volume, v
- (e) relative humidity, ϕ



ASHRAE PSYCHROMETRIC CHART NO.1

NORMAL TEMPERATURE

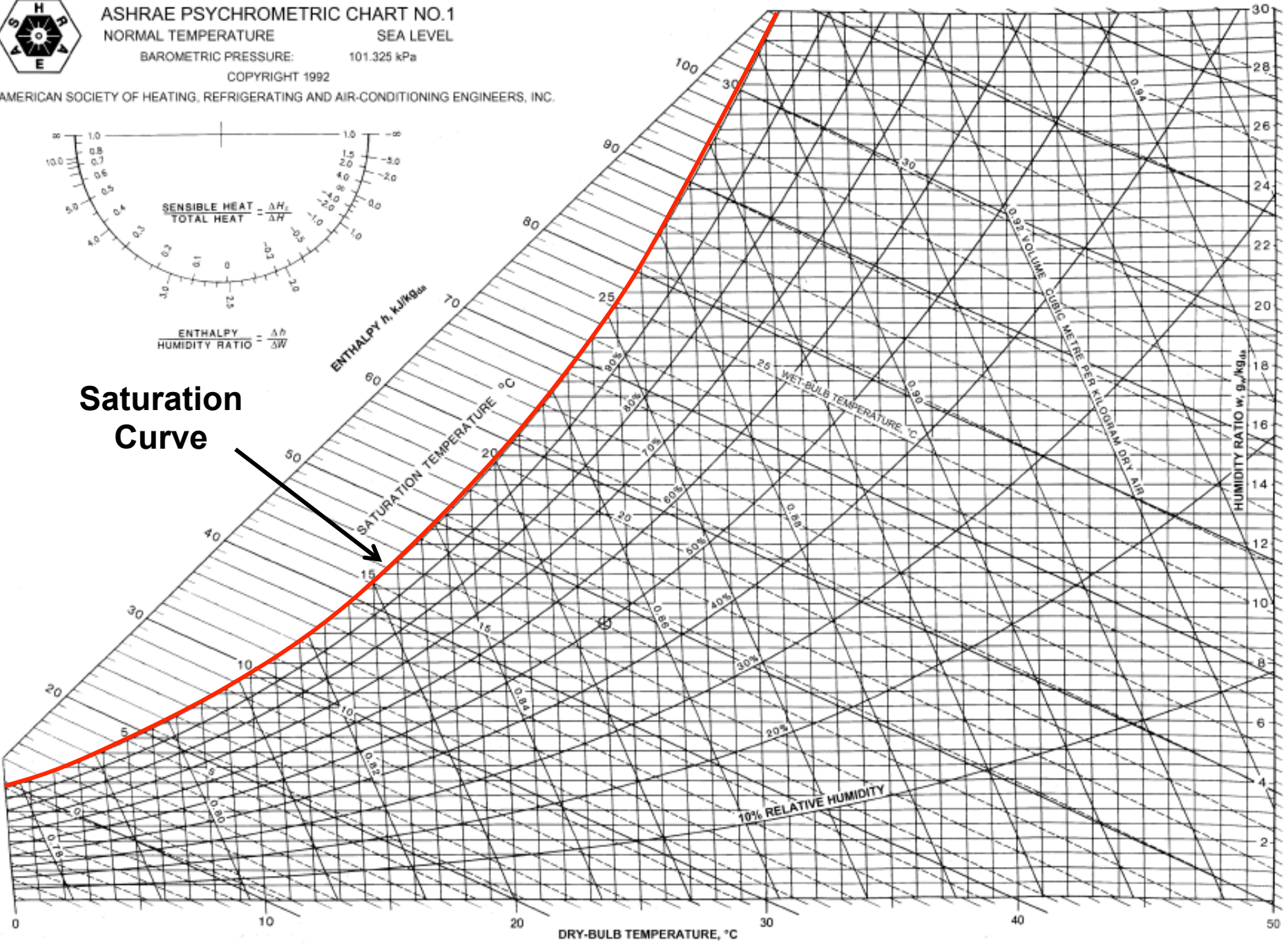
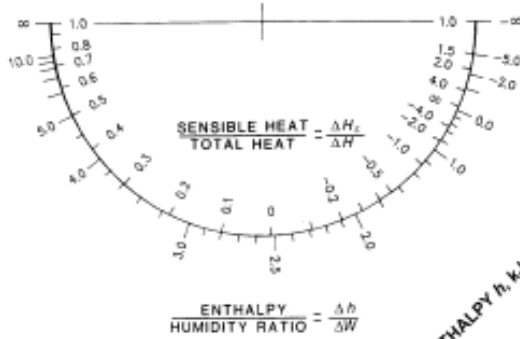
SEA LEVEL

BAROMETRIC PRESSURE:

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Saturation Curve





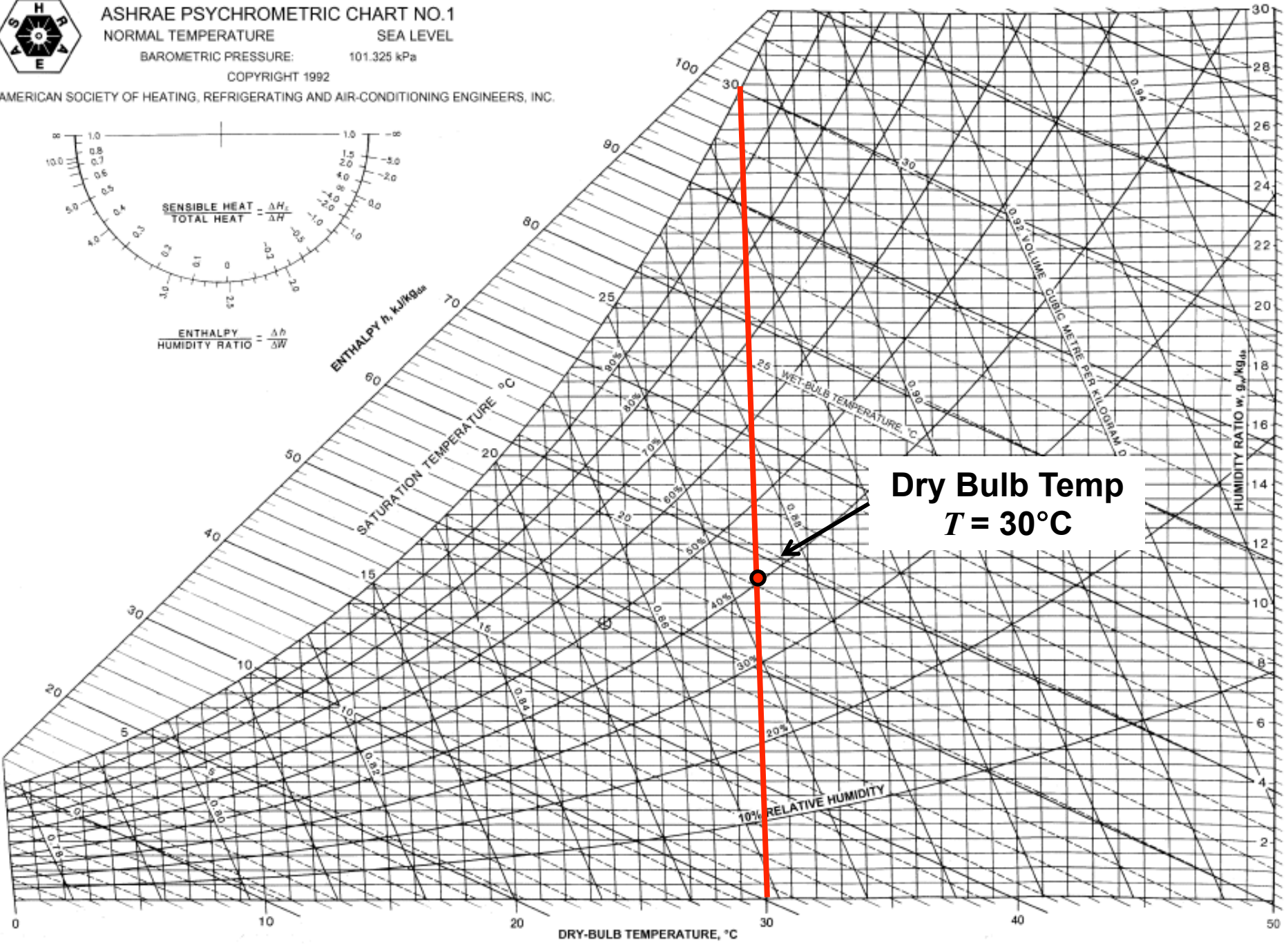
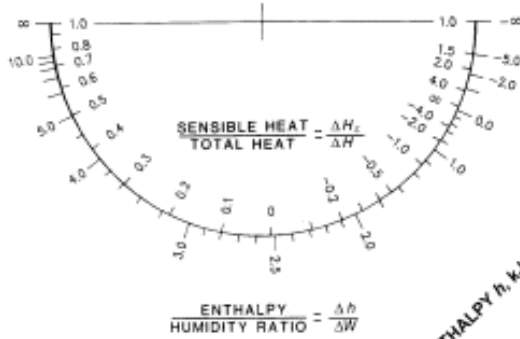
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NORMAL TEMPERATURE SEA LEVEL

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Dry Bulb Temp
 $T = 30^{\circ}\text{C}$





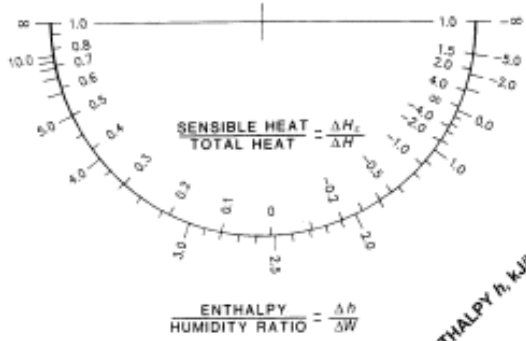
ASHRAE PSYCHROMETRIC CHART NO.1

NORMAL TEMPERATURE SEA LEVEL

BAROMETRIC PRESSURE: 101.325 kPa

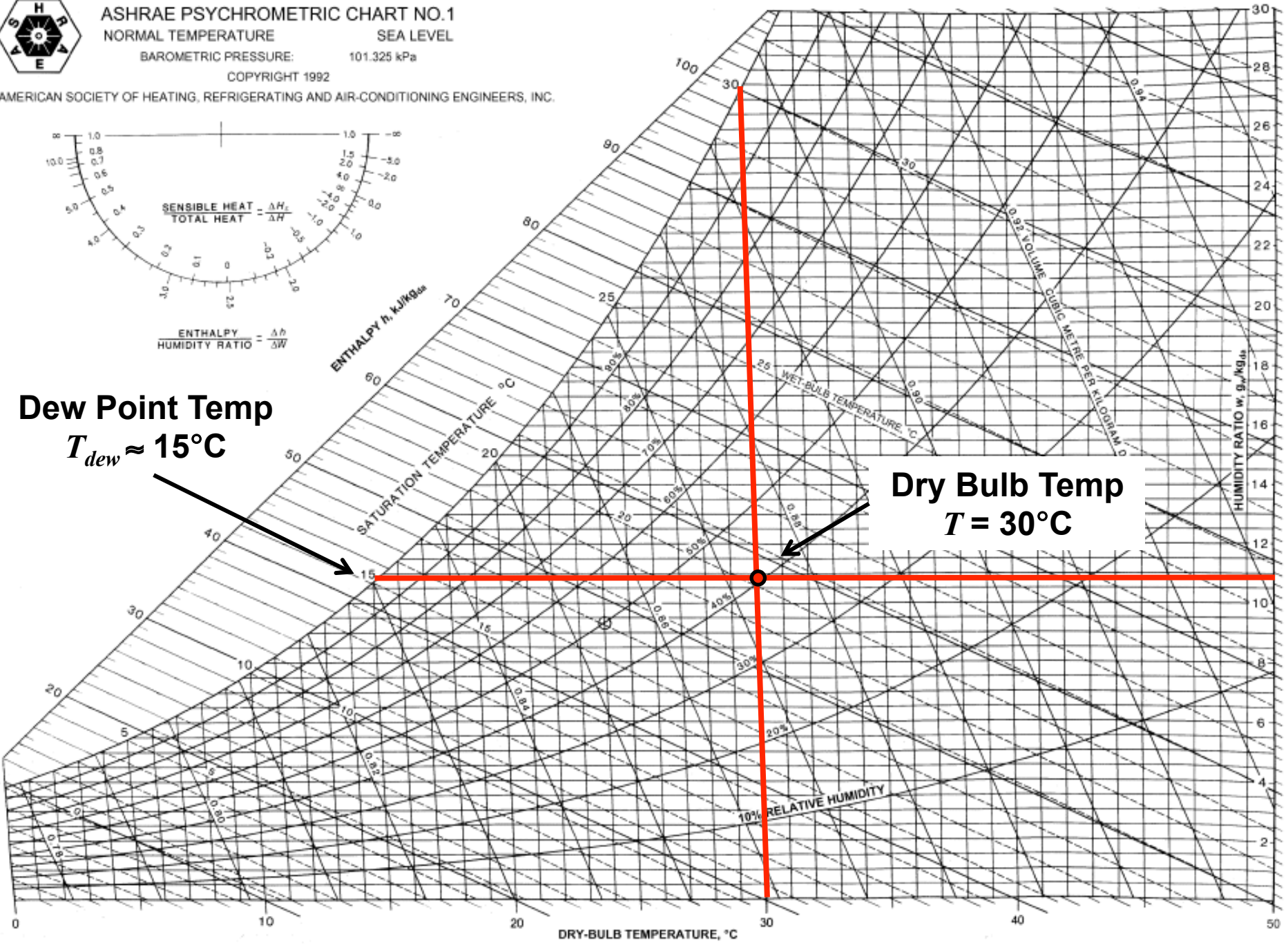
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Dew Point Temp
 $T_{dew} \approx 15^\circ\text{C}$

Dry Bulb Temp
 $T = 30^\circ\text{C}$





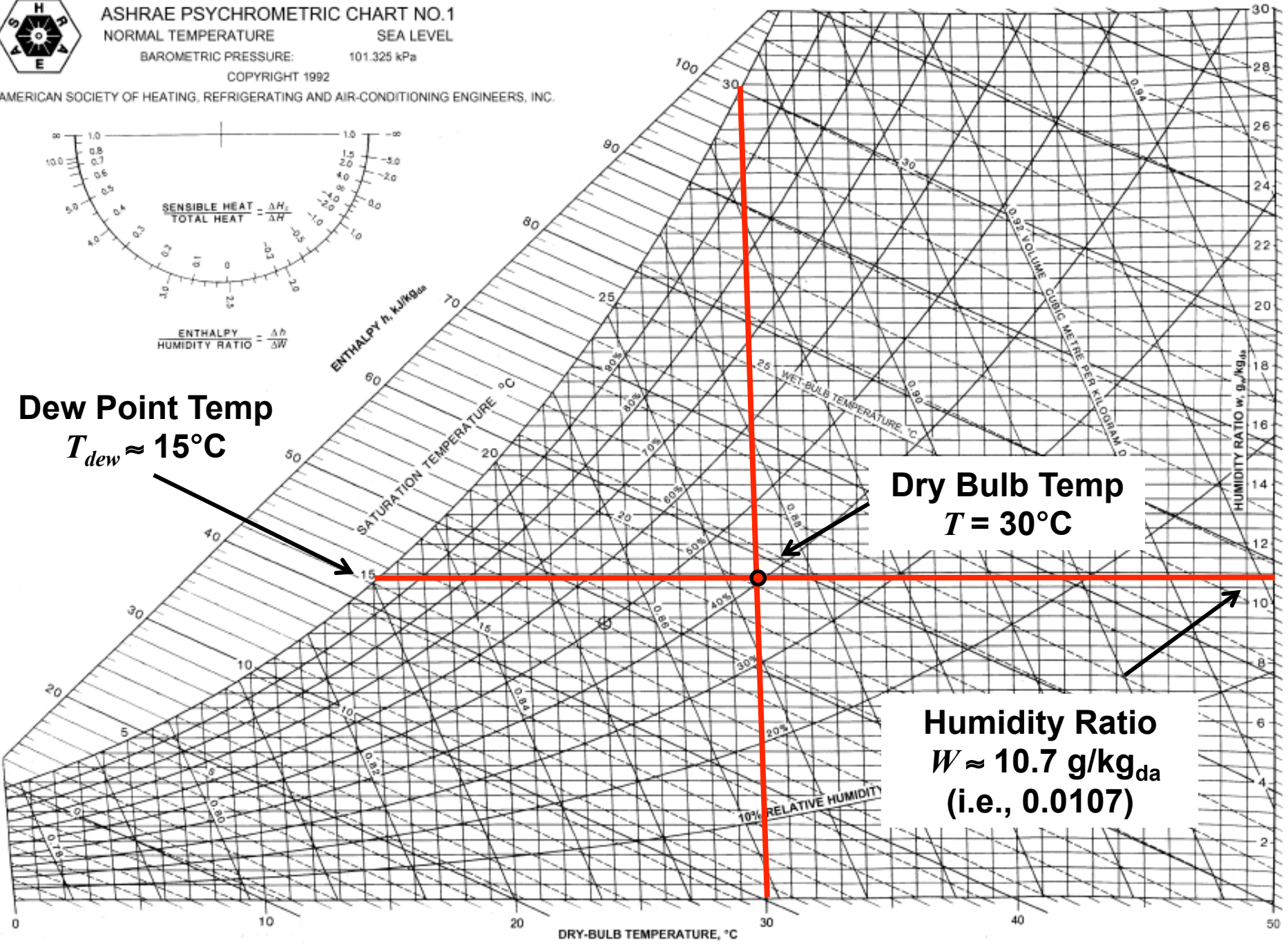
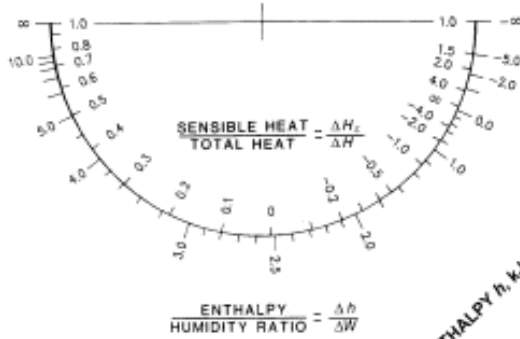
ASHRAE PSYCHROMETRIC CHART NO.1

NORMAL TEMPERATURE SEA LEVEL

BAROMETRIC PRESSURE: 101.325 kPa

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Dew Point Temp
 $T_{dew} \approx 15^\circ\text{C}$

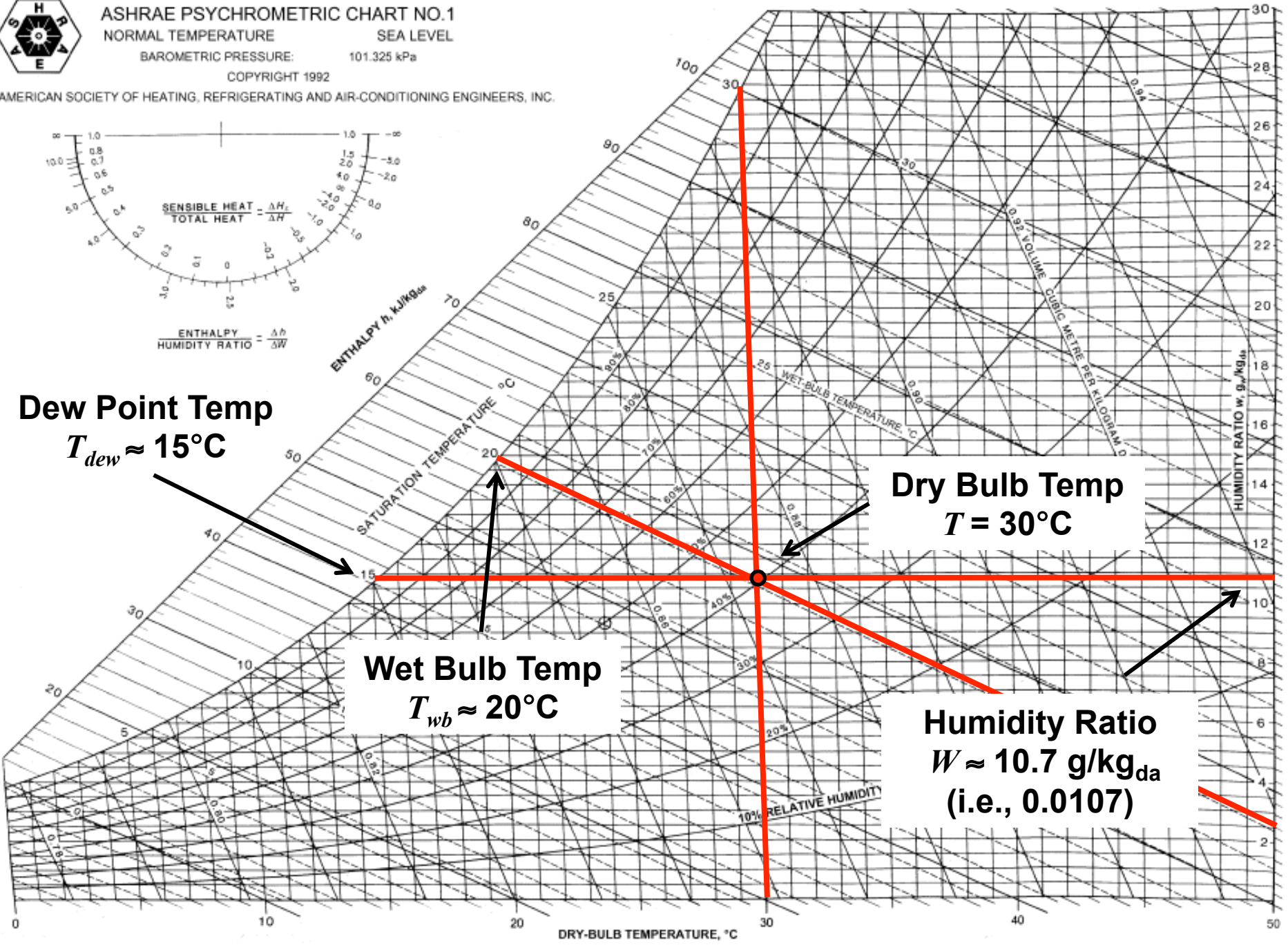
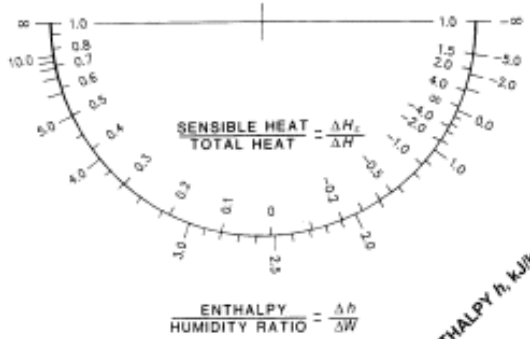
Dry Bulb Temp
 $T = 30^\circ\text{C}$

Humidity Ratio
 $W \approx 10.7 \text{ g/kg}_{da}$
(i.e., 0.0107)



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NORMAL TEMPERATURE SEA LEVEL
BAROMETRIC PRESSURE: 101.325 kPa
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Dew Point Temp
 $T_{dew} \approx 15^\circ\text{C}$

Dry Bulb Temp
 $T = 30^\circ\text{C}$

Wet Bulb Temp
 $T_{wb} \approx 20^\circ\text{C}$

Humidity Ratio
 $W \approx 10.7 \text{ g/kg}_{da}$
(i.e., 0.0107)



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 NORMAL TEMPERATURE SEA LEVEL
 BAROMETRIC PRESSURE: 101.325 kPa
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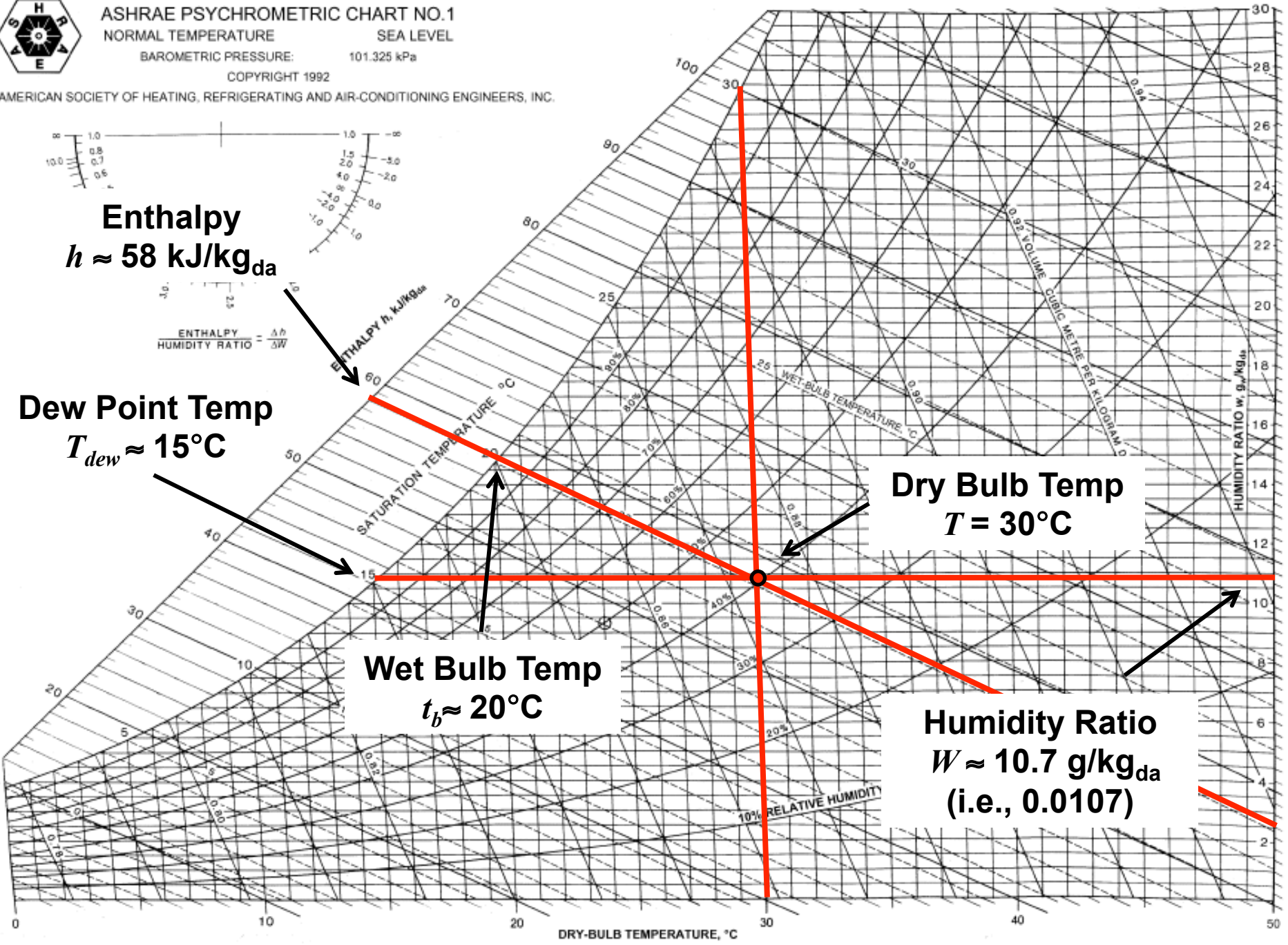
Enthalpy
 $h \approx 58 \text{ kJ/kg}_{da}$

Dew Point Temp
 $T_{dew} \approx 15^\circ\text{C}$

Wet Bulb Temp
 $t_b \approx 20^\circ\text{C}$

Dry Bulb Temp
 $T = 30^\circ\text{C}$

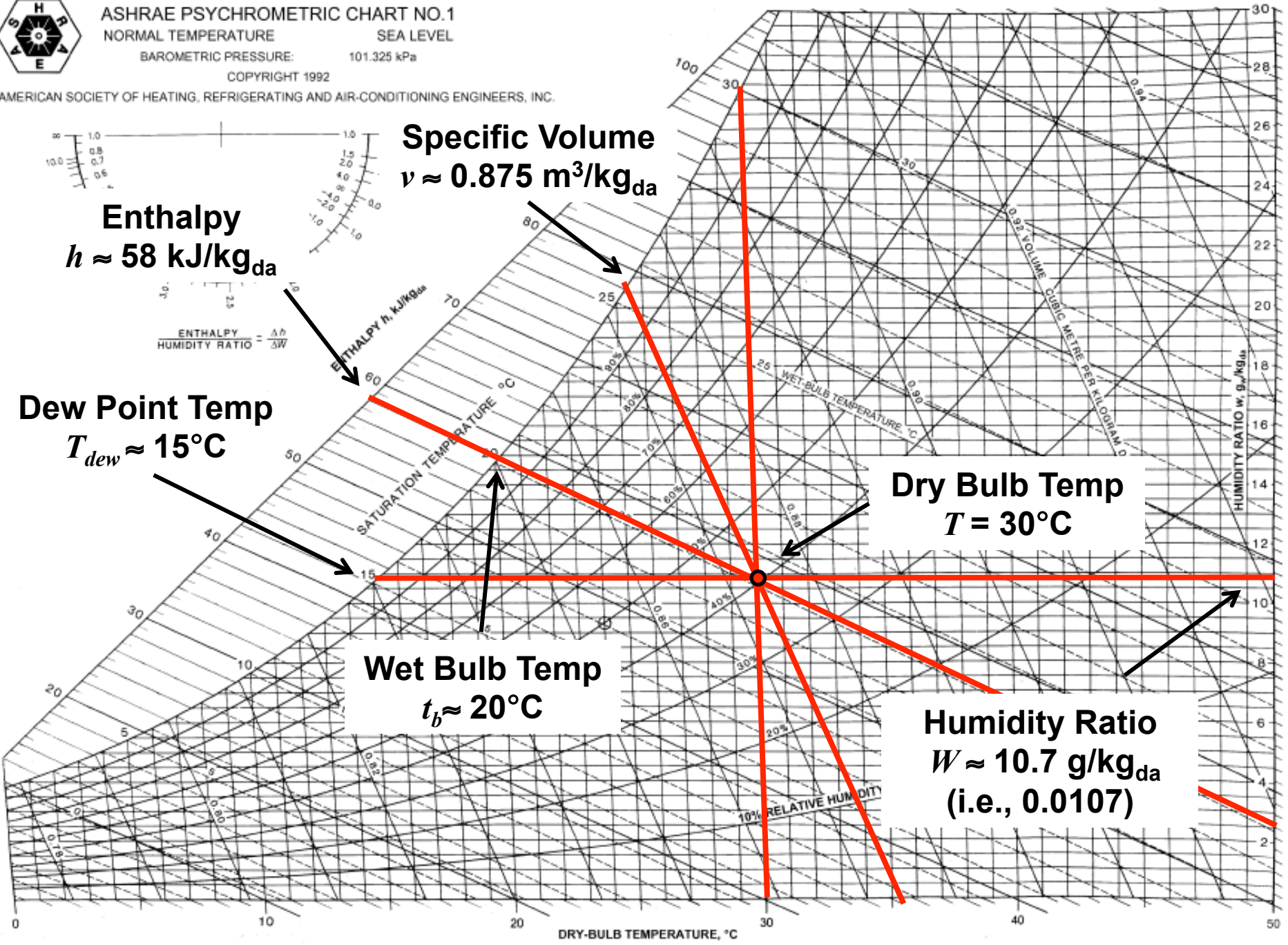
Humidity Ratio
 $W \approx 10.7 \text{ g/kg}_{da}$
 (i.e., 0.0107)





ASHRAE PSYCHROMETRIC CHART NO.1
 NORMAL TEMPERATURE SEA LEVEL
 BAROMETRIC PRESSURE: 101.325 kPa
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Specific Volume
 $v \approx 0.875 \text{ m}^3/\text{kg}_{da}$

Enthalpy
 $h \approx 58 \text{ kJ/kg}_{da}$

Dew Point Temp
 $T_{dew} \approx 15^\circ\text{C}$

Dry Bulb Temp
 $T = 30^\circ\text{C}$

Wet Bulb Temp
 $t_b \approx 20^\circ\text{C}$

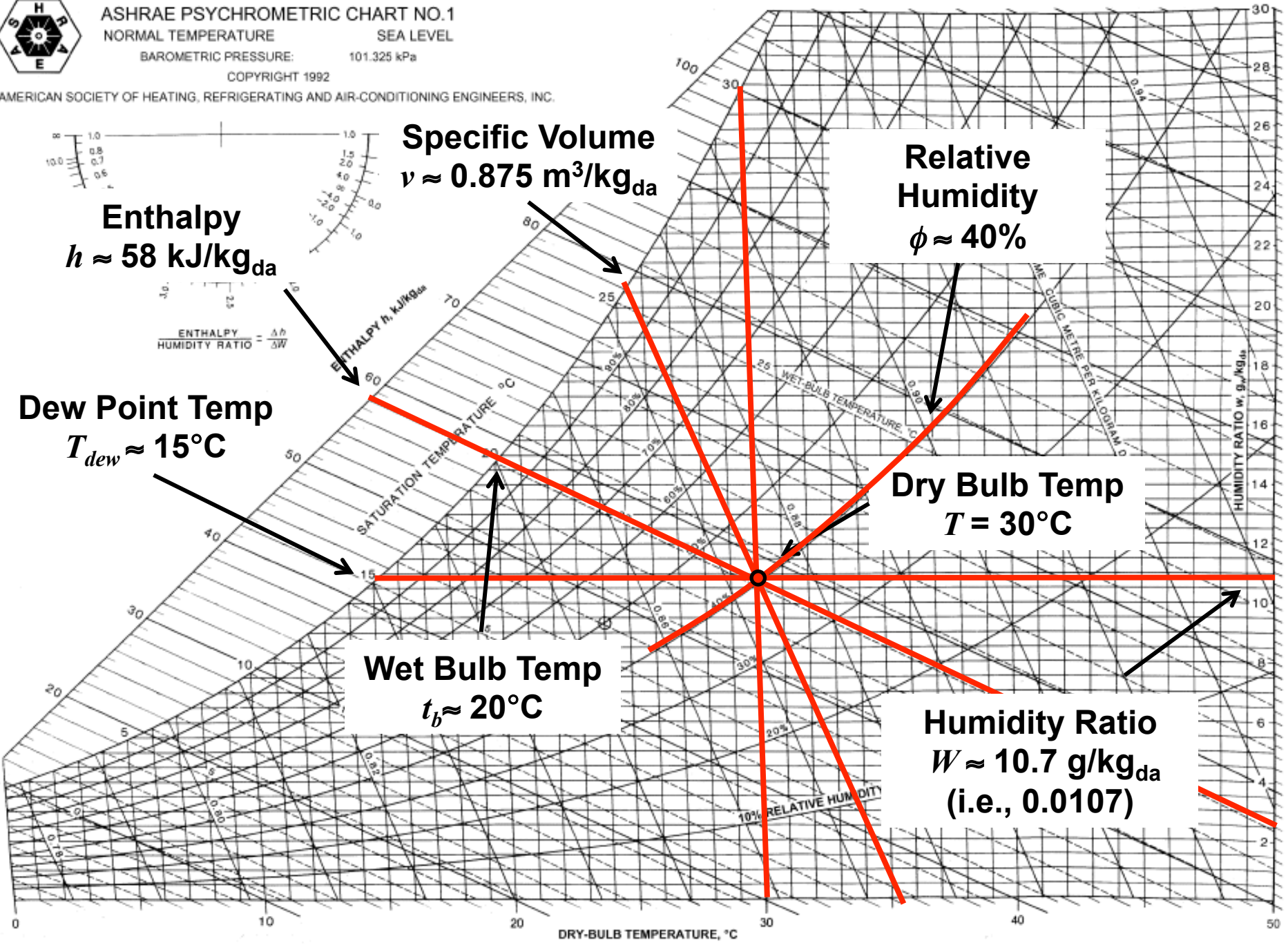
Humidity Ratio
 $W \approx 10.7 \text{ g/kg}_{da}$
 (i.e., 0.0107)

ENTHALPY
 $\frac{\Delta h}{\Delta W}$



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Dew Point Temp
 $T_{dew} \approx 15^\circ\text{C}$

Enthalpy
 $h \approx 58 \text{ kJ/kg}_{da}$

Specific Volume
 $v \approx 0.875 \text{ m}^3/\text{kg}_{da}$

Relative Humidity
 $\phi \approx 40\%$

Dry Bulb Temp
 $T = 30^\circ\text{C}$

Wet Bulb Temp
 $t_b \approx 20^\circ\text{C}$

Humidity Ratio
 $W \approx 10.7 \text{ g/kg}_{da}$
 (i.e., 0.0107)

ENTHALPY
 HUMIDITY RATIO = $\frac{\Delta h}{\Delta W}$

DRY-BULB TEMPERATURE, °C

HUMIDITY RATIO w , g_w/kg_{da}

30
28
26
24
22
20
18
16
14
12
10
8
6
4
2

0

10

20

30

40

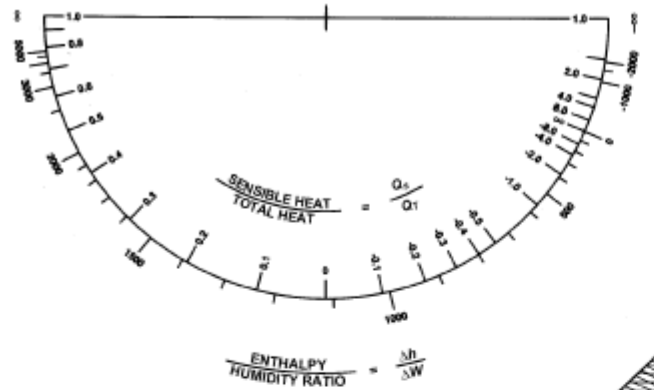
50

Psychrometrics: IP units example

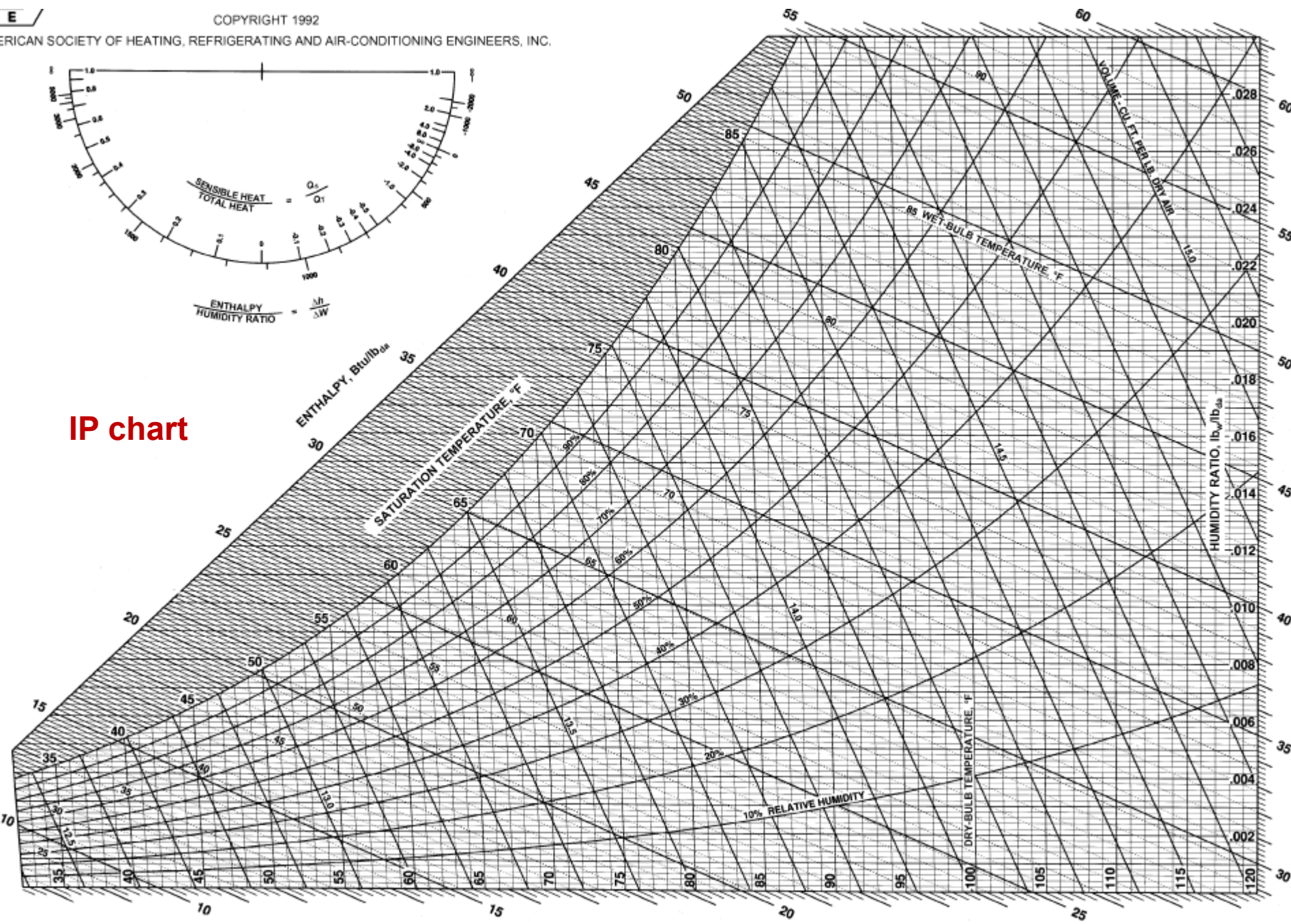
- Moist air exists at 100°F dry bulb, 65°F wet bulb and 14.696 psia

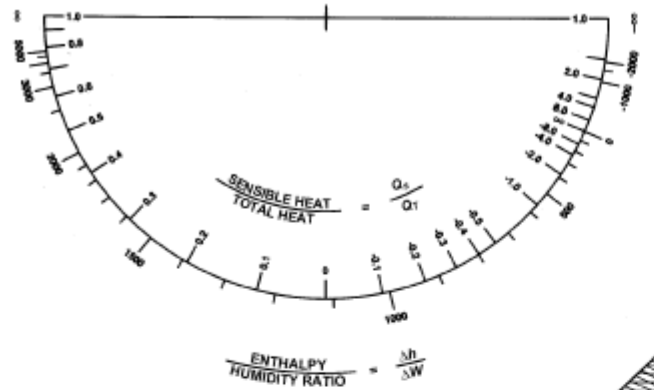
Find:

- a) Humidity ratio
- b) Enthalpy
- c) Dew-point temperature
- d) Relative humidity
- e) Specific volume

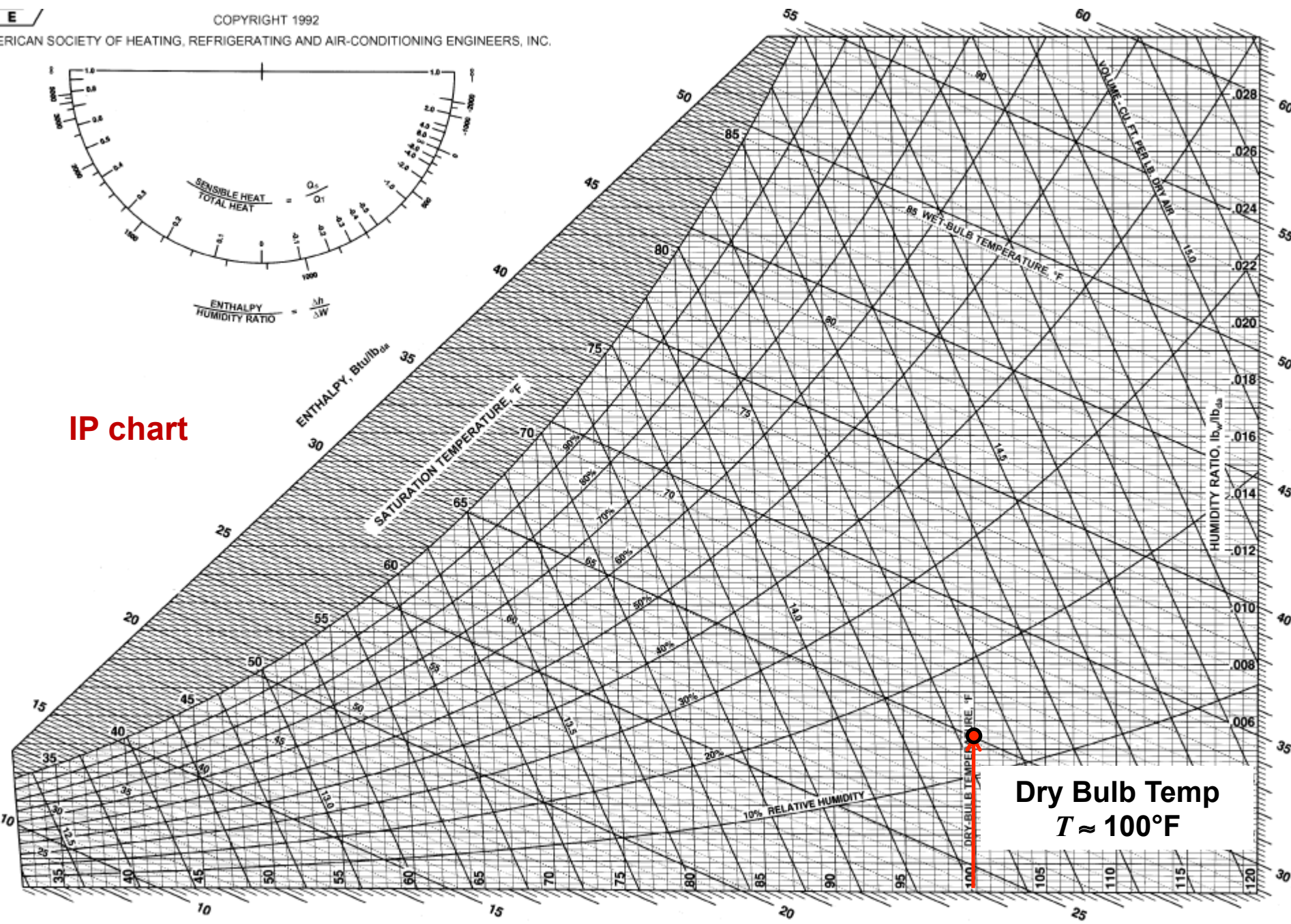


IP chart

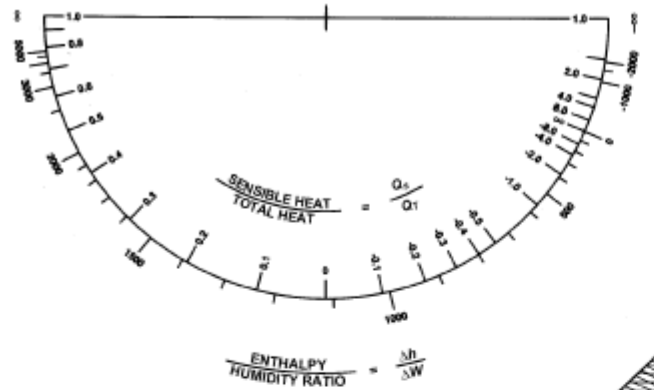




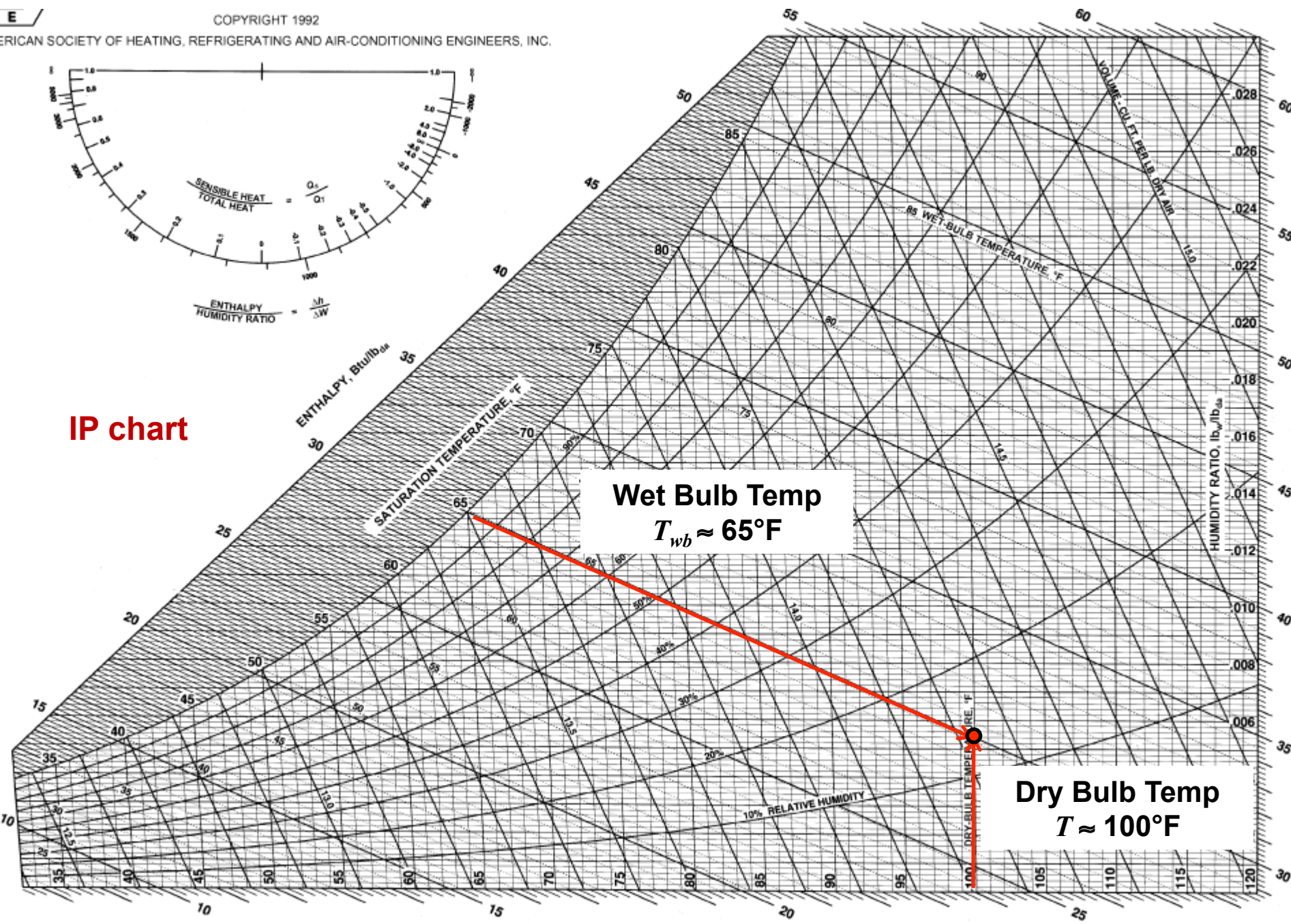
IP chart



Dry Bulb Temp
 $T \approx 100^\circ\text{F}$

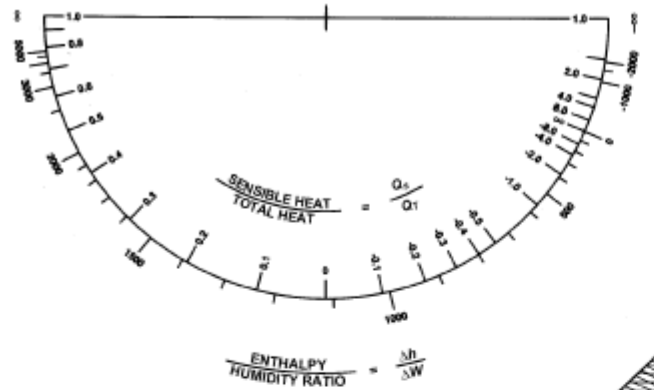


IP chart



Wet Bulb Temp
 $T_{wb} \approx 65^\circ\text{F}$

Dry Bulb Temp
 $T \approx 100^\circ\text{F}$

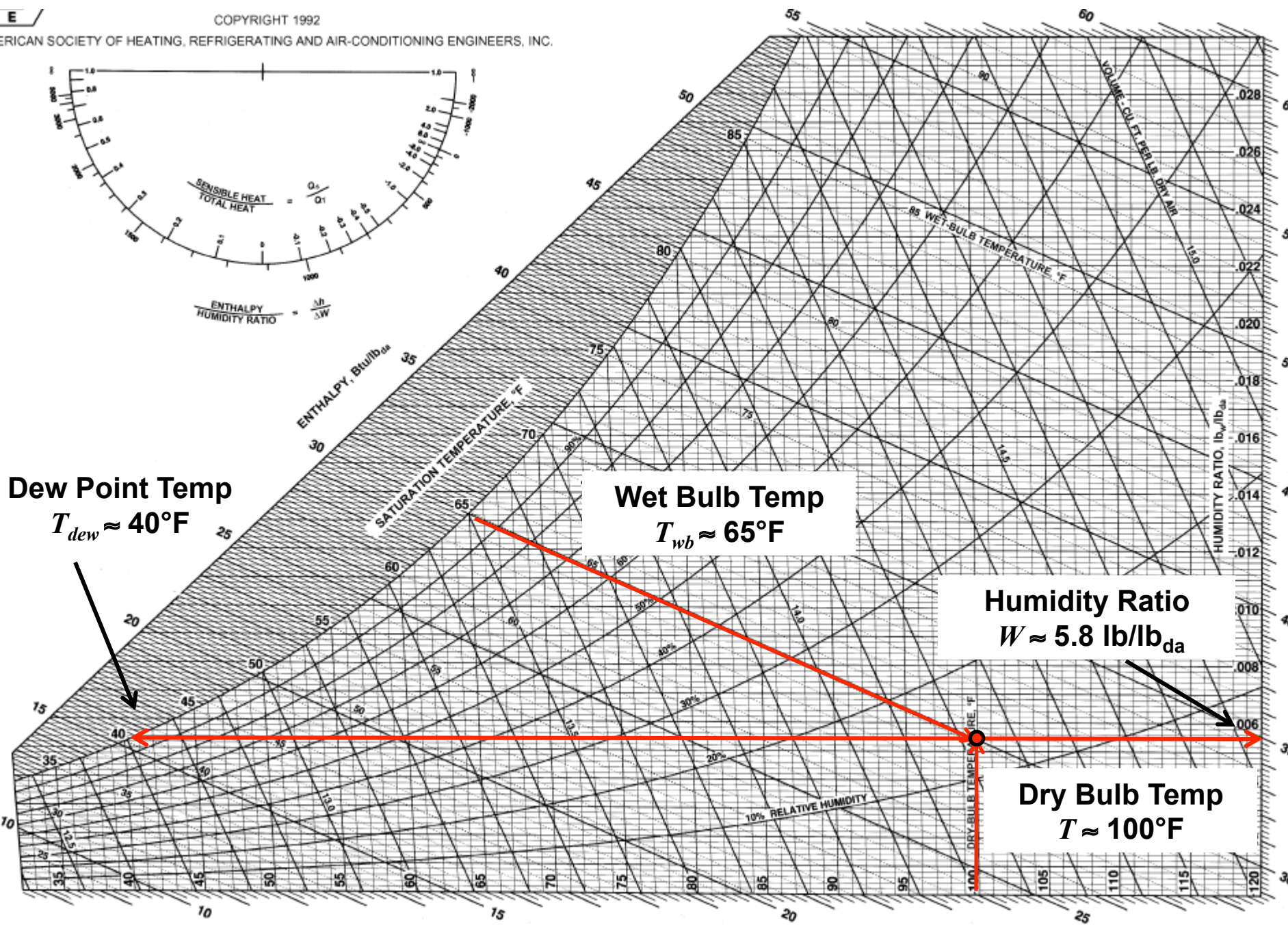


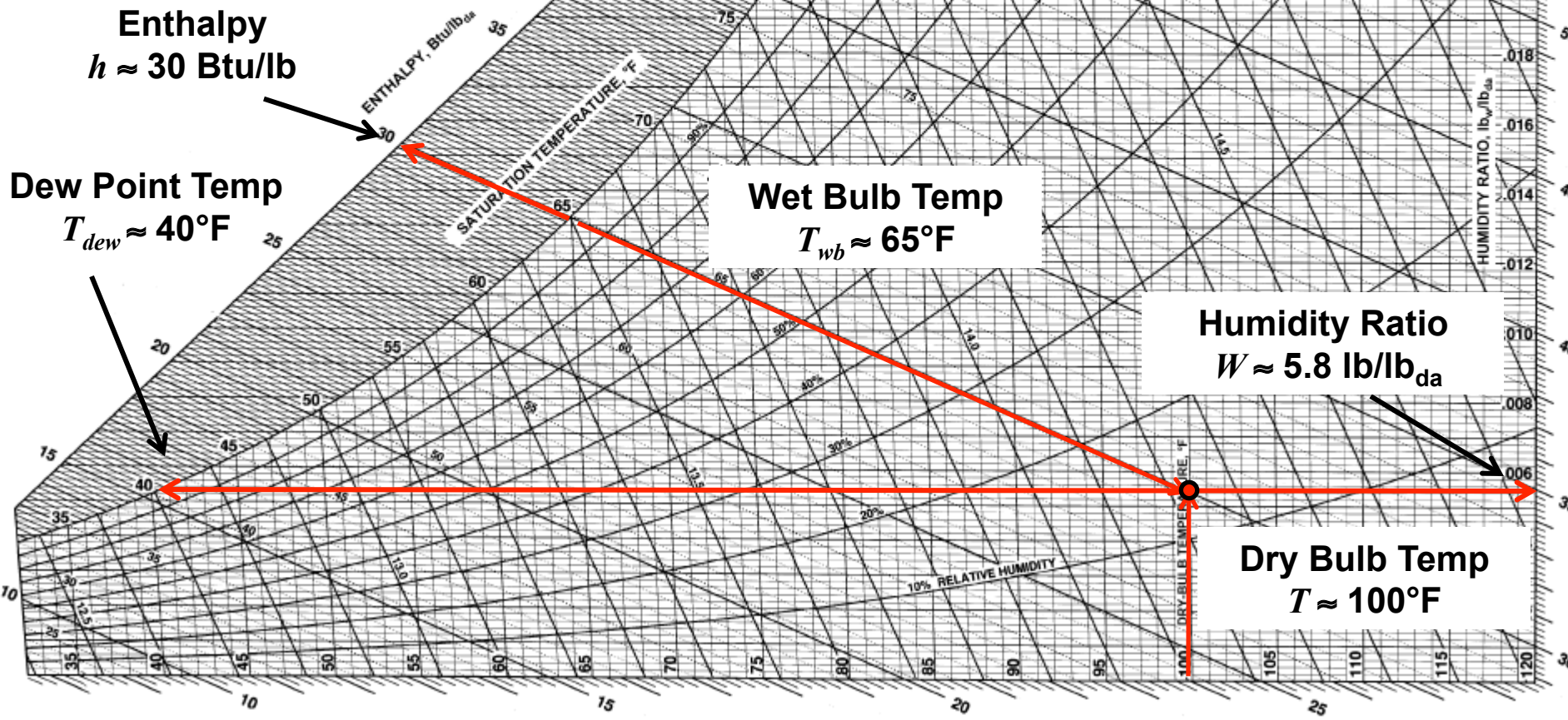
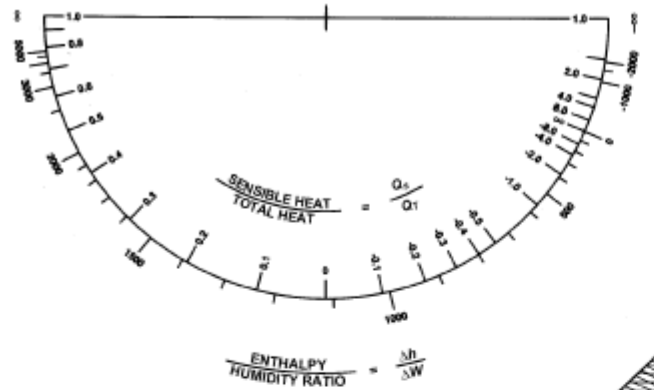
Dew Point Temp
 $T_{dew} \approx 40^\circ\text{F}$

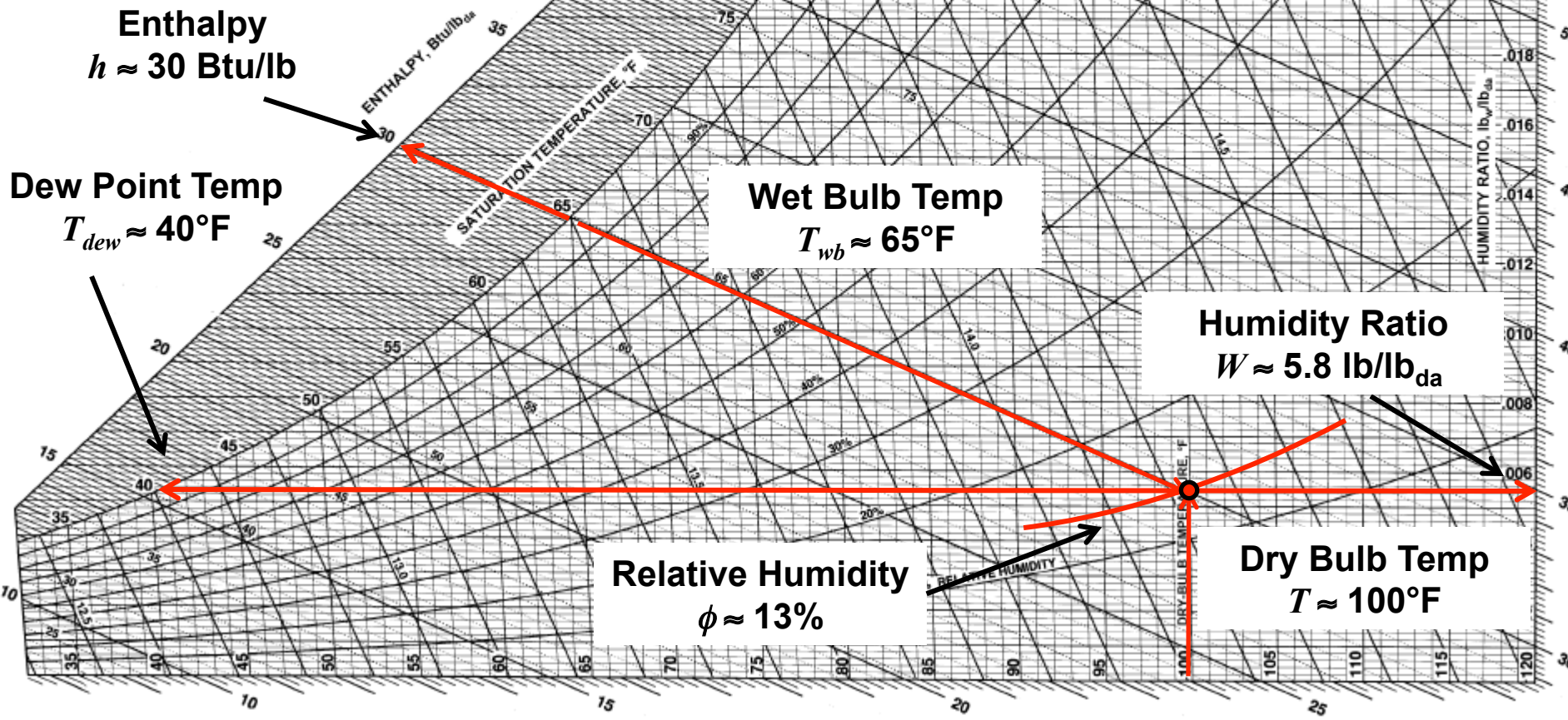
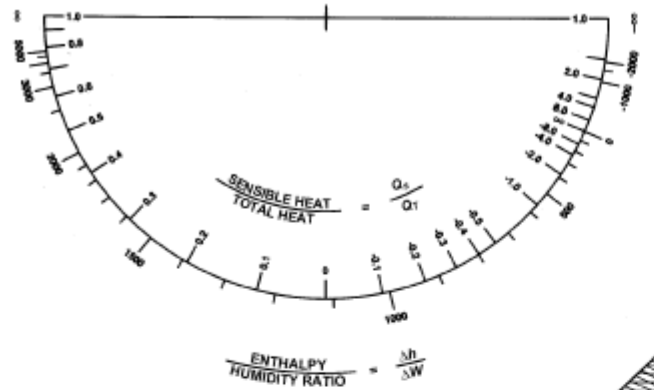
Wet Bulb Temp
 $T_{wb} \approx 65^\circ\text{F}$

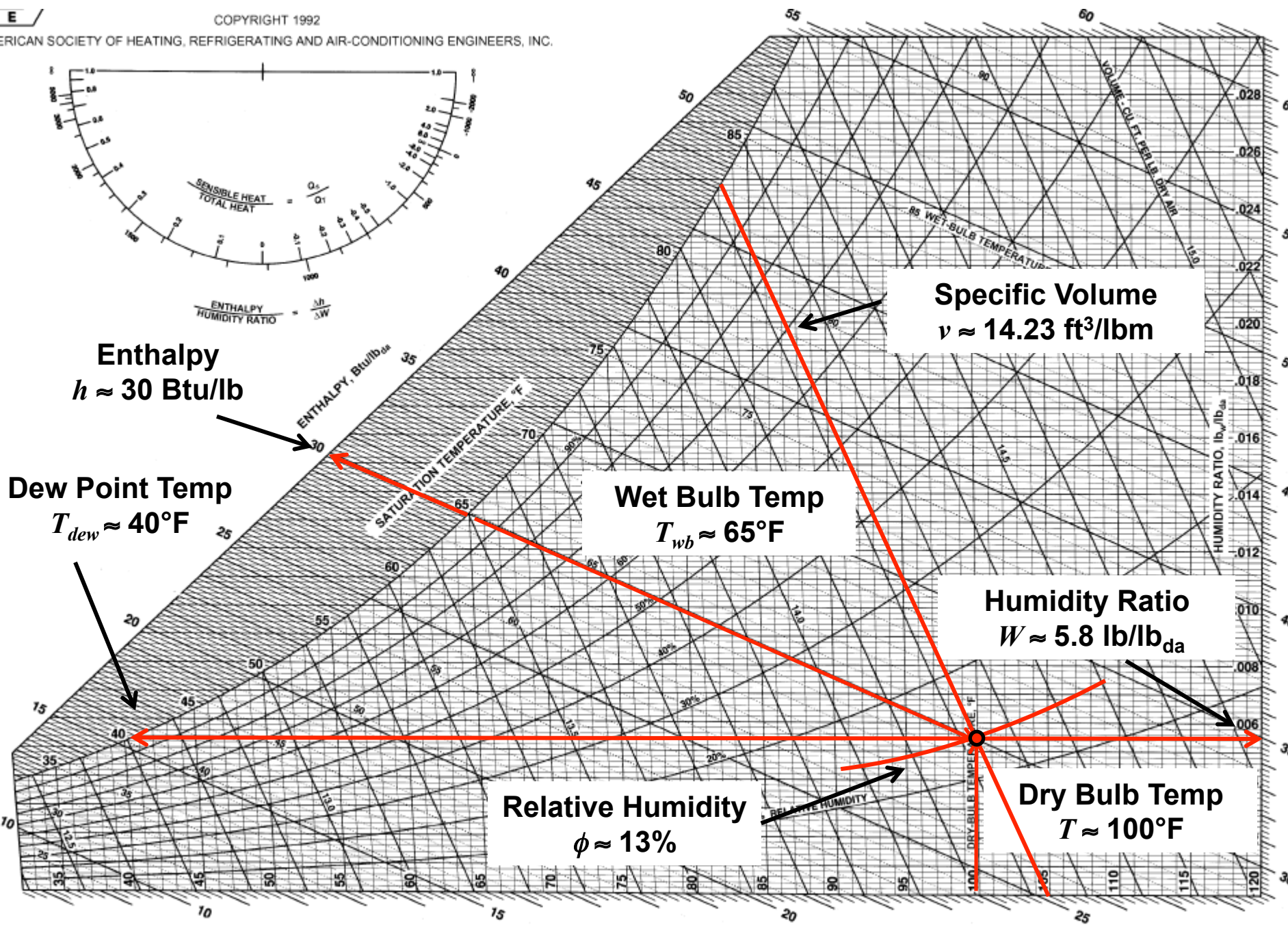
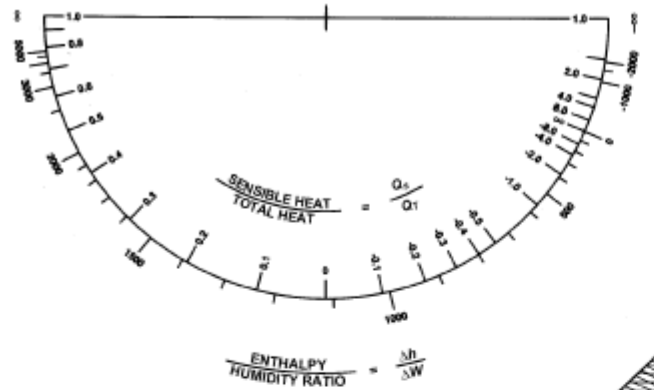
Humidity Ratio
 $W \approx 5.8 \text{ lb/lb}_{da}$

Dry Bulb Temp
 $T \approx 100^\circ\text{F}$

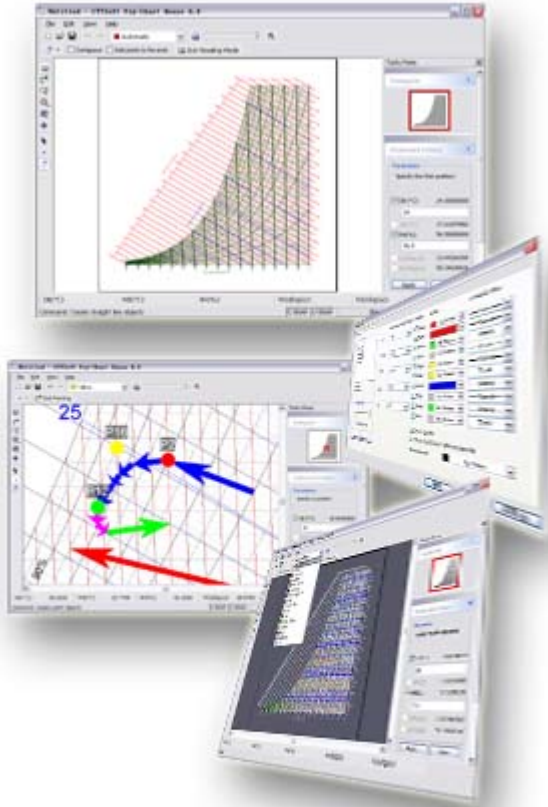








Applying psychrometrics



- We can also use psychrometric charts or software
 - Psych and Psychpro
 - Very popular psych chart and analysis software
 - I think at least one of these is in the AM 218 lab
- There are a bunch of online calculators as well
 - <http://www.psychrometric-calculator.com>
 - <http://www.sugartech.co.za/psychro/>
 - <http://www.wolframalpha.com/examples/Psychrometrics.html>
- And smart phone apps too
- You can also make your own (i.e., in Excel)
 - You will have a HW problem where you have to do this

Psychrometrics also involves learning how to use and combine those quantities to determine things like sensible and latent heating and cooling loads (i.e., **processes**) (covered in a future lecture)