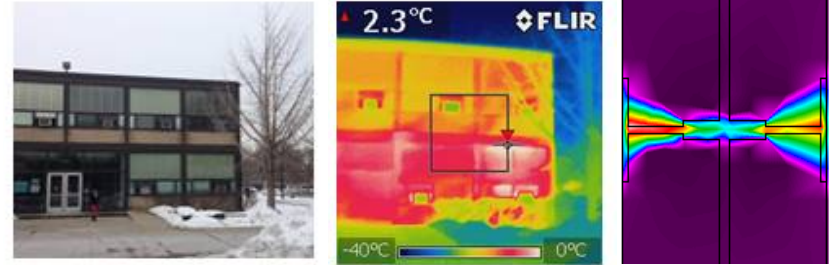


CAE 331/513

Building Science

Fall 2015



Week 5: September 22, 2015

Psychrometrics (chart and definitions)

Built
Environment
Research

@ IIT



*Advancing energy, environmental, and
sustainability research within the built environment*

www.built-envi.com

Twitter: [@built_envi](https://twitter.com/built_envi)

Dr. Brent Stephens, Ph.D.

Civil, Architectural and Environmental Engineering

Illinois Institute of Technology

brent@iit.edu

Guest lecture today:



Dr. Josephine Lau
Associate Professor
Architectural Engineering
University of Nebraska-Lincoln

“A quest for healthy and sustainable building environments and solutions”

Topics include:

- Demand controlled ventilation (DCV)
- Ultraviolet germicidal air cleaning
- School indoor environments and student achievement

Today, Tuesday, September 22, 2015
Wishnick Hall Room 116

Race to Zero: DOE student design competition



- Registration due November 12
- Submittals due March 24
- Competition held April 16-17 at NREL

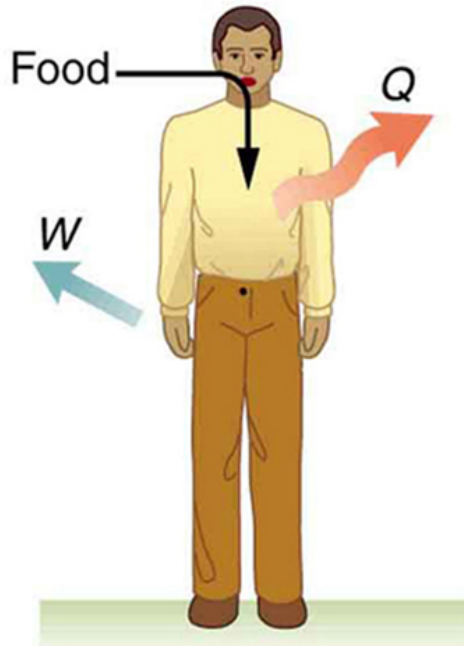
We are planning to have a CAE 497 Special Problems course in Spring 2016 devoted to this competition

Email me if you would like to participate!

<http://energy.gov/eere/buildings/us-department-energy-race-zero-student-design-competition>

Last time

- Human thermal comfort



$$\dot{Q} = MA_{skin}$$

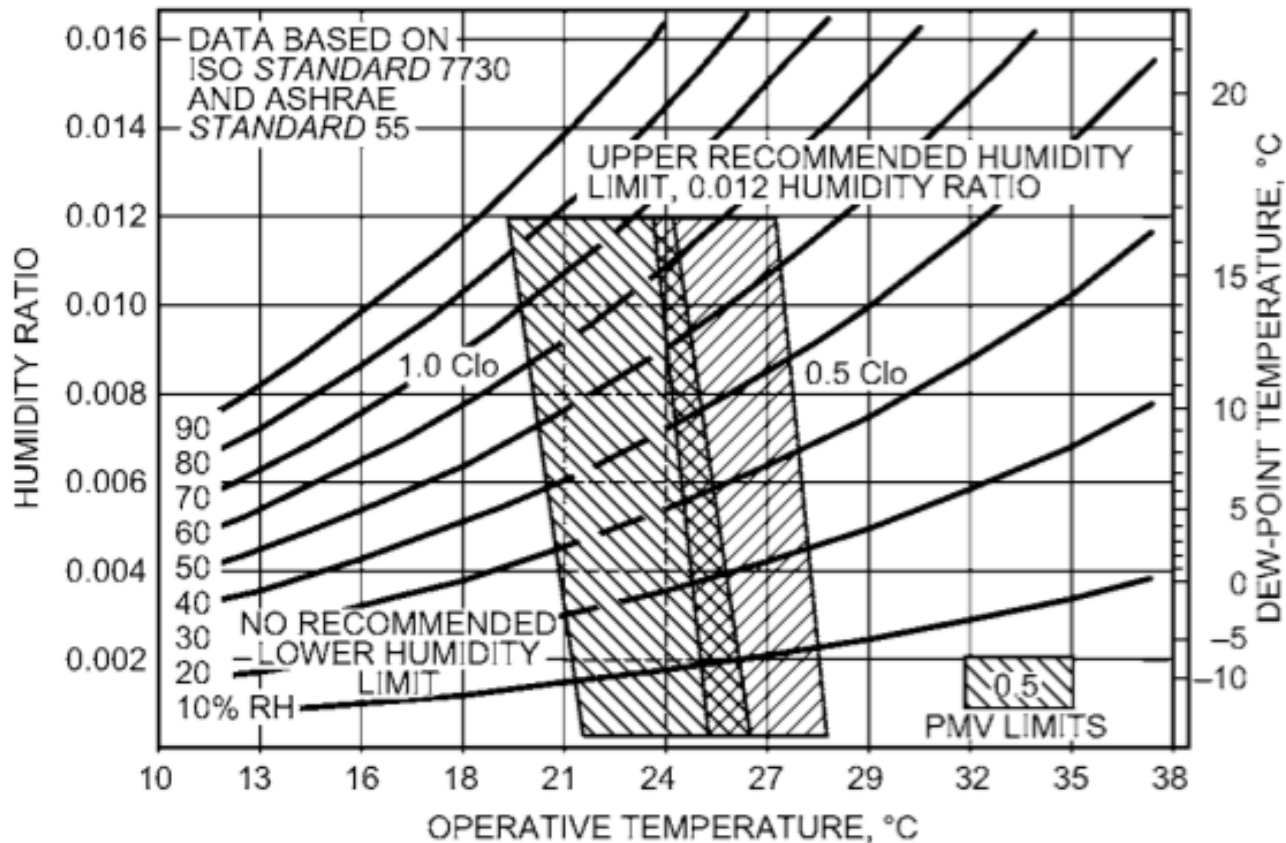
$$1 \text{ met} = 18.4 \frac{\text{Btu}}{\text{h} \cdot \text{ft}^2} = 58 \frac{\text{W}}{\text{m}^2}$$

$$A_D = 0.202m^{0.425}l^{0.725}$$

A_D = DuBois surface area, m^2
 m = mass, kg
 l = height, m

-3	-2	-1	0	+1	+2	+3
cold	cool	slightly cool	neutral	slightly warm	warm	hot

ASHRAE comfort zone



**3 equations
for operative
temperatures:**

$$t_o = \frac{(h_r t_{mr} + h_c t_a)}{h_r + h_c}$$

$$t_o = \frac{(t_{mr} + (t_a \times \sqrt{10v}))}{1 + \sqrt{10v}}$$

$$t_o = \frac{(t_a + t_{mr})}{2}$$

Fig. 5 ASHRAE Summer and Winter Comfort Zones
 [Acceptable ranges of operative temperature and humidity with
 air speed ≤ 0.2 m/s for people wearing 1.0 and 0.5 clo clothing
 during primarily sedentary activity (≤ 1.1 met)].

ASHRAE comfort zone: CBE Thermal Comfort Tool

CBE Thermal Comfort Tool

ASHRAE-55

Compare

Ranges

Select method:

PMV method

Air temperature

24.6 °C

Use operative temperature

Mean radiant temperature

26 °C

Air speed

0.07

Humidity

50

Metabolic

1.3

Clothing

0.55

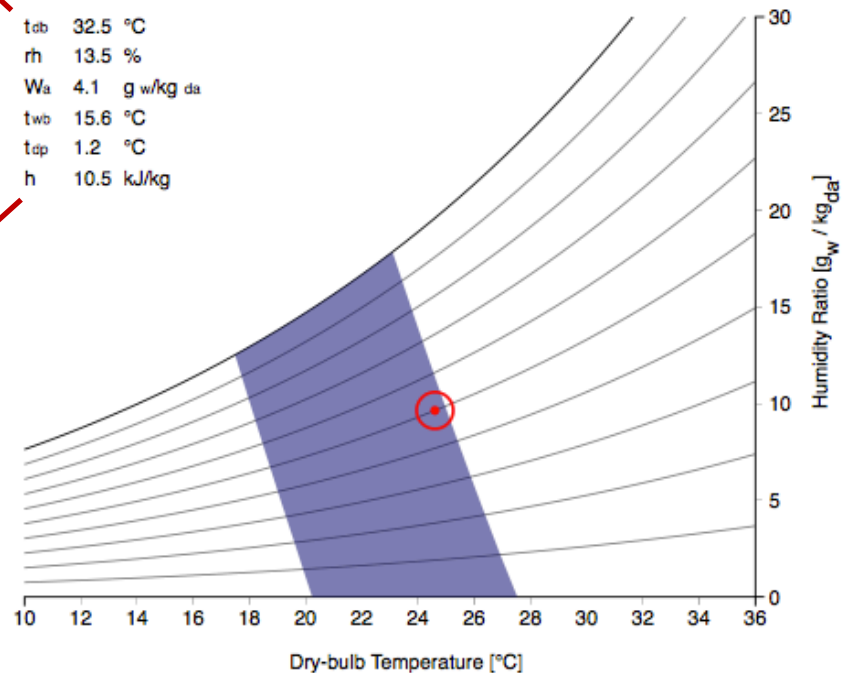
t_{db} 32.5 °C
 rh 13.5 %
 W_a 4.1 g w/kg da
 t_{wb} 15.6 °C
 t_{dp} 1.2 °C
 h 10.5 kJ/kg

✓ Complies with ASHRAE Standard 55-2010

PMV 0.44
PPD 9%
Sensation Neutral
SET 26.7°C

Psychrometric chart (air temperature)

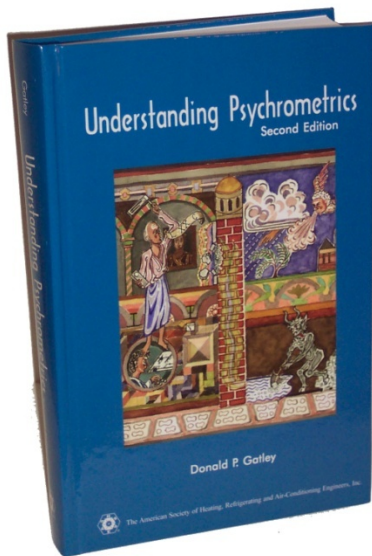
t_{db} 32.5 °C
 rh 13.5 %
 W_a 4.1 g w/kg da
 t_{wb} 15.6 °C
 t_{dp} 1.2 °C
 h 10.5 kJ/kg



PSYCHROMETRICS

Psychrometrics

Psychrometrics is the science and engineering of air/vapor mixtures



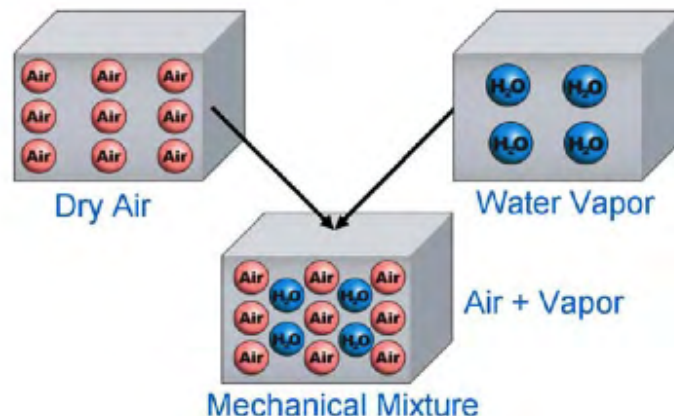
- For architectural engineers, the vapor is usually **water vapor**
- In building science and engineering we use psychrometrics to relate the thermodynamic and physical properties of moist air

Applying psychrometrics

- We need to understand **air temperature** and **moisture content** to understand **human thermal comfort**
 - In **hot, humid** weather we design HVAC systems to **remove** moisture by dehumidification/cooling
 - In **dry, cold** weather, we **add** moisture by humidifiers
- We are also concerned about moisture for structural, aesthetic, and indoor air quality reasons
 - Energy use too
- Psychrometrics also involves learning how to use and combine a variety of moist air parameters

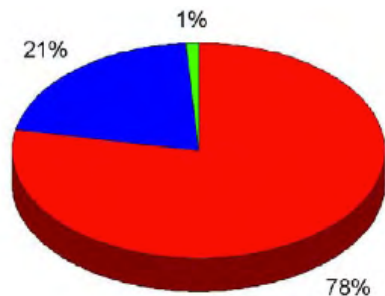
Some definitions for psychrometrics

- **Atmospheric air** contains:
 - Many gaseous components
 - Water vapor
 - Contaminants (particulate matter and gaseous pollutants)
- **Dry air** is atmospheric air with all of the water vapor removed
- **Moist air** is a two-component mixture of dry air and water vapor



Standard composition of dry air

Gas	Molecular weight (g/mol)	Volume %
Nitrogen (N ₂)	32.000	78.084
Oxygen (O ₂)	28.016	20.946
Argon (Ar)	39.444	0.9340
Carbon Dioxide (CO ₂)	44.010	0.03697
Neon (Ne)	20.179	0.00182
Helium (He)	4.002	0.00052
Methane (CH ₄)	16.042	0.00014
Krypton	83.800	0.00010



■ Nitrogen ■ Oxygen ■ Other Gases

Where does water fit in?

Standard composition of **moist** air

Gas	Molecular weight (g/mol)	Volume %
Nitrogen (N ₂)	32.000	78.084%
Oxygen (O ₂)	28.016	20.946%
Water (H₂O)	18.015	0 to 4%
Argon (Ar)	39.444	0.9340%
Carbon Dioxide (CO ₂)	44.010	0.03697%
Neon (Ne)	20.179	0.00182%
Helium (He)	4.002	0.00052%
Methane (CH ₄)	16.042	0.00014%
Krypton	83.800	0.00010%

Key terms for describing moist air

- To describe and deal with moist air, we need to be able to describe the relative portions of dry air and water vapor
- There are several different equivalent measures
 - Which one you use depends on what data you have to start with and what quantity you are trying to find
 - If you know two properties, you can usually get all the others

Key terms for describing moist air

Key terms to learn today:

1. Dry bulb temperature
2. Vapor pressure
3. Saturation
4. Relative humidity
5. Absolute humidity (or humidity ratio)
6. Dew point temperature
7. Wet bulb temperature
8. Enthalpy
9. Density
10. Specific volume

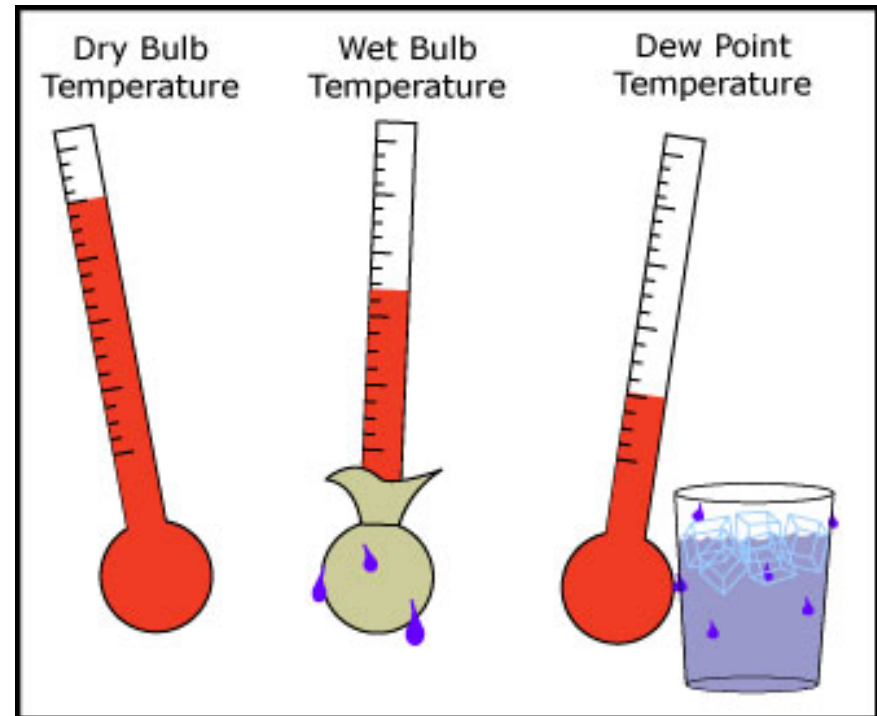
Three different temperatures: T , T_{dew} and T_{wb}

The standard temperature, T , we are all familiar with is called the **dry-bulb** temperature, or T_d

- It is a measure of internal energy

We can also define:

- **Dew-point** temperature, T_{dew}
 - Temperature at which water vapor changes into liquid (condensation)
 - Air is maximally **saturated** with water vapor
 - **Wet-bulb** temperature, T_{wb}
 - The temperature that a parcel of air would have if it were cooled to saturation (100% **relative humidity**) by the evaporation of water into it
- ✓ The energy needed to evaporate liquid water (heat of vaporization) is taken from the air in the form of sensible heat and converted to latent heat, which lowers the temperature at constant enthalpy



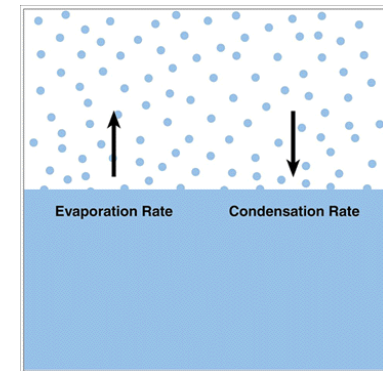
Units of Celsius, Fahrenheit, or Kelvin

Key concepts: Vapor pressure and Saturation

- Air can hold moisture (i.e., **water vapor**)
- **Vapor pressure** is a measurement of the amount of water vapor in a volume/parcel of air (ideal gas)

$$p_w$$

*Units of pressure, Pa or kPa
(partial pressure)



- The amount of moisture air can hold in vapor form before condensation occurs is dependent on temperature
 - We call the limit **saturation**

$$p_{ws}$$

*Units of pressure, Pa or kPa



Key concept: Relative humidity, ϕ

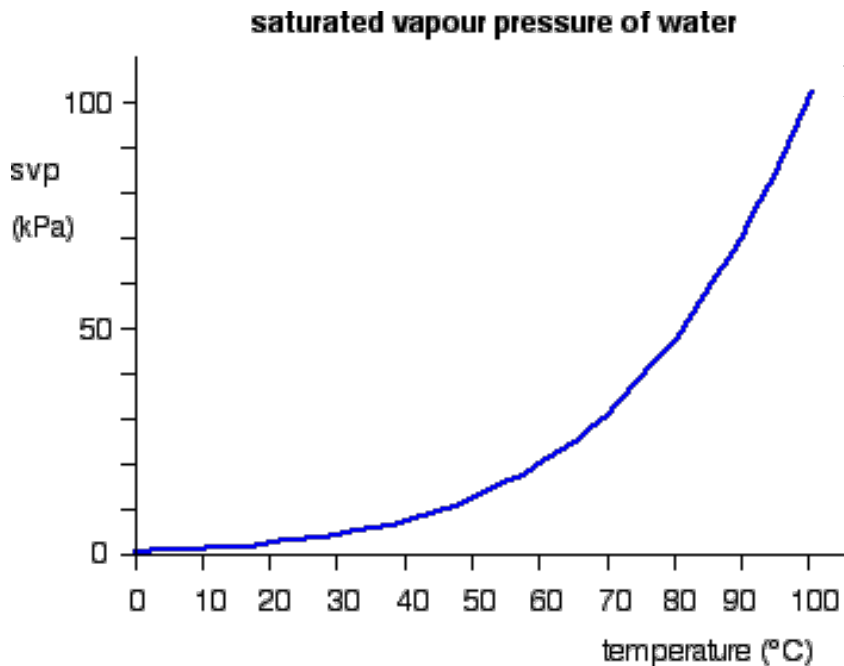
- **Relative humidity** (RH, or ϕ) is the ratio of the vapor pressure of water vapor in a sample of air to the **saturation** vapor pressure at the dry bulb temperature of the sample
- Relative humidity \neq absolute humidity



$$\phi = \frac{p_w}{p_{ws}}$$

Key concept: Saturation vapor pressure, p_{ws}

- The **saturation vapor pressure** is the partial pressure of water vapor at saturation (p_{ws}) *Units of pressure, Pa or kPa
 - Cannot absorb any more moisture at that temperature
- We can look up p_{ws} in tables (as a function of T)
 - Table 3 in Ch.1 of 2013 ASHRAE Fundamentals
- We can also use empirical equations:



Equation for p_{ws} :

$$\ln p_{ws} = \frac{C_8}{T} + C_9 + C_{10}T + C_{11}T^2 + C_{12}T^3 + C_{13} \ln T$$

where

$$C_8 = -5.800\ 220\ 6\ E+03$$

$$C_9 = 1.391\ 499\ 3\ E+00$$

$$C_{10} = -4.864\ 023\ 9\ E-02$$

$$C_{11} = 4.176\ 476\ 8\ E-05$$

$$C_{12} = -1.445\ 209\ 3\ E-08$$

$$C_{13} = 6.545\ 967\ 3\ E+00$$

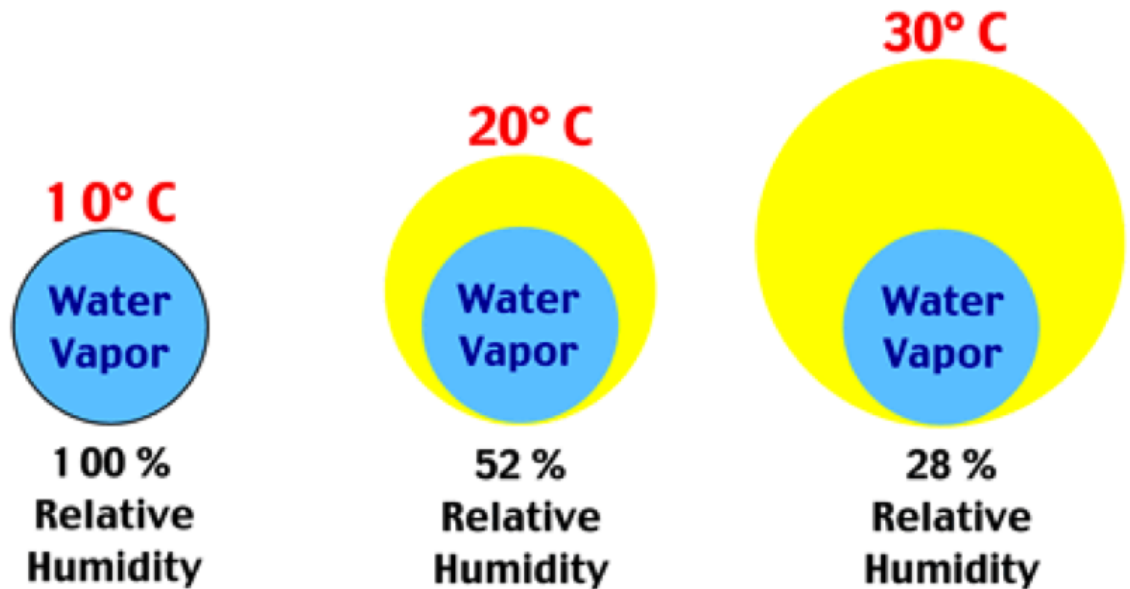
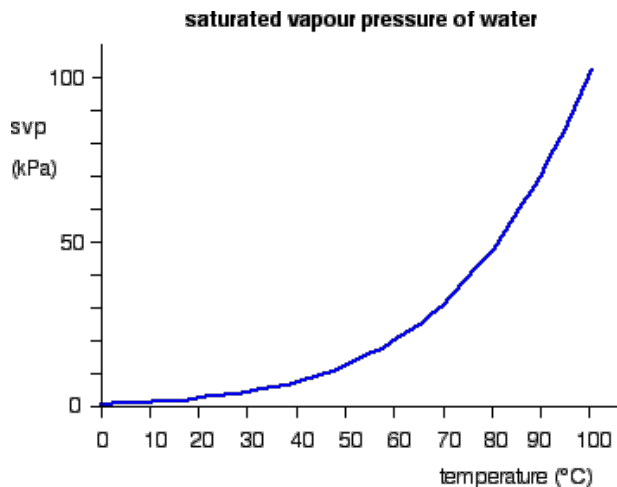
p_{ws} = saturation pressure, Pa

T = absolute temperature, K = °C + 273.15

Relative humidity and temperature

- **Relative humidity** (RH, or ϕ) is a function of temperature

$$\phi = \frac{p_w}{p_{ws}}$$



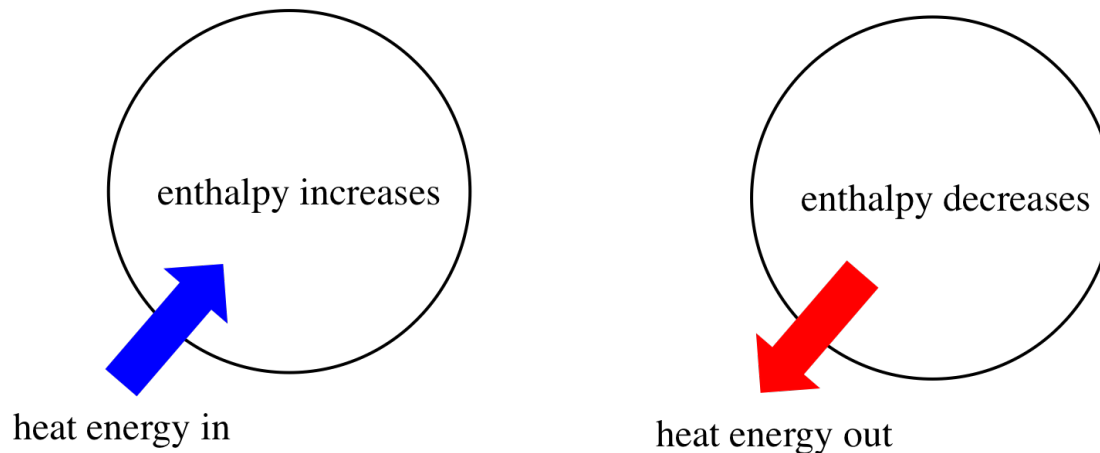
Key concept: Humidity ratio, W

- The **humidity ratio** is a direct measure of the **moisture content** of a parcel of air (a measure of absolute humidity)
- Simply, the humidity ratio is the mass quantity of water vapor that exists in a mass parcel of dry air
 - Units of mass of water vapor per mass of dry air
 - kg/kg ($\text{kg}_w/\text{kg}_{da}$)
 - g/kg (g_w/g_{da})

$$W = \frac{\text{mass of water vapor}}{\text{mass of dry air}} \left[\frac{\text{kg}_w}{\text{kg}_{da}} \right]$$

Key concept: **Enthalpy**

- **Enthalpy** is a measure of the amount of energy in a system
 - Units of Joules or BTU (or J/kg or BTU/lb)
- The enthalpy of moist air is the total enthalpy of the dry air plus the water vapor mixture per mass of moist air
- Includes:
 - Enthalpy of dry air, or **sensible** heat
 - Enthalpy of evaporated water, or **latent** heat



Key concept: Density and specific volume

Air density

- Density is a measure of the mass of moist air per unit volume of air
- Includes mass of dry air + water vapor

$$\rho = \frac{\text{mass of moist air}}{\text{volume of moist air}} \quad \left[\frac{\text{kg}}{\text{m}^3} \right]$$

Specific volume

- Specific volume is the volume of unit mass of dry air at a given temperature, expressed as m³/kg (inverse of dry air density)

$$v = \frac{\text{volume of dry air}}{\text{mass of dry air}} \quad \left[\frac{\text{m}^3}{\text{kg}_{\text{da}}} \right]$$

The Psychrometric Chart

- There are both simple and complex ways to estimate these properties
 - Equations and tables (more complex, save for next lecture)
 - Graphically using ...
- **The Psychrometric Chart**
 - Plots dry bulb temperature (T) on the x-axis and humidity ratio (W) on the y-axis
 - Shows relationships between T and W and relative humidity, wet-bulb temperature, vapor pressure, specific volume, and enthalpy
 - Charts are unique at each value of atmospheric pressure (p)
- Both SI and IP versions are on BB in the materials folder



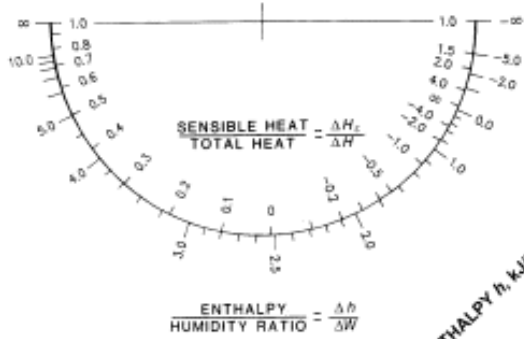
ASHRAE PSYCHROMETRIC CHART NO. 1

NORMAL TEMPERATURE SEA LEVEL

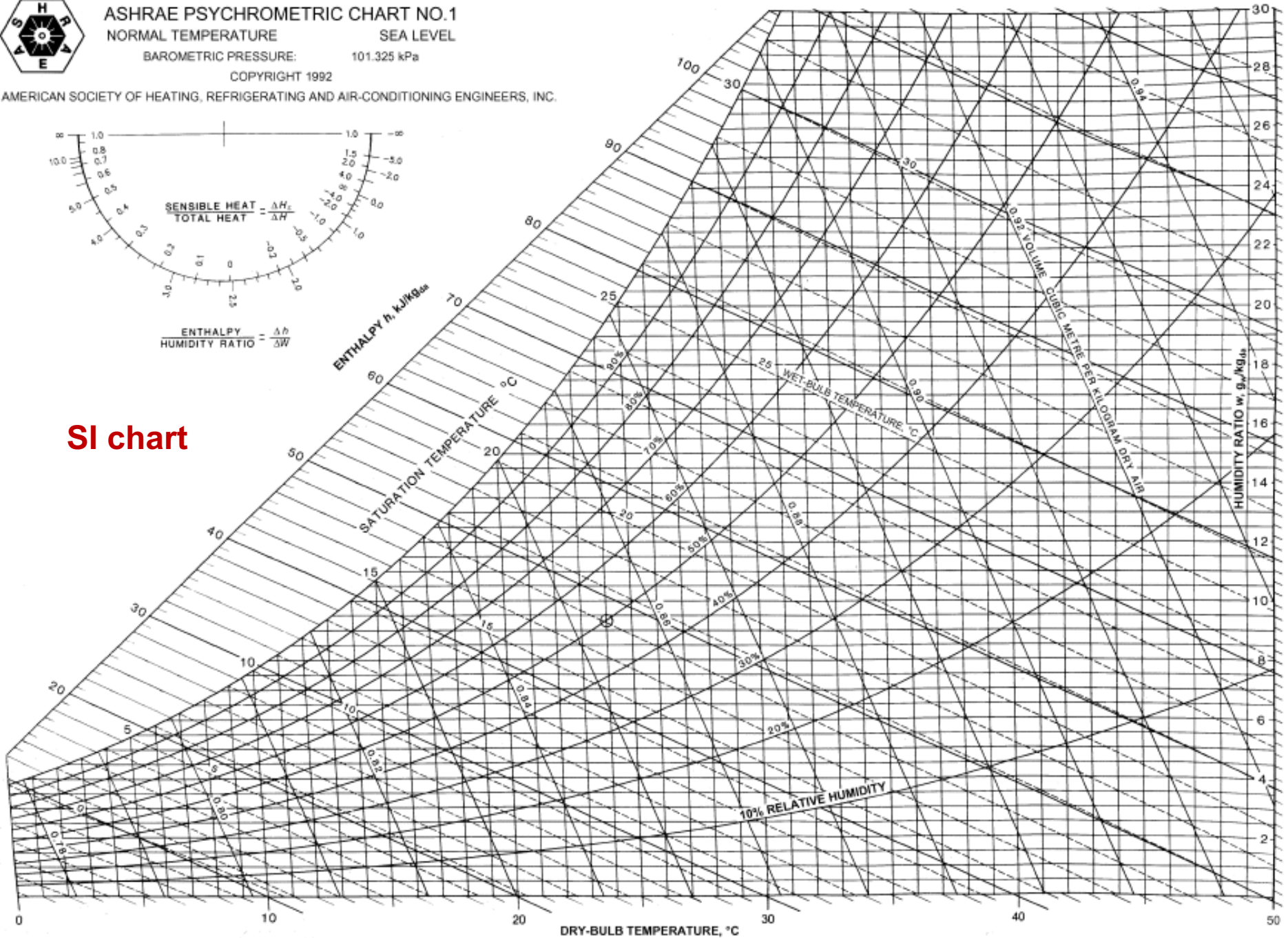
BAROMETRIC PRESSURE: 101.325 kPa

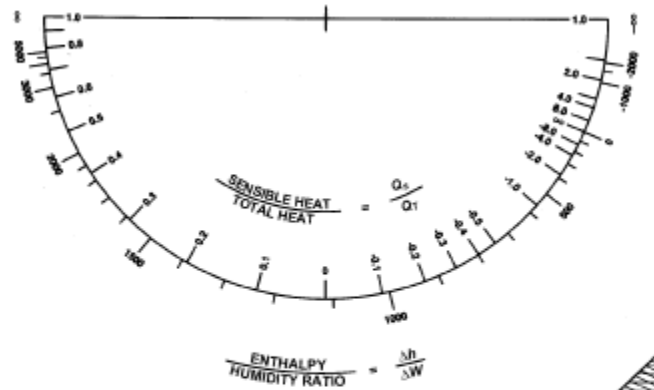
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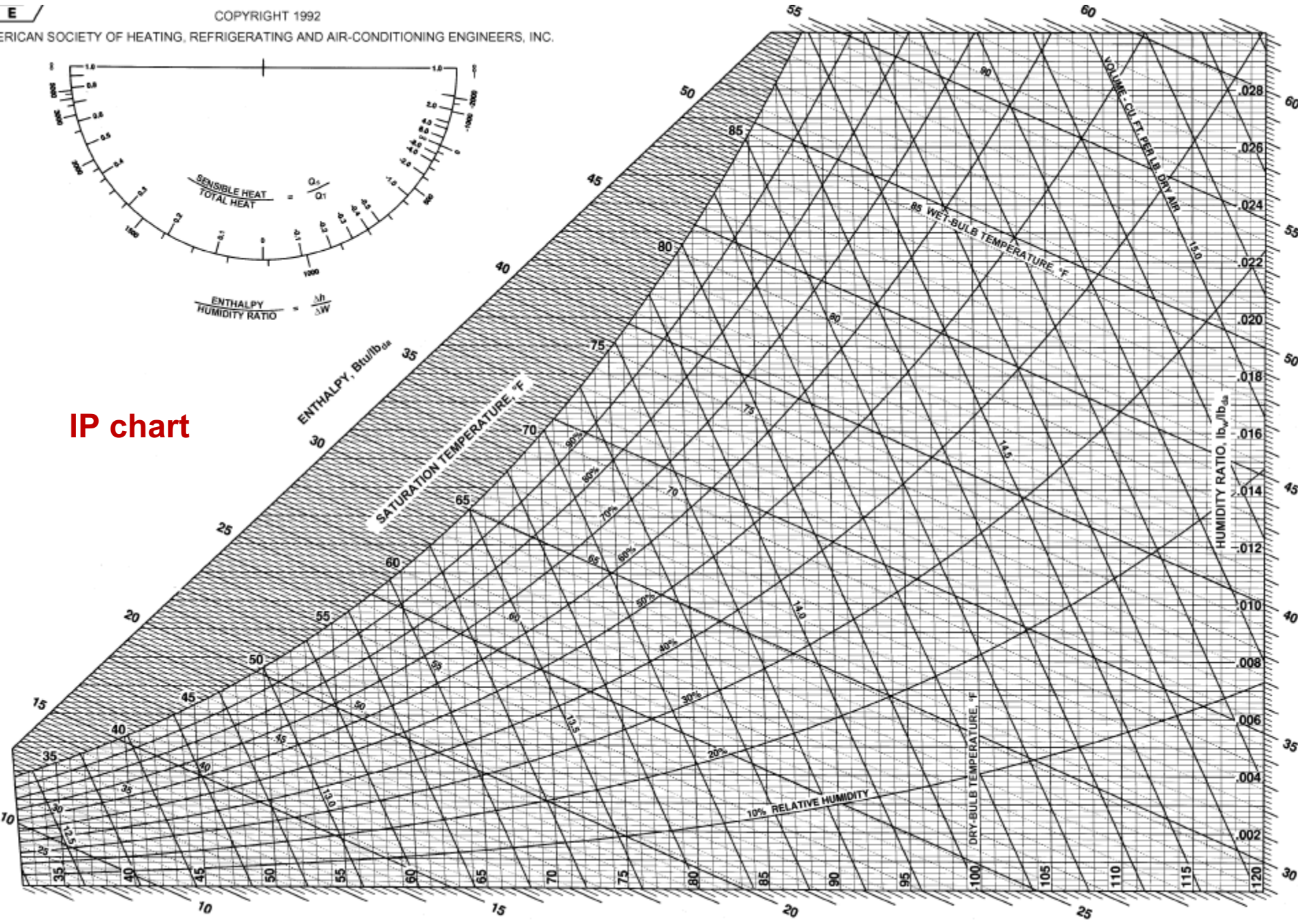


SI chart



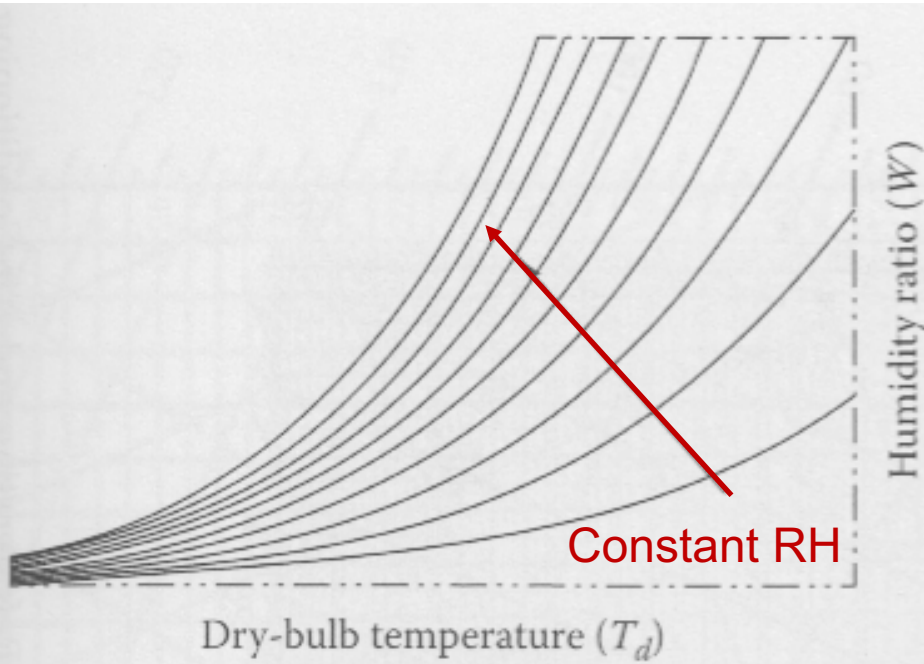


IP chart

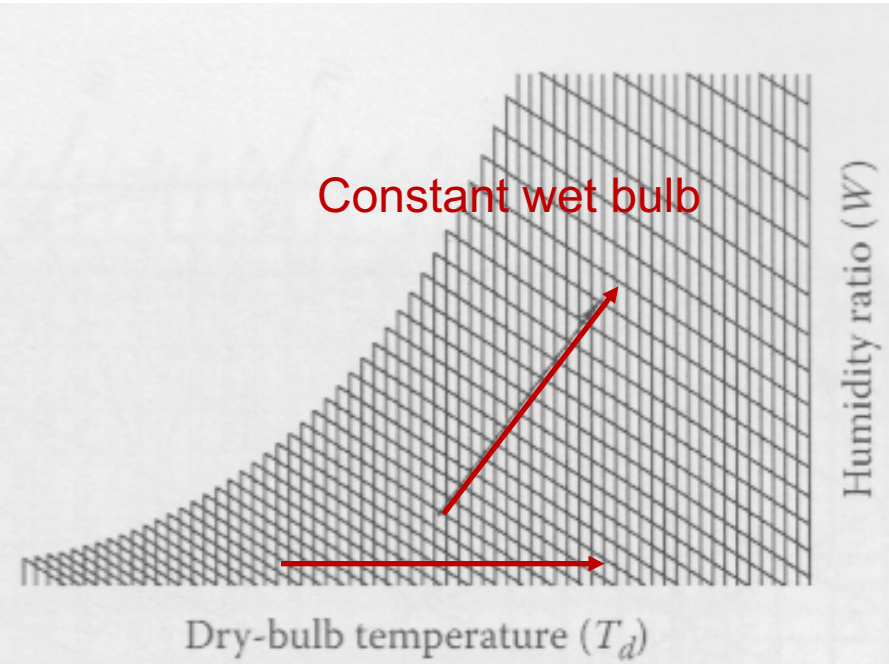


Deciphering the psychrometric chart

Lines of constant RH



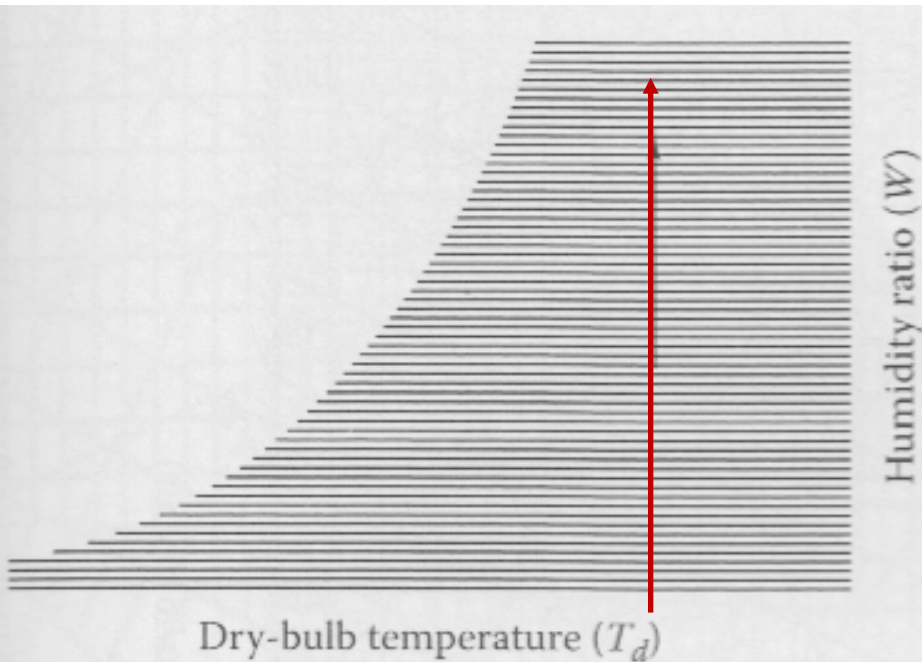
Lines of constant wet-bulb and dry-bulb



Constant dry bulb temperature

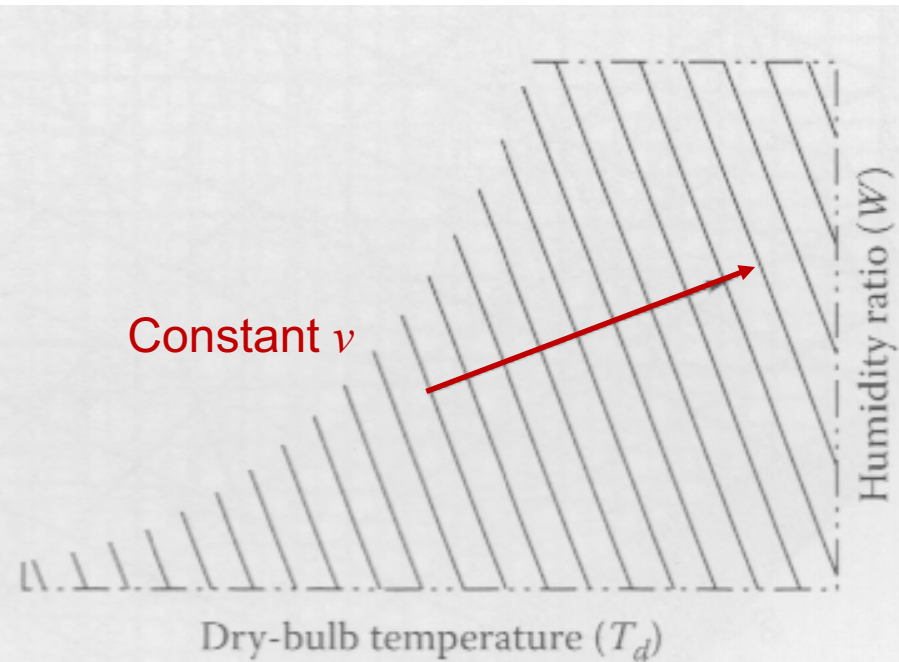
Deciphering the psychrometric chart

Lines of constant humidity ratio



Constant W

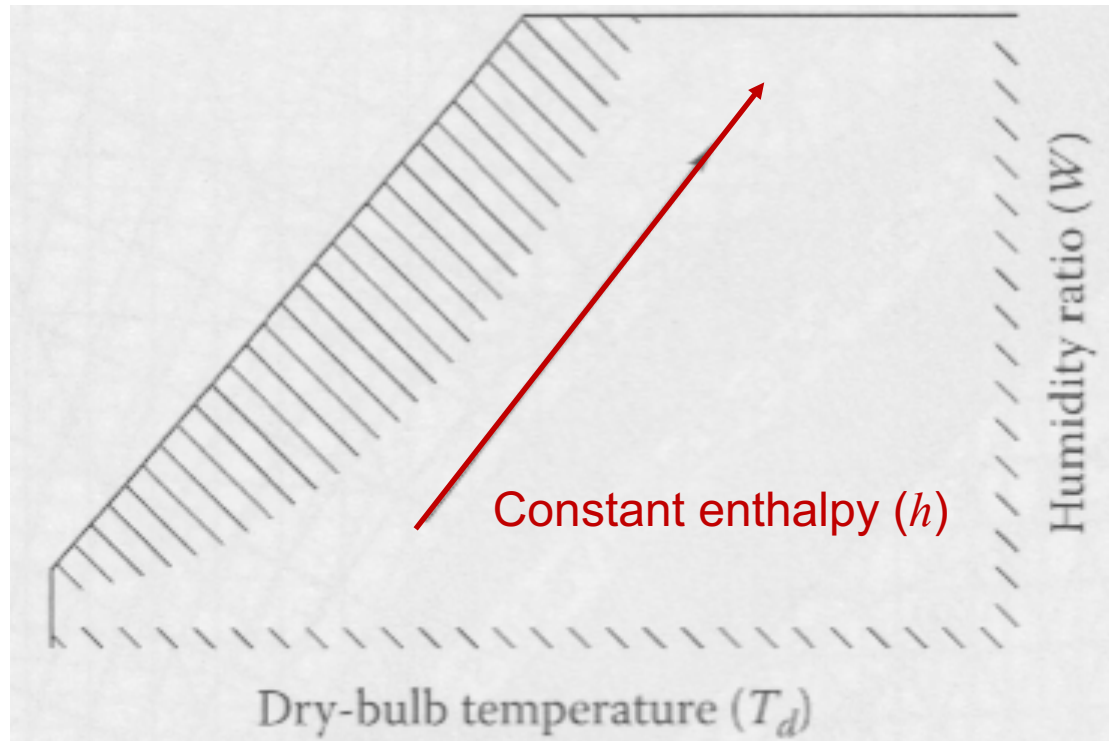
Lines of constant specific volume



Constant v

Deciphering the psychrometric chart

Lines of constant enthalpy





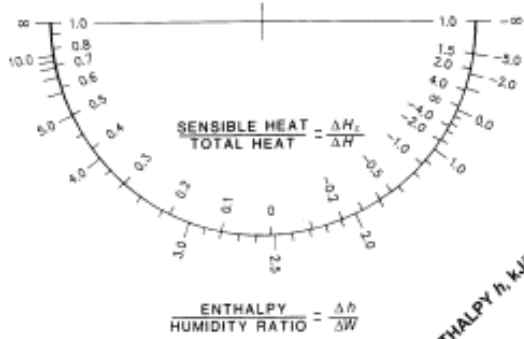
ASHRAE PSYCHROMETRIC CHART NO. 1

NORMAL TEMPERATURE SEA LEVEL

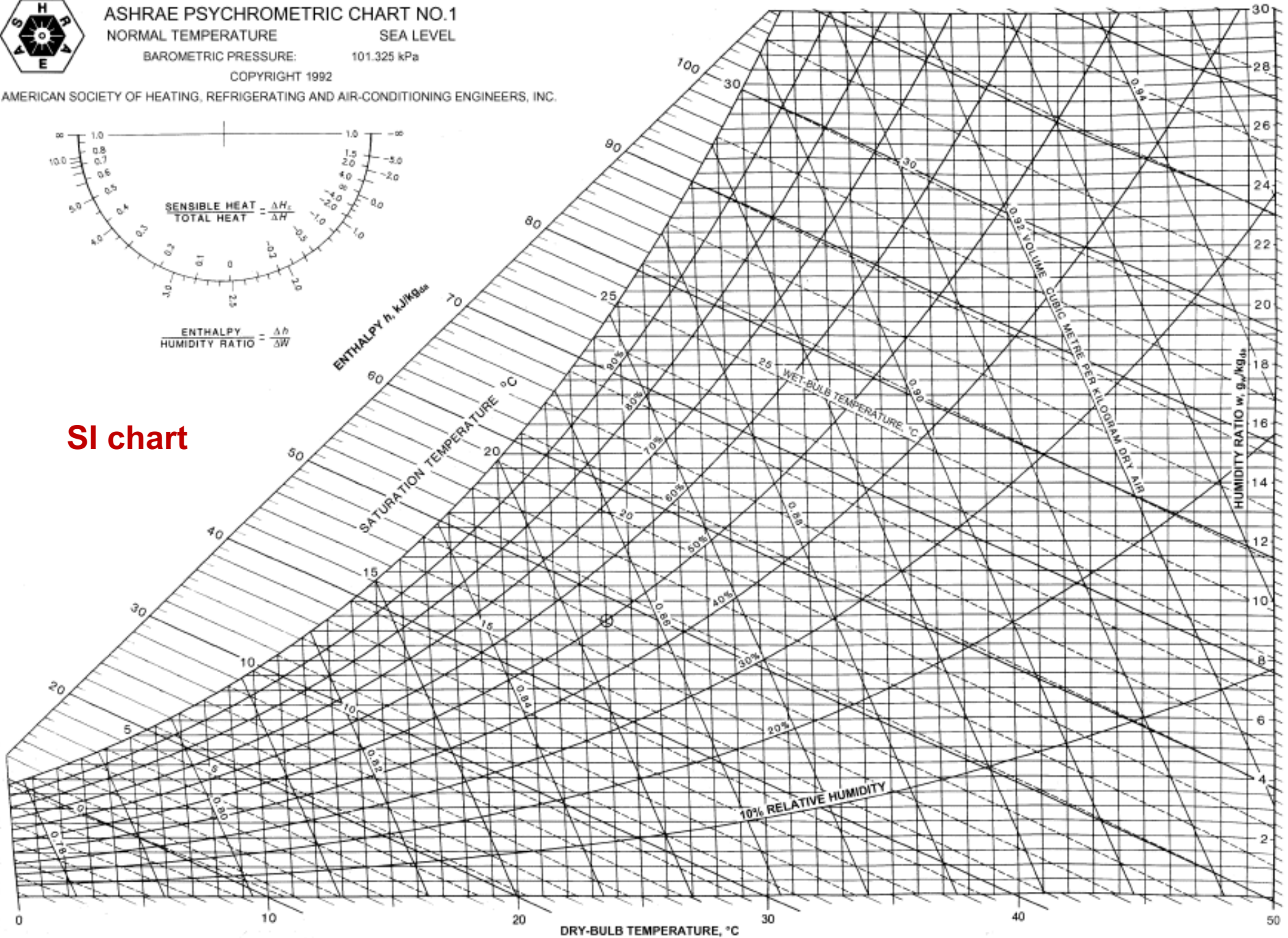
BAROMETRIC PRESSURE: 101.325 kPa

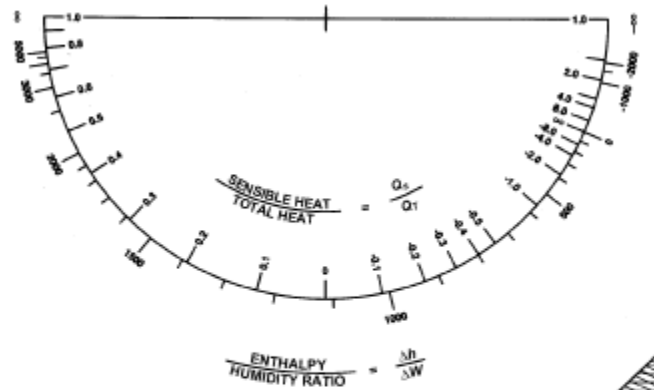
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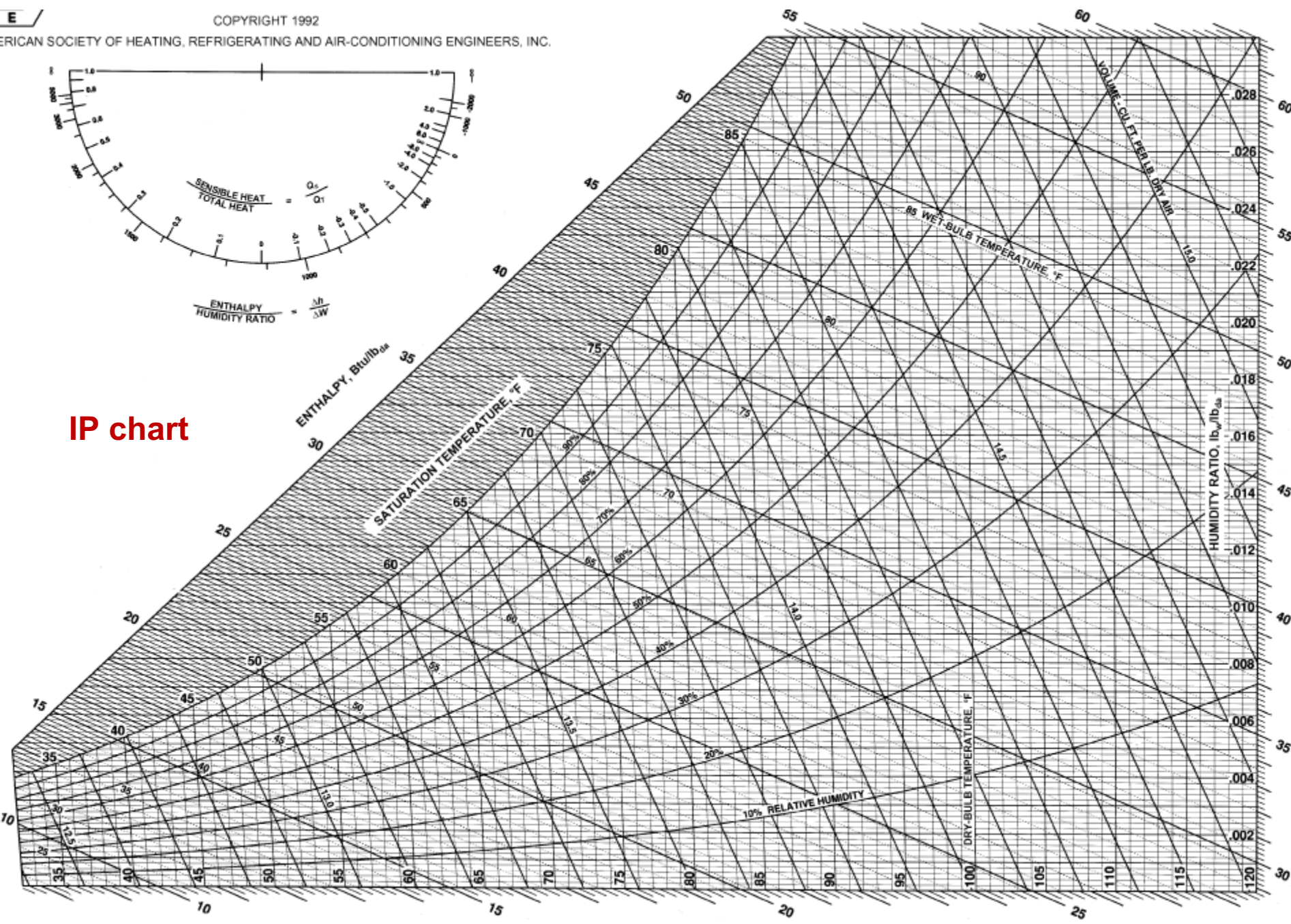


SI chart





IP chart



Some psychrometric examples

Moist air exists at 22°C dry-bulb temperature with 50% RH

Find the following:

- (a) the humidity ratio, W
- (b) dew point temperature, T_{dew}
- (c) wet-bulb temperature, T_{wb}
- (d) enthalpy, h
- (e) specific volume, v
- (f) dry air density, ρ



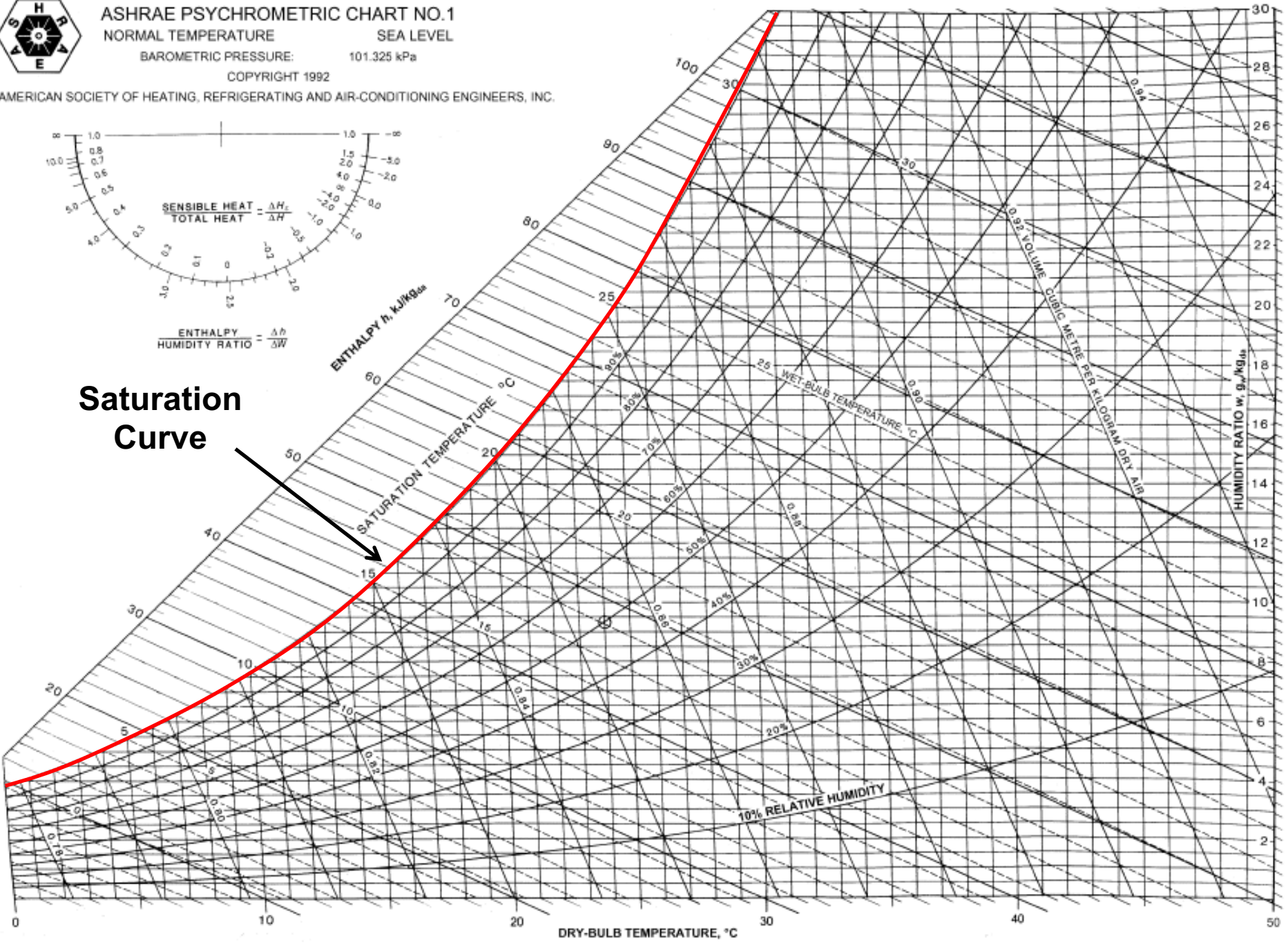
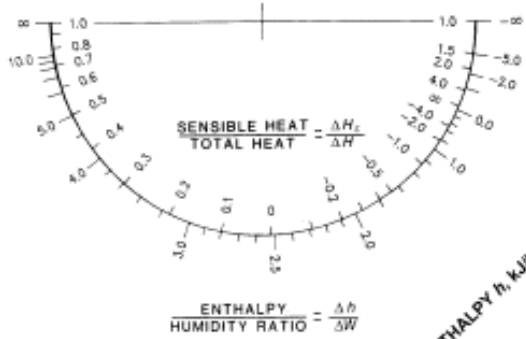
ASHRAE PSYCHROMETRIC CHART NO. 1

NORMAL TEMPERATURE SEA LEVEL

BAROMETRIC PRESSURE: 101.325 kPa

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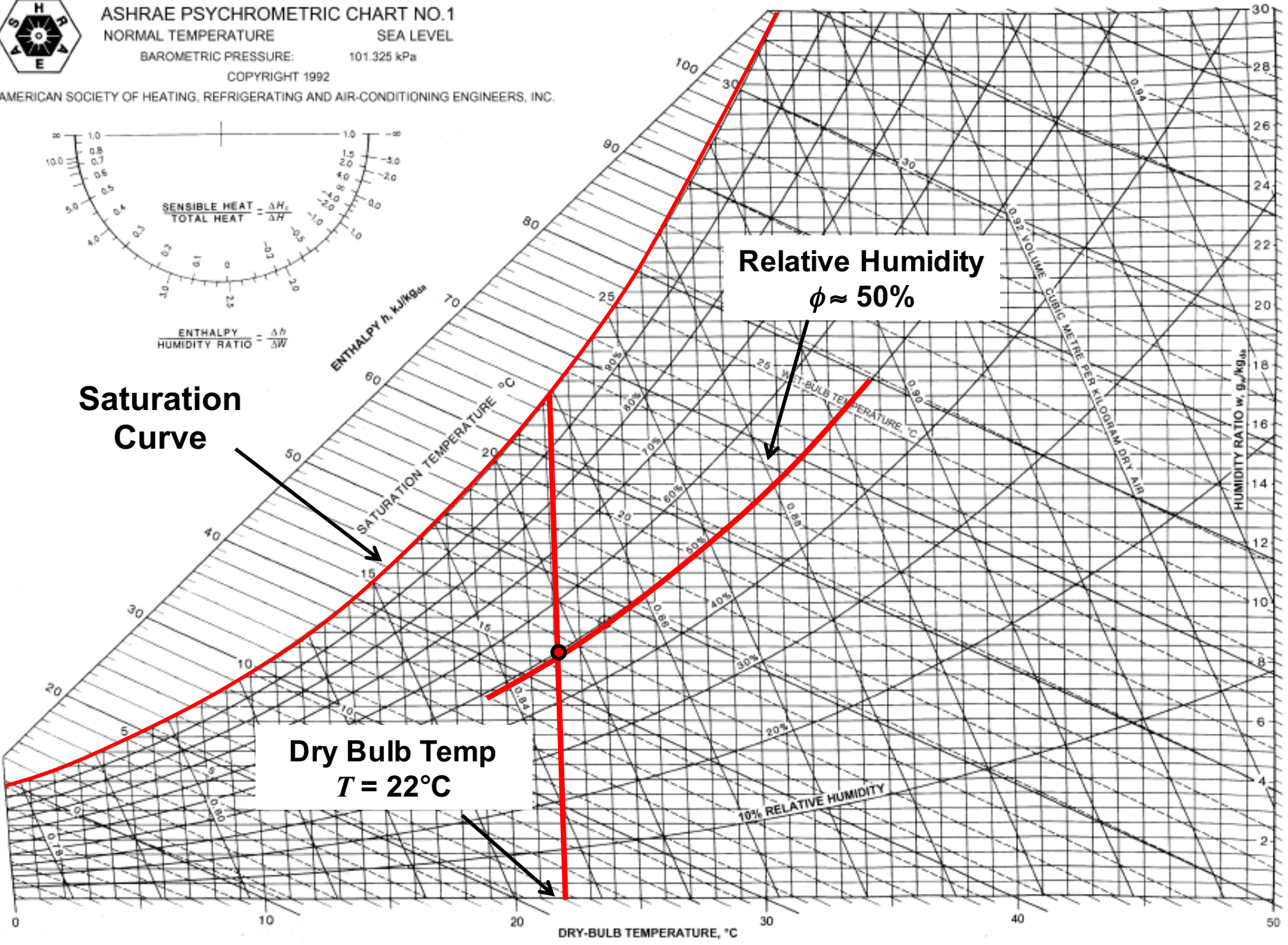
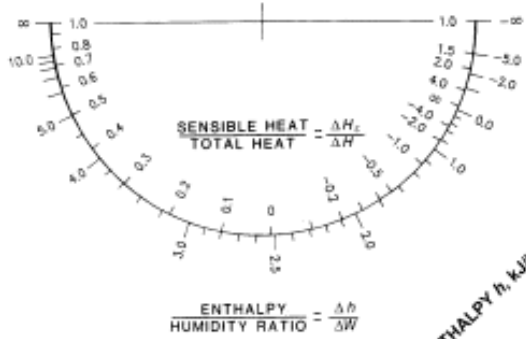
Saturation Curve





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NORMAL TEMPERATURE SEA LEVEL
BAROMETRIC PRESSURE: 101.325 kPa
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Saturation Curve

Relative Humidity $\phi \approx 50\%$

Dry Bulb Temp $T = 22^\circ\text{C}$

DRY-BULB TEMPERATURE, °C



ASHRAE PSYCHROMETRIC CHART NO.1

NORMAL TEMPERATURE

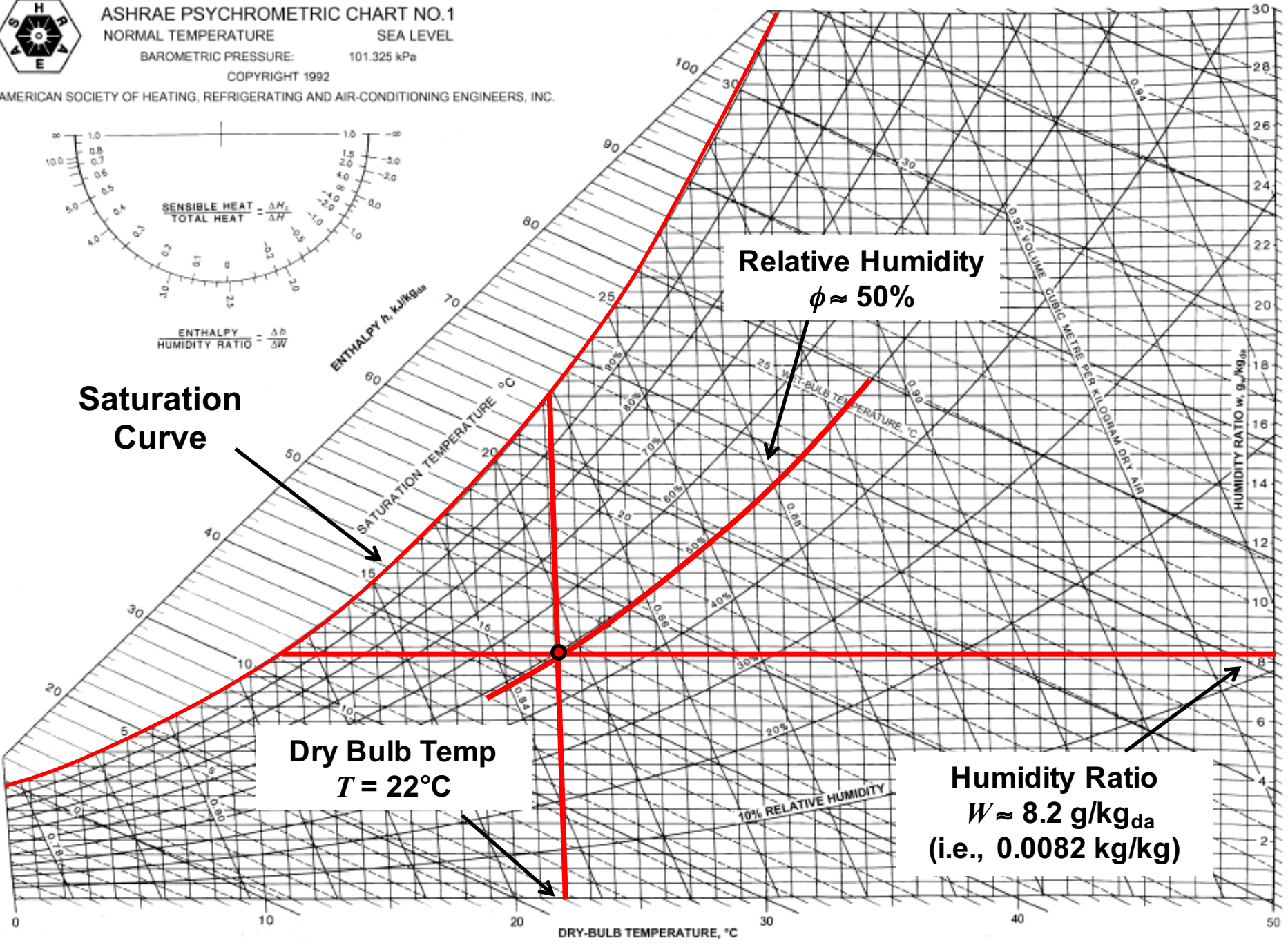
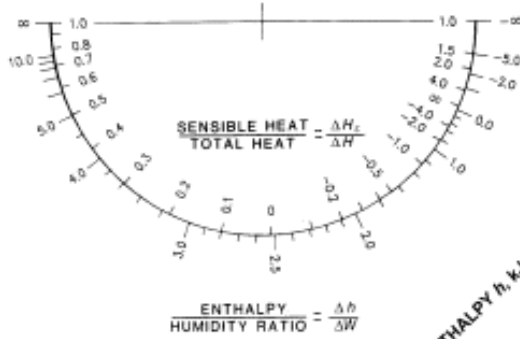
SEA LEVEL

BAROMETRIC PRESSURE:

101.325 kPa

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Saturation Curve

Relative Humidity
 $\phi \approx 50\%$

Dry Bulb Temp
 $T = 22^\circ\text{C}$

Humidity Ratio
 $W \approx 8.2 \text{ g/kg}_{da}$
(i.e., 0.0082 kg/kg)

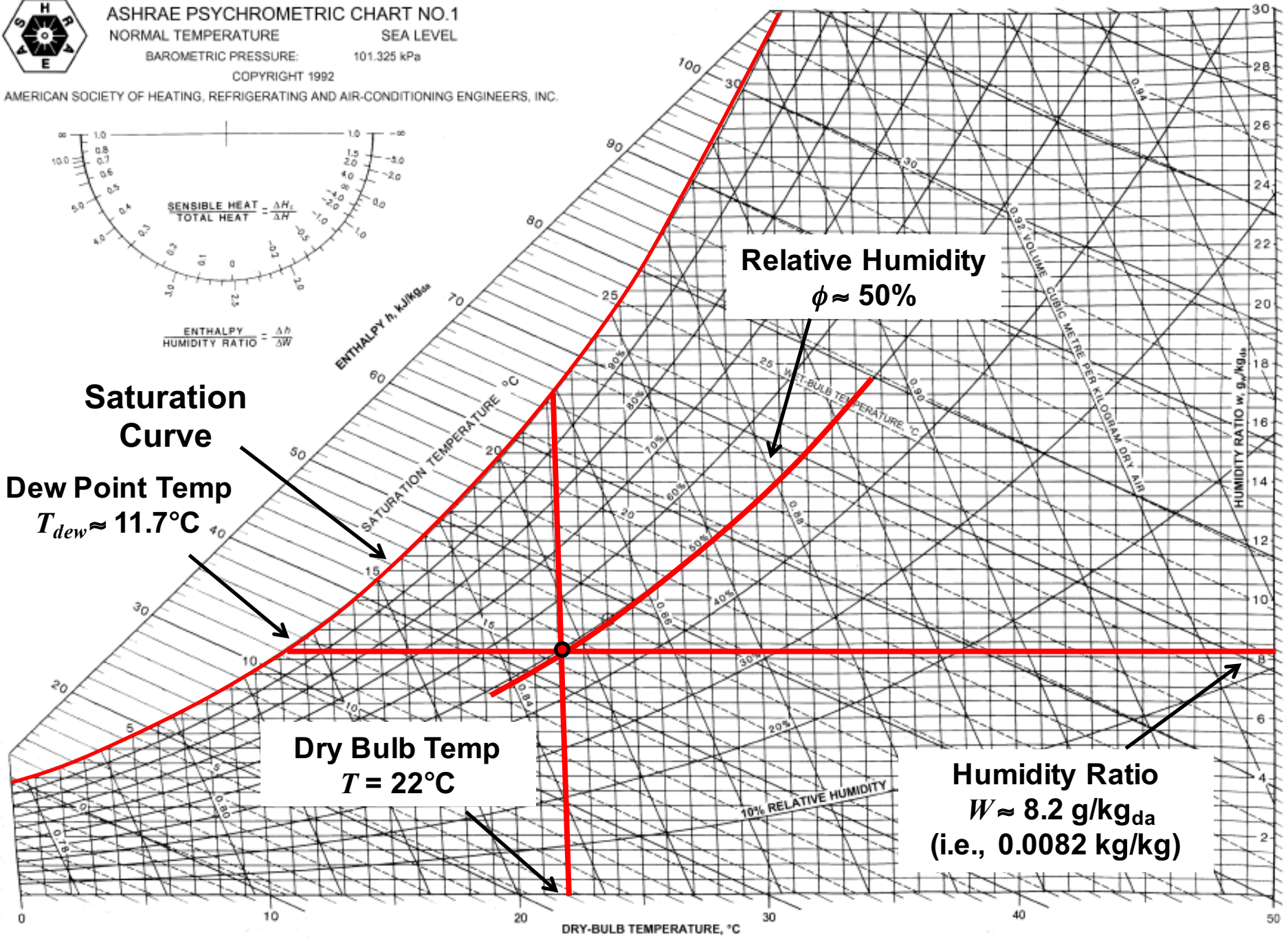
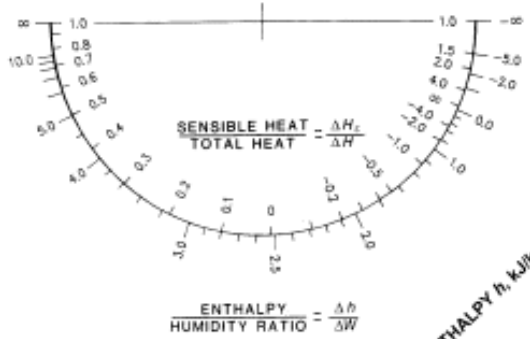
DRY-BULB TEMPERATURE, °C

HUMIDITY RATIO w , g/kg_{da}



ASHRAE PSYCHROMETRIC CHART NO.1
 NORMAL TEMPERATURE SEA LEVEL
 BAROMETRIC PRESSURE: 101.325 kPa
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Relative Humidity
 $\phi \approx 50\%$

Saturation Curve
 Dew Point Temp
 $T_{dew} \approx 11.7^\circ\text{C}$

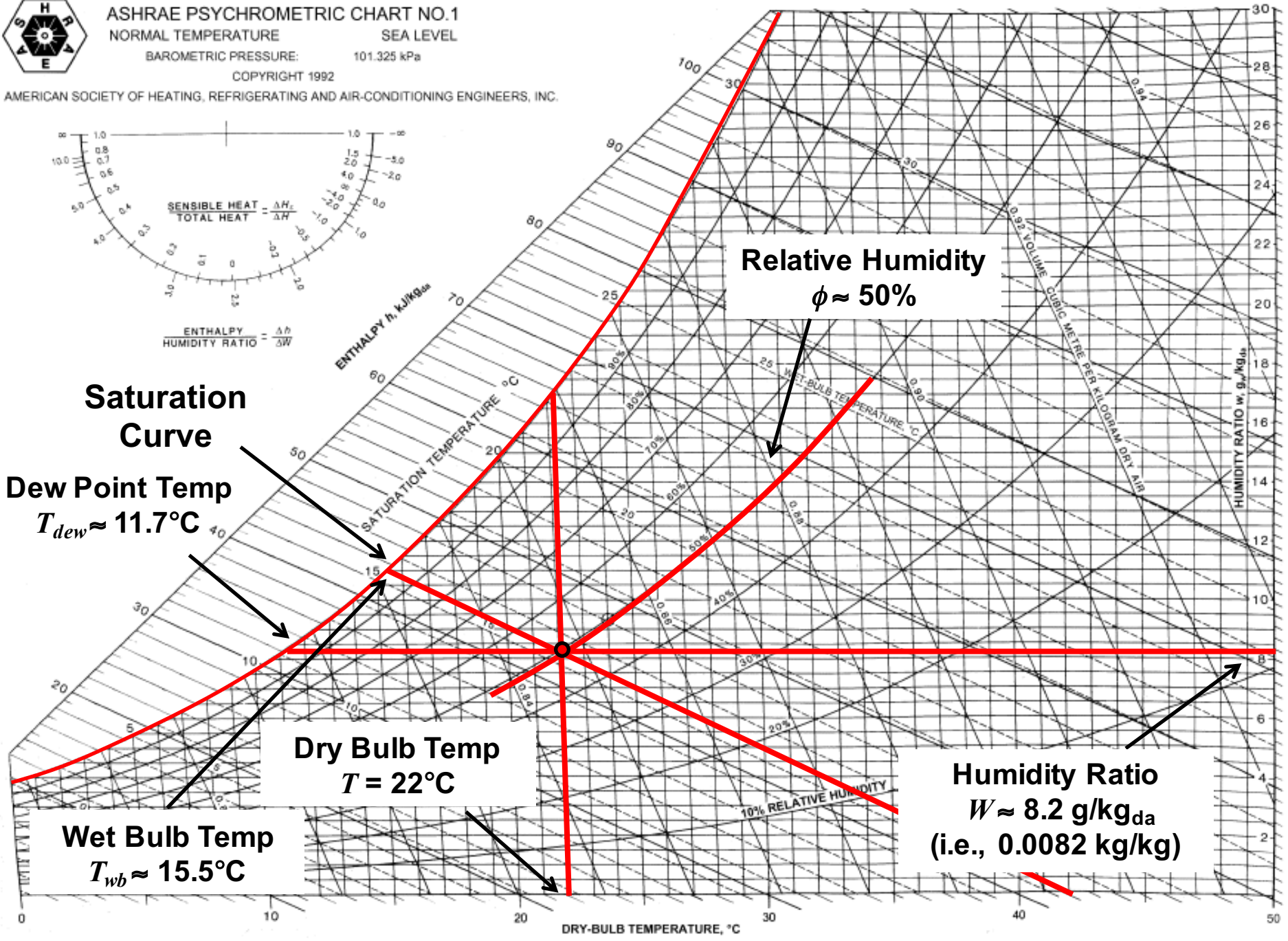
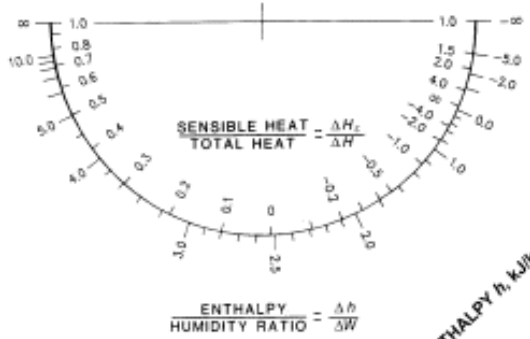
Dry Bulb Temp
 $T = 22^\circ\text{C}$

Humidity Ratio
 $W \approx 8.2 \text{ g/kg}_{da}$
 (i.e., 0.0082 kg/kg)



ASHRAE PSYCHROMETRIC CHART NO.1
 NORMAL TEMPERATURE SEA LEVEL
 BAROMETRIC PRESSURE: 101.325 kPa
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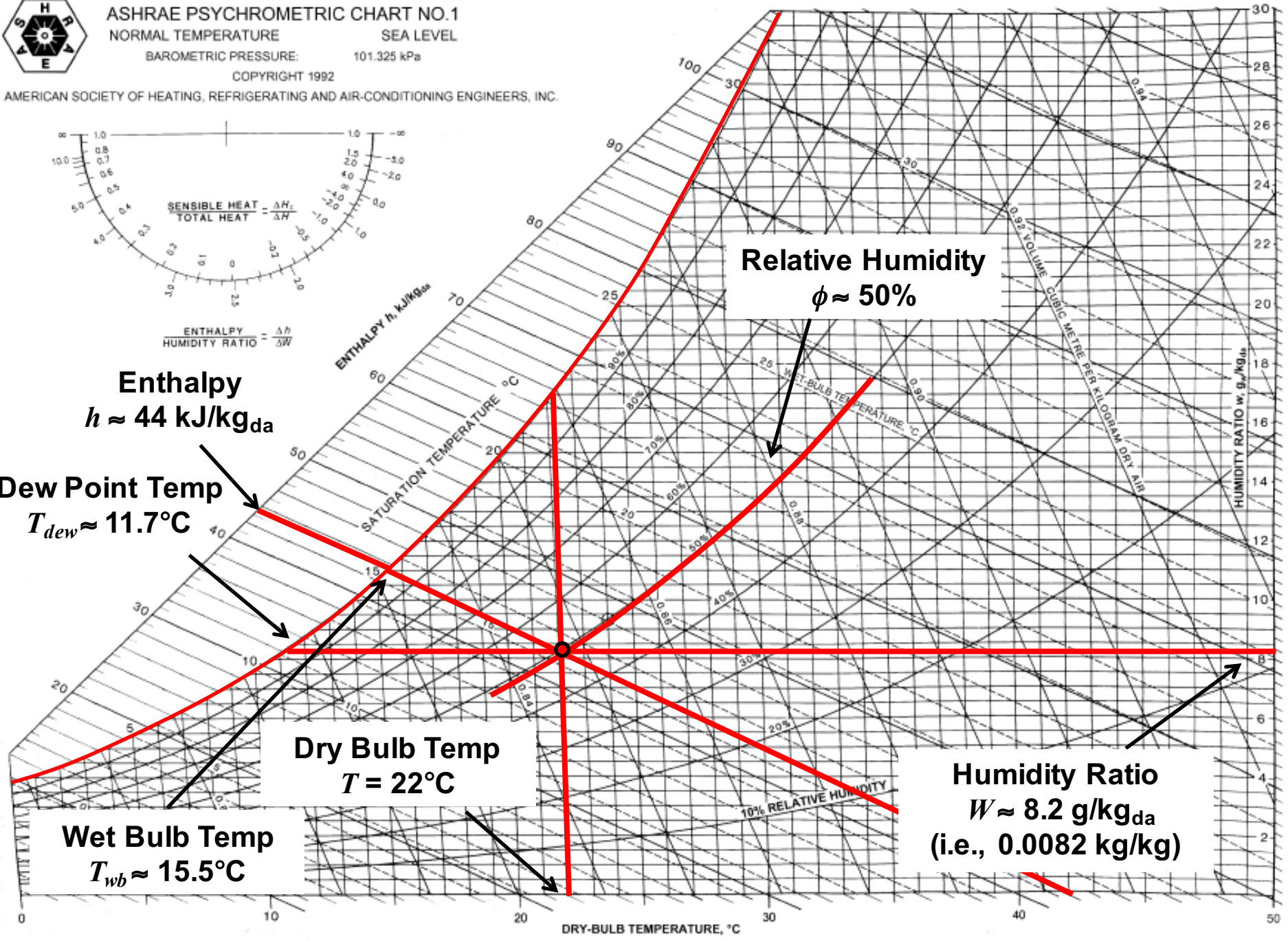
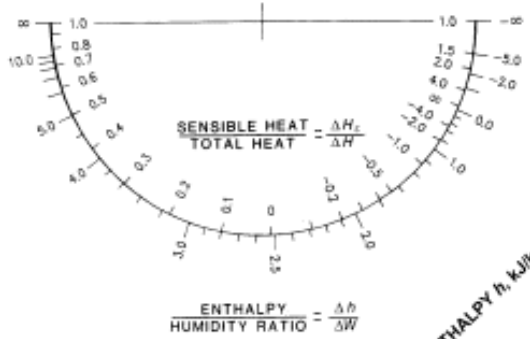
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 NORMAL TEMPERATURE SEA LEVEL
 BAROMETRIC PRESSURE: 101.325 kPa
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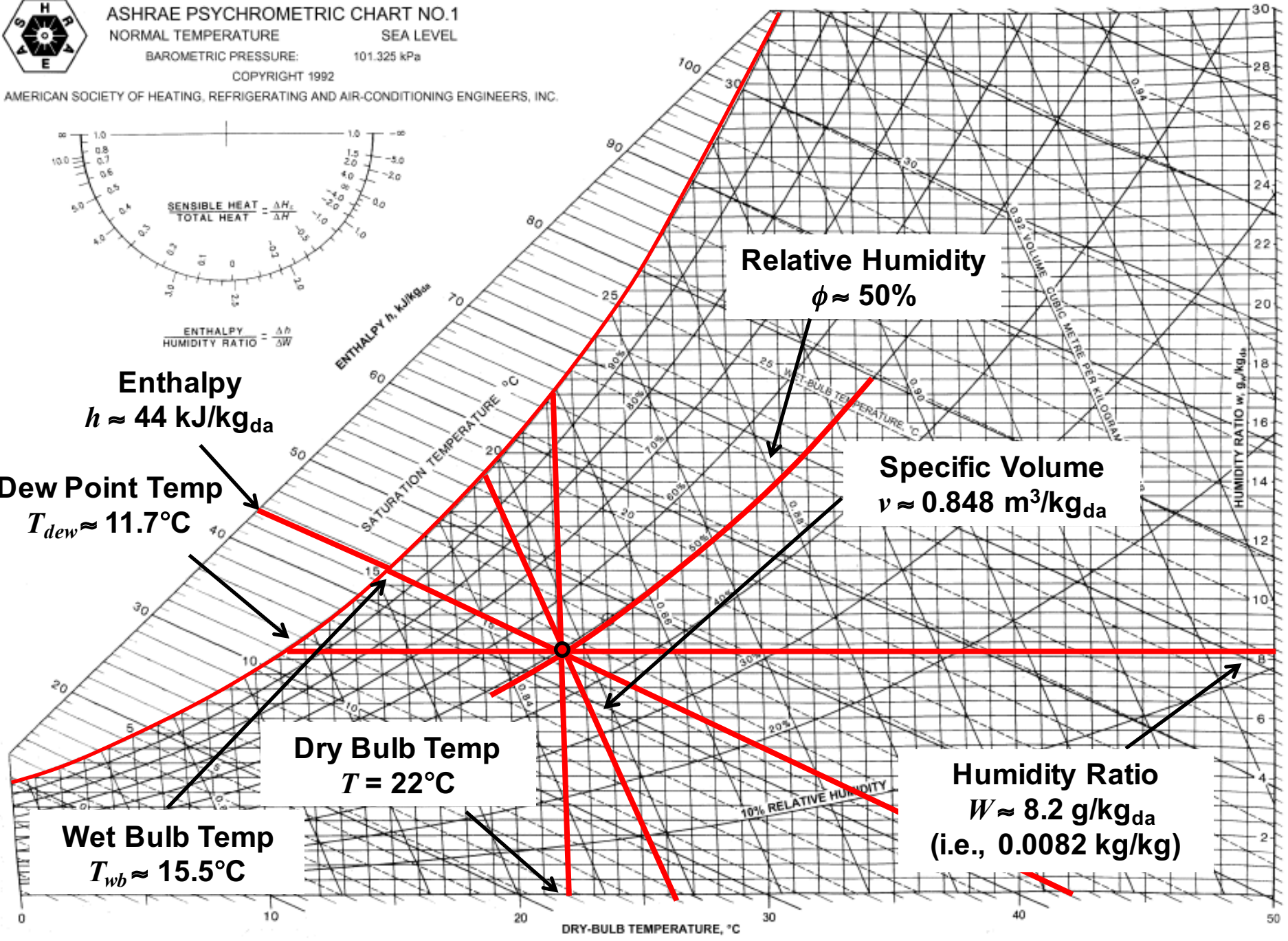
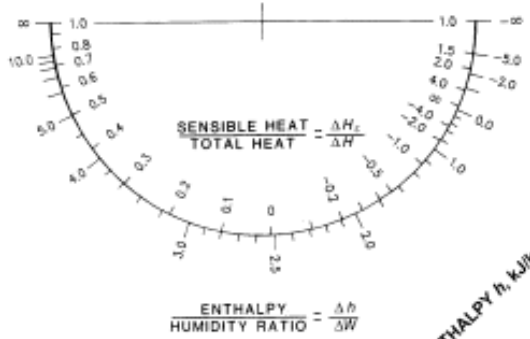
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 NORMAL TEMPERATURE SEA LEVEL
 BAROMETRIC PRESSURE: 101.325 kPa
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Relative Humidity
 $\phi \approx 50\%$

Specific Volume
 $v \approx 0.848 \text{ m}^3/\text{kg}_{da}$

Humidity Ratio
 $W \approx 8.2 \text{ g}/\text{kg}_{da}$
 (i.e., $0.0082 \text{ kg}/\text{kg}$)

Dry Bulb Temp
 $T = 22^\circ\text{C}$

Wet Bulb Temp
 $T_{wb} \approx 15.5^\circ\text{C}$

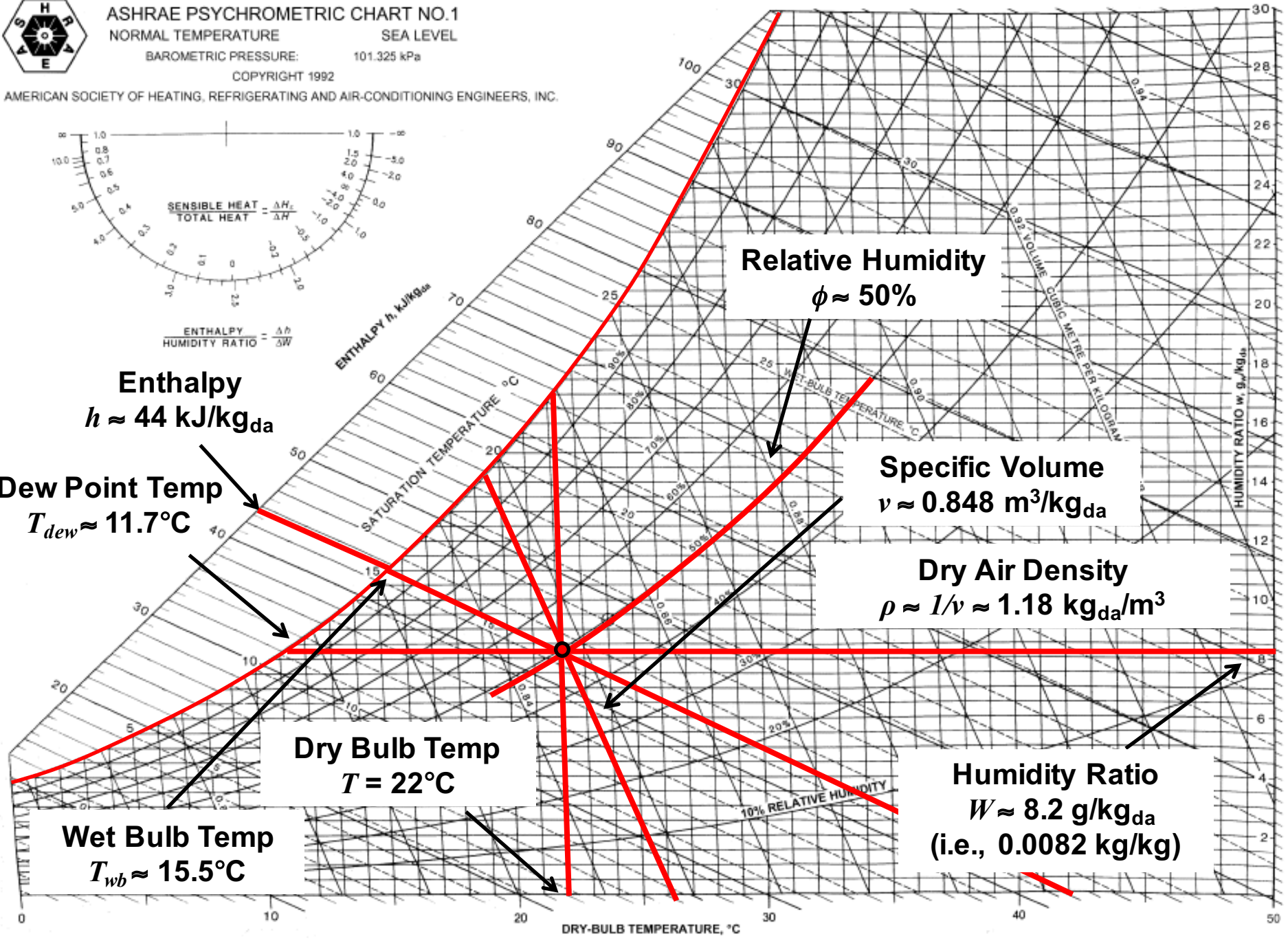
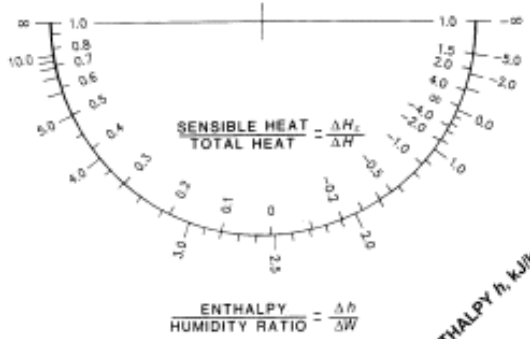
Dew Point Temp
 $T_{dew} \approx 11.7^\circ\text{C}$

Enthalpy
 $h \approx 44 \text{ kJ}/\text{kg}_{da}$



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 NORMAL TEMPERATURE SEA LEVEL
 BAROMETRIC PRESSURE: 101.325 kPa
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Relative Humidity
 $\phi \approx 50\%$

Specific Volume
 $v \approx 0.848 \text{ m}^3/\text{kg}_{da}$

Dry Air Density
 $\rho \approx 1/v \approx 1.18 \text{ kg}_{da}/\text{m}^3$

Humidity Ratio
 $W \approx 8.2 \text{ g}/\text{kg}_{da}$
 (i.e., 0.0082 kg/kg)

Enthalpy
 $h \approx 44 \text{ kJ}/\text{kg}_{da}$

Dew Point Temp
 $T_{dew} \approx 11.7^\circ\text{C}$

Dry Bulb Temp
 $T = 22^\circ\text{C}$

Wet Bulb Temp
 $T_{wb} \approx 15.5^\circ\text{C}$

Some psychrometric examples

Moist air exists at 30°C dry-bulb temperature with a 15°C dew point temperature

Find the following:

- (a) the humidity ratio, W
- (b) wet-bulb temperature, T_{wb}
- (c) enthalpy, h
- (d) specific volume, ν
- (e) relative humidity, ϕ



ASHRAE PSYCHROMETRIC CHART NO.1

NORMAL TEMPERATURE

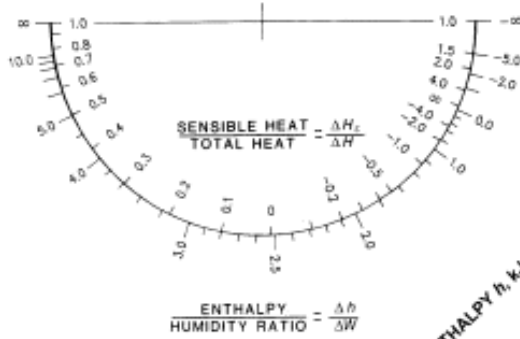
SEA LEVEL

BAROMETRIC PRESSURE:

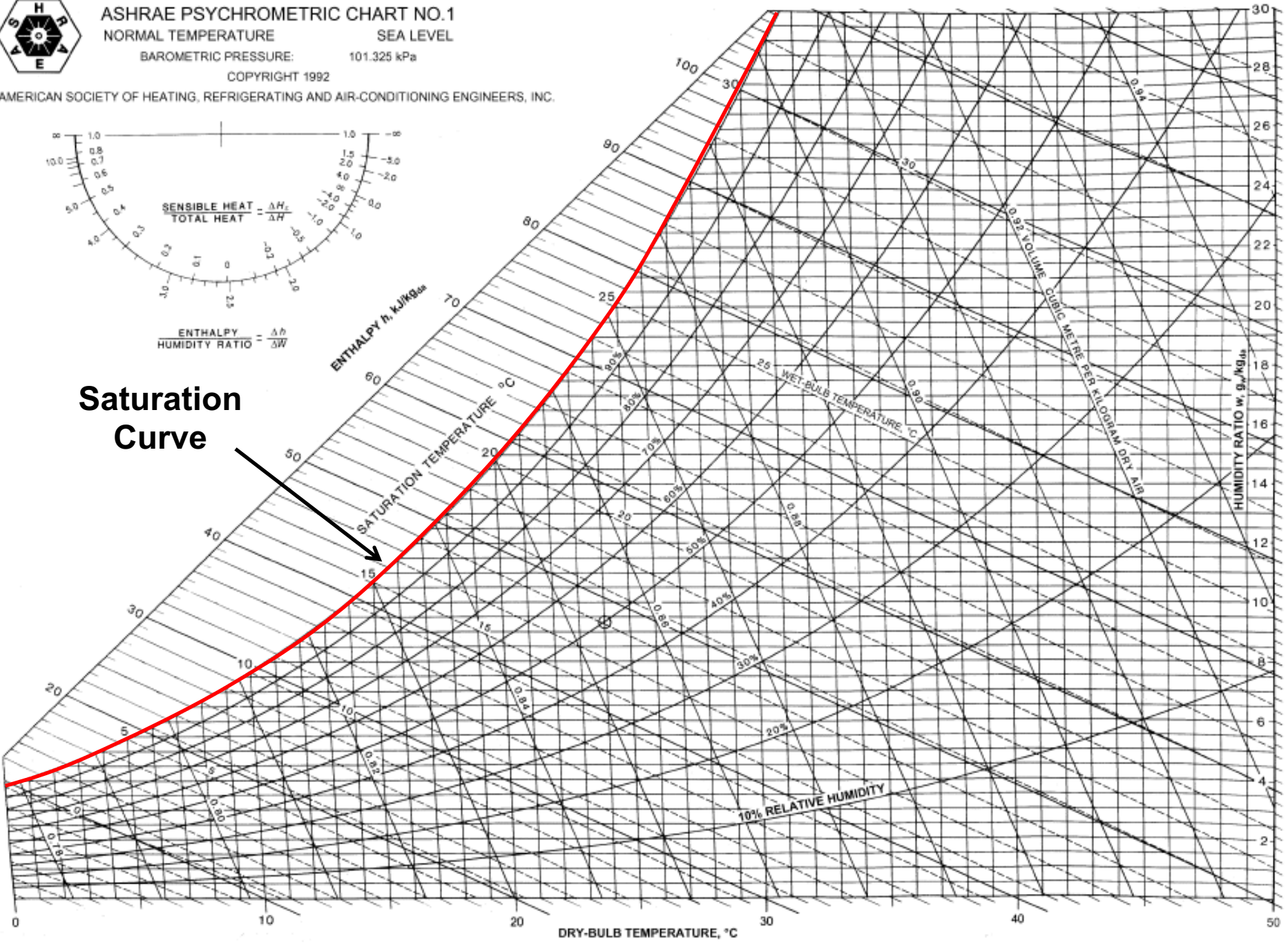
101.325 kPa

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Saturation Curve





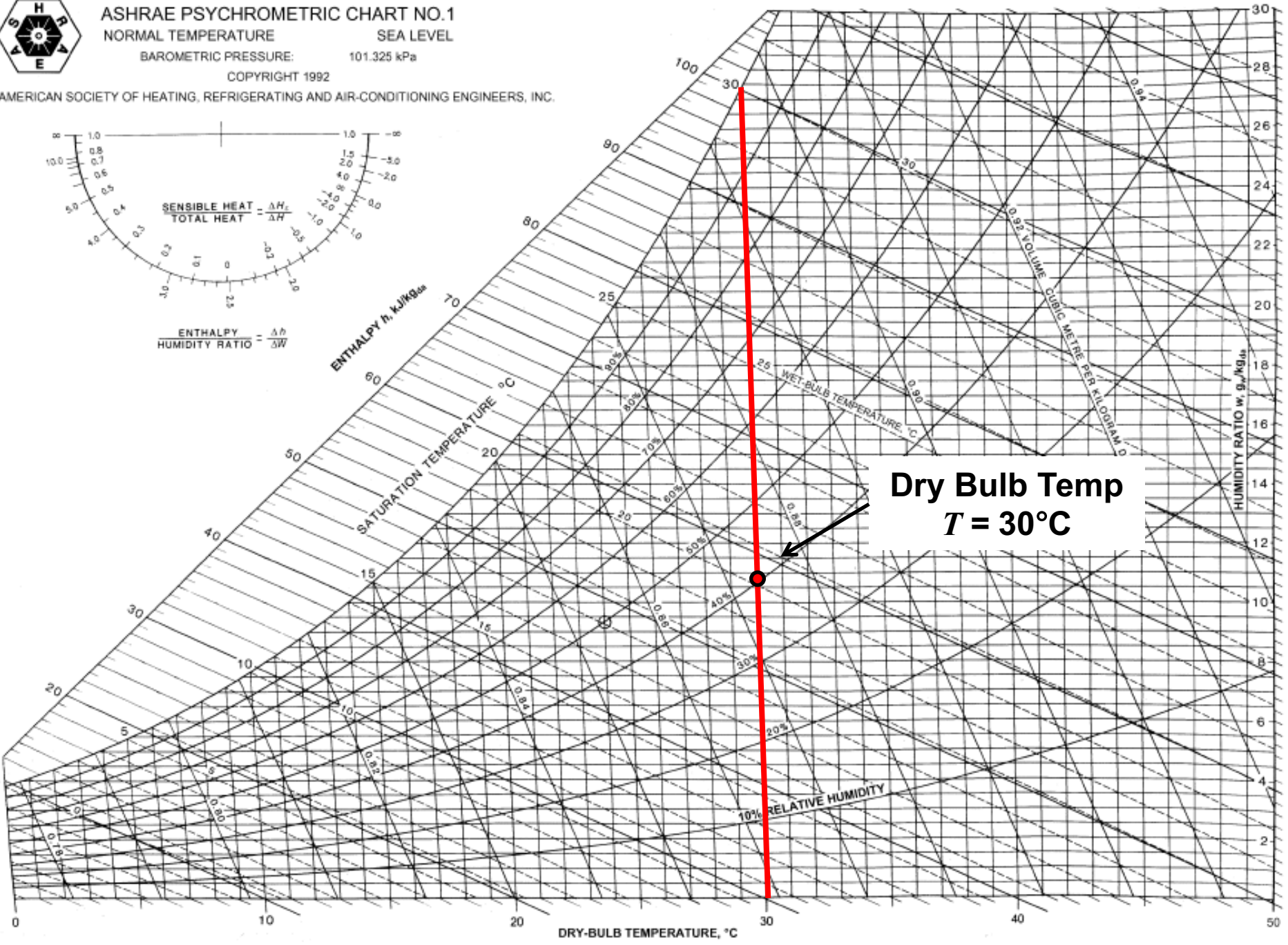
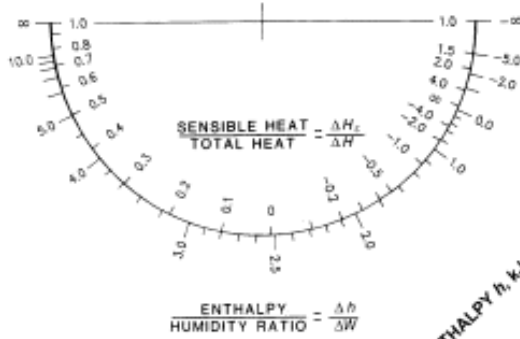
ASHRAE PSYCHROMETRIC CHART NO. 1

NORMAL TEMPERATURE SEA LEVEL

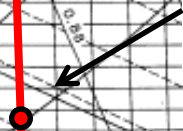
BAROMETRIC PRESSURE: 101.325 kPa

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Dry Bulb Temp
 $T = 30^{\circ}\text{C}$





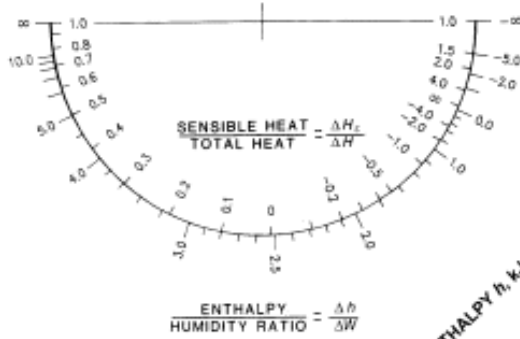
ASHRAE PSYCHROMETRIC CHART NO.1

NORMAL TEMPERATURE SEA LEVEL

BAROMETRIC PRESSURE: 101.325 kPa

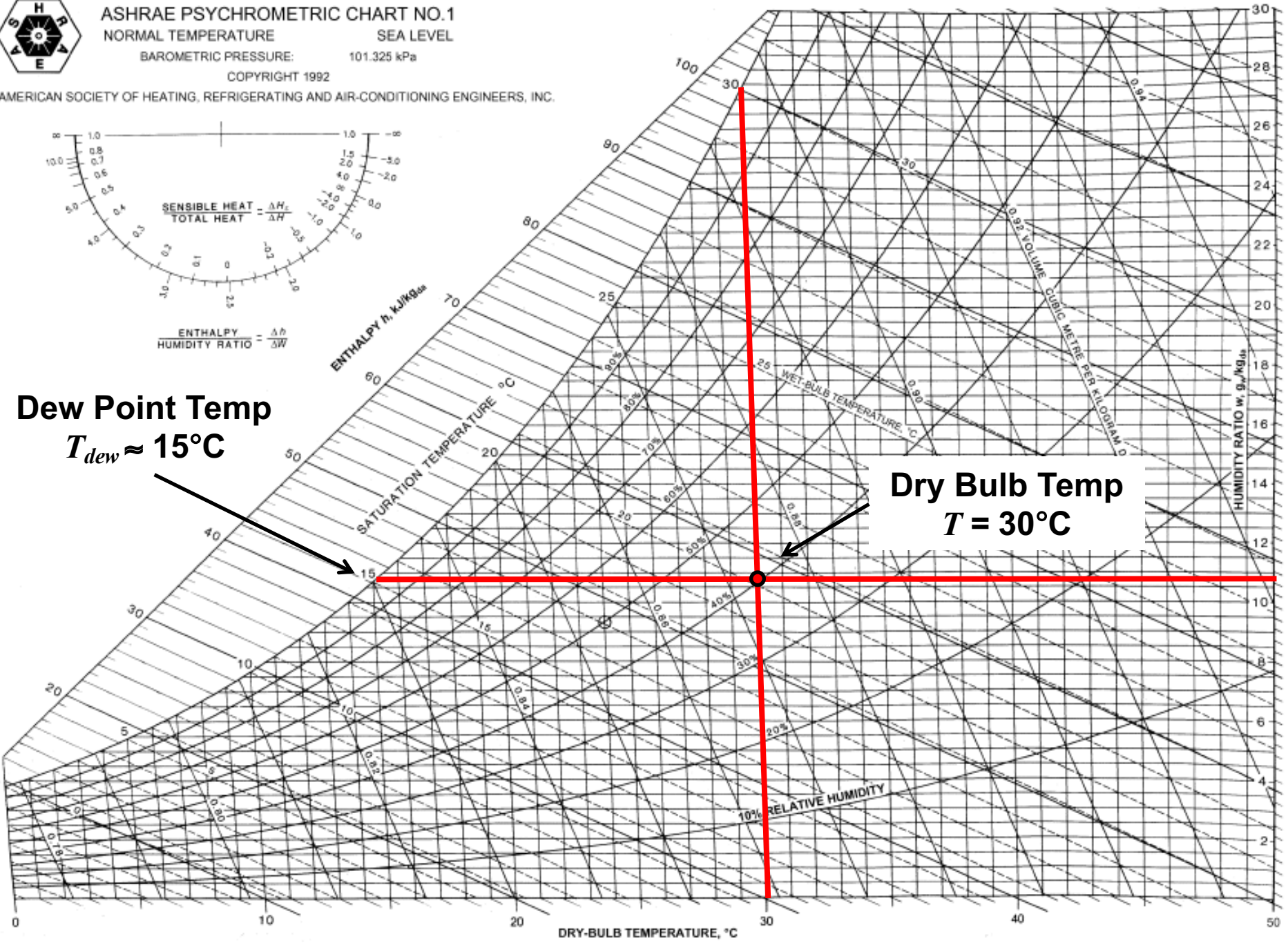
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Dew Point Temp
 $T_{dew} \approx 15^\circ\text{C}$

Dry Bulb Temp
 $T = 30^\circ\text{C}$





ASHRAE PSYCHROMETRIC CHART NO.1

NORMAL TEMPERATURE

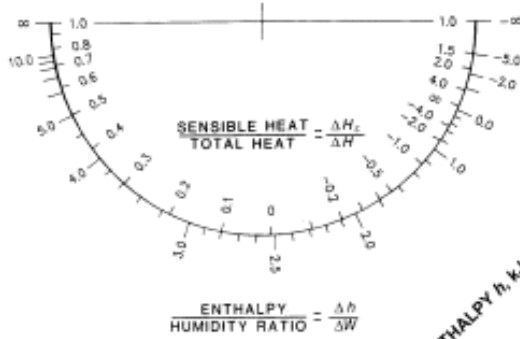
SEA LEVEL

BAROMETRIC PRESSURE:

101.325 kPa

COPYRIGHT 1992

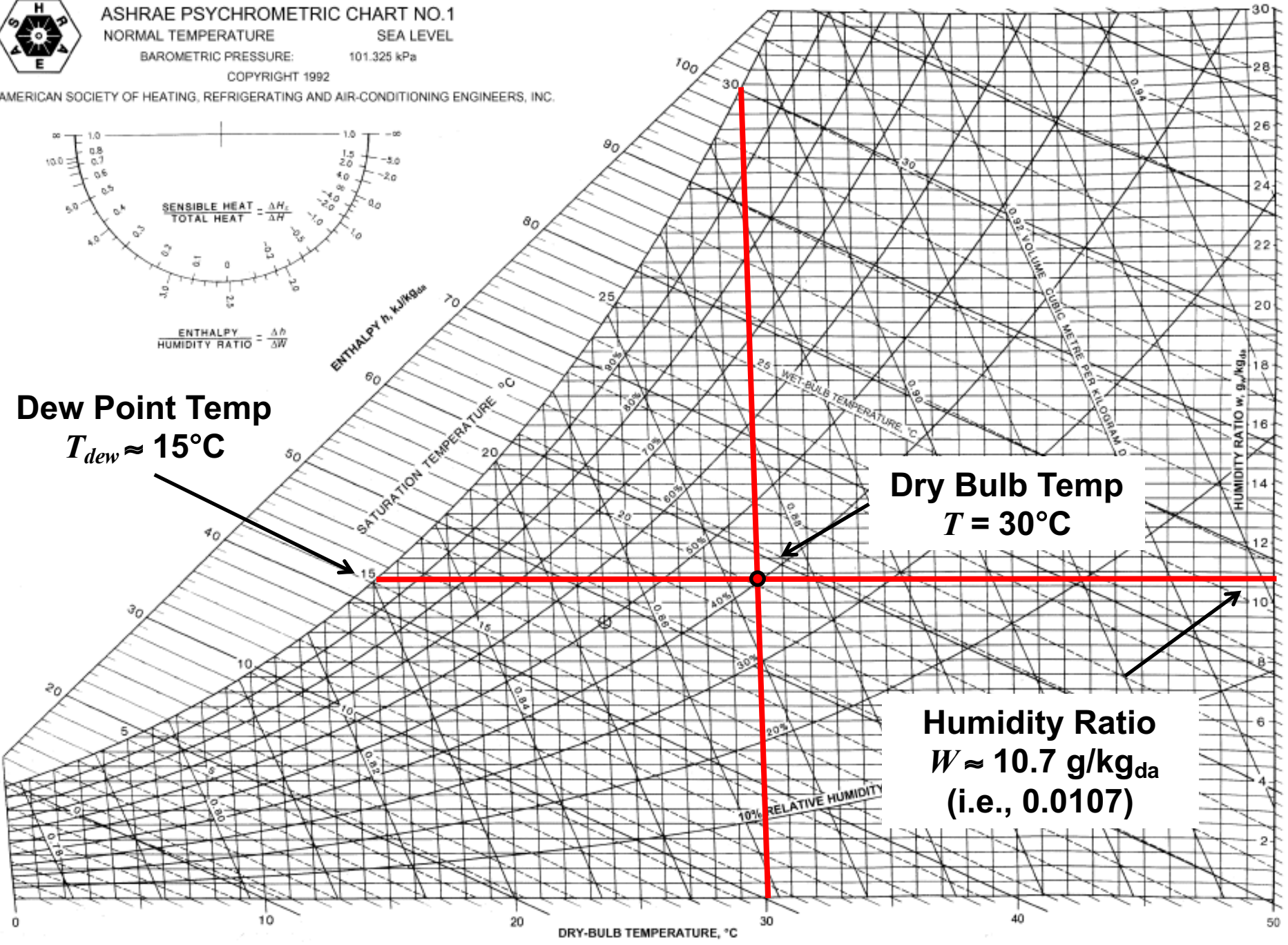
AMERICAN SOCIETY OF HEATING, REFRIGERATING AND AIR-CONDITIONING ENGINEERS, INC.



Dew Point Temp
 $T_{dew} \approx 15^\circ\text{C}$

Dry Bulb Temp
 $T = 30^\circ\text{C}$

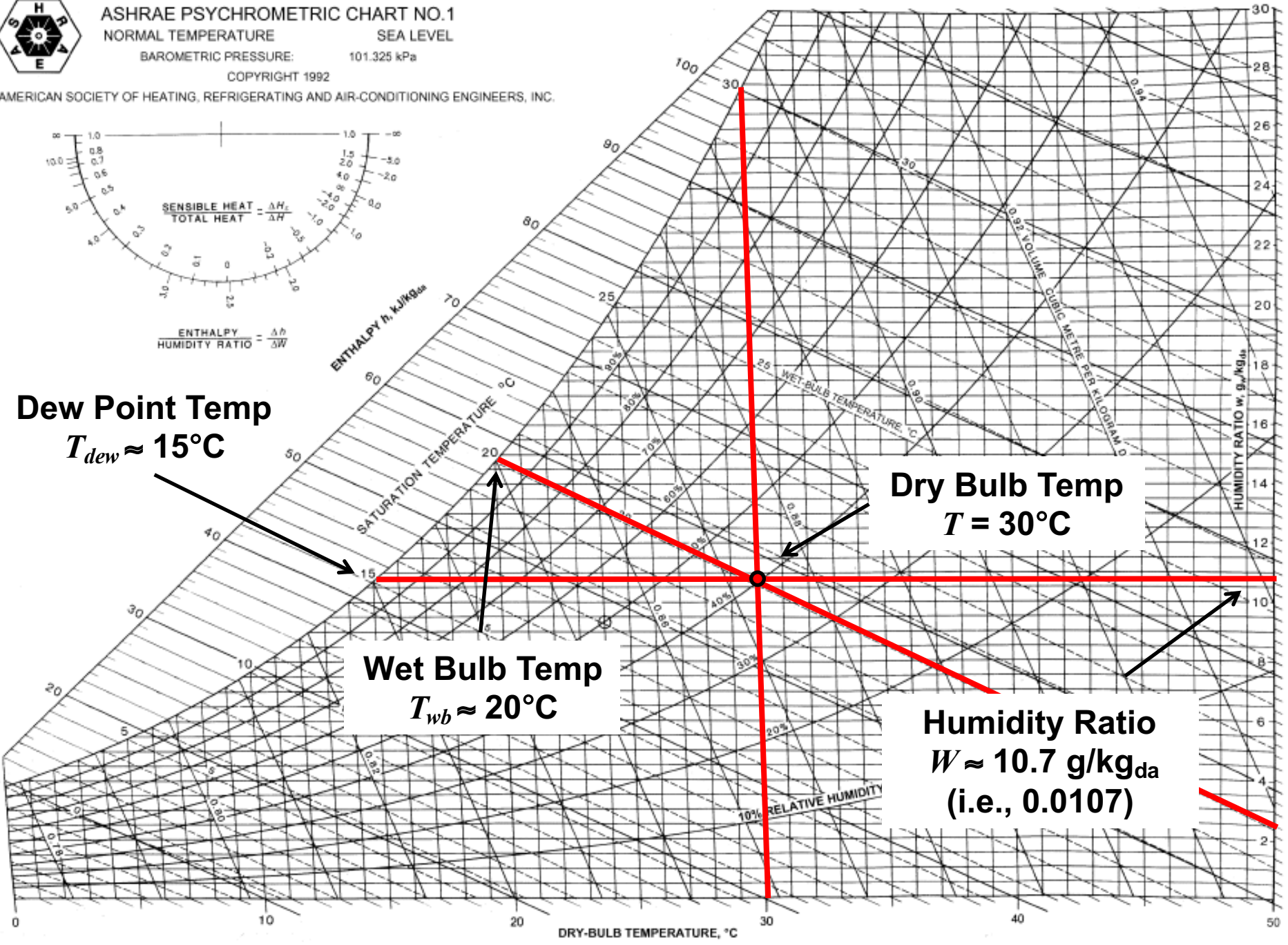
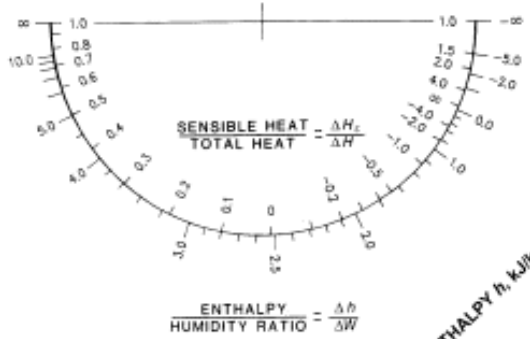
Humidity Ratio
 $W \approx 10.7 \text{ g/kg}_{da}$
(i.e., 0.0107)





ASHRAE PSYCHROMETRIC CHART NO.1
NORMAL TEMPERATURE
SEA LEVEL
BAROMETRIC PRESSURE: 101.325 kPa
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ASHRAE PSYCHROMETRIC CHART NO.1
 NORMAL TEMPERATURE SEA LEVEL
 BAROMETRIC PRESSURE: 101.325 kPa
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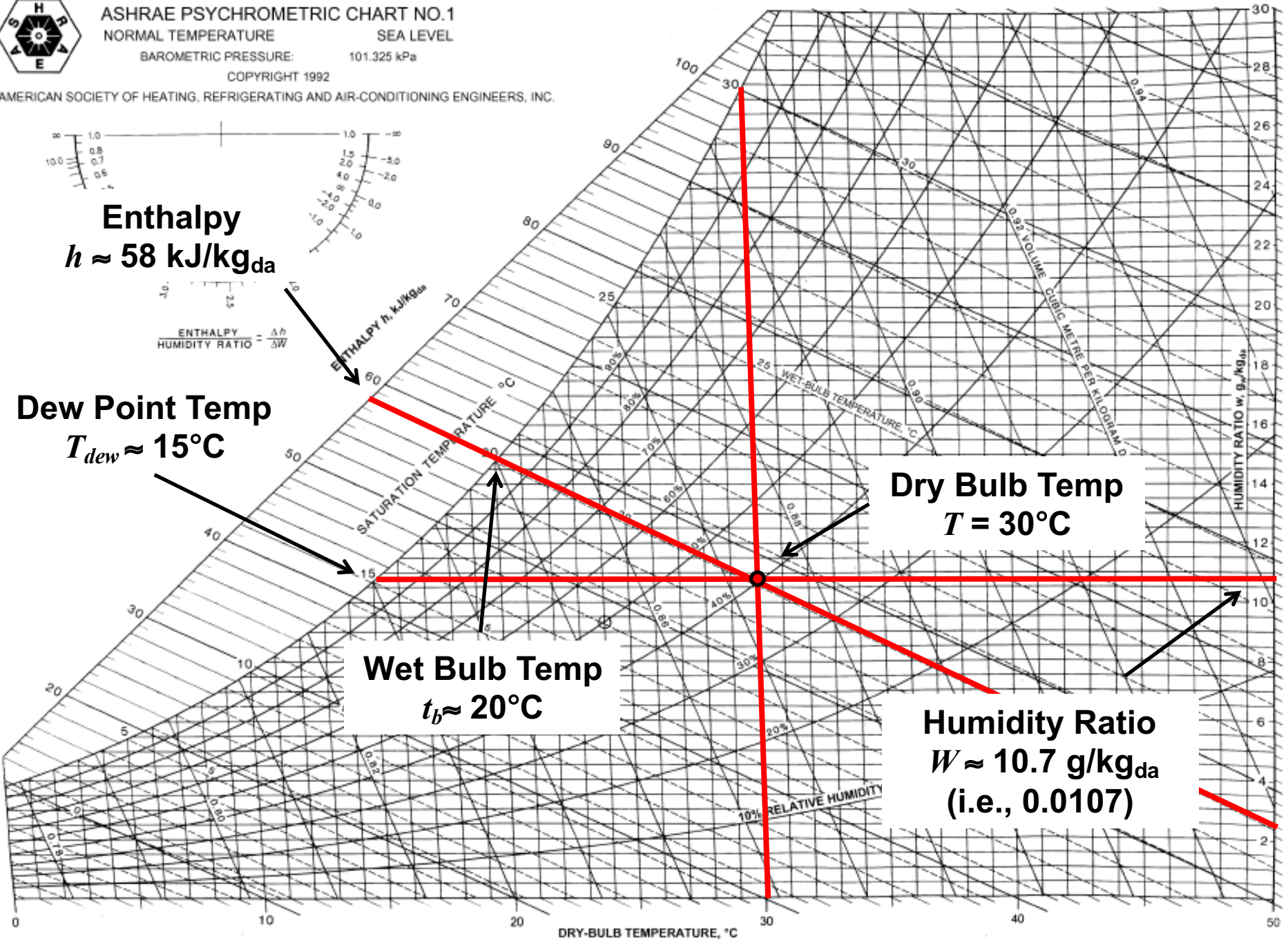
Enthalpy
 $h \approx 58 \text{ kJ/kg}_{da}$

Dew Point Temp
 $T_{dew} \approx 15^\circ\text{C}$

Wet Bulb Temp
 $t_b \approx 20^\circ\text{C}$

Dry Bulb Temp
 $T = 30^\circ\text{C}$

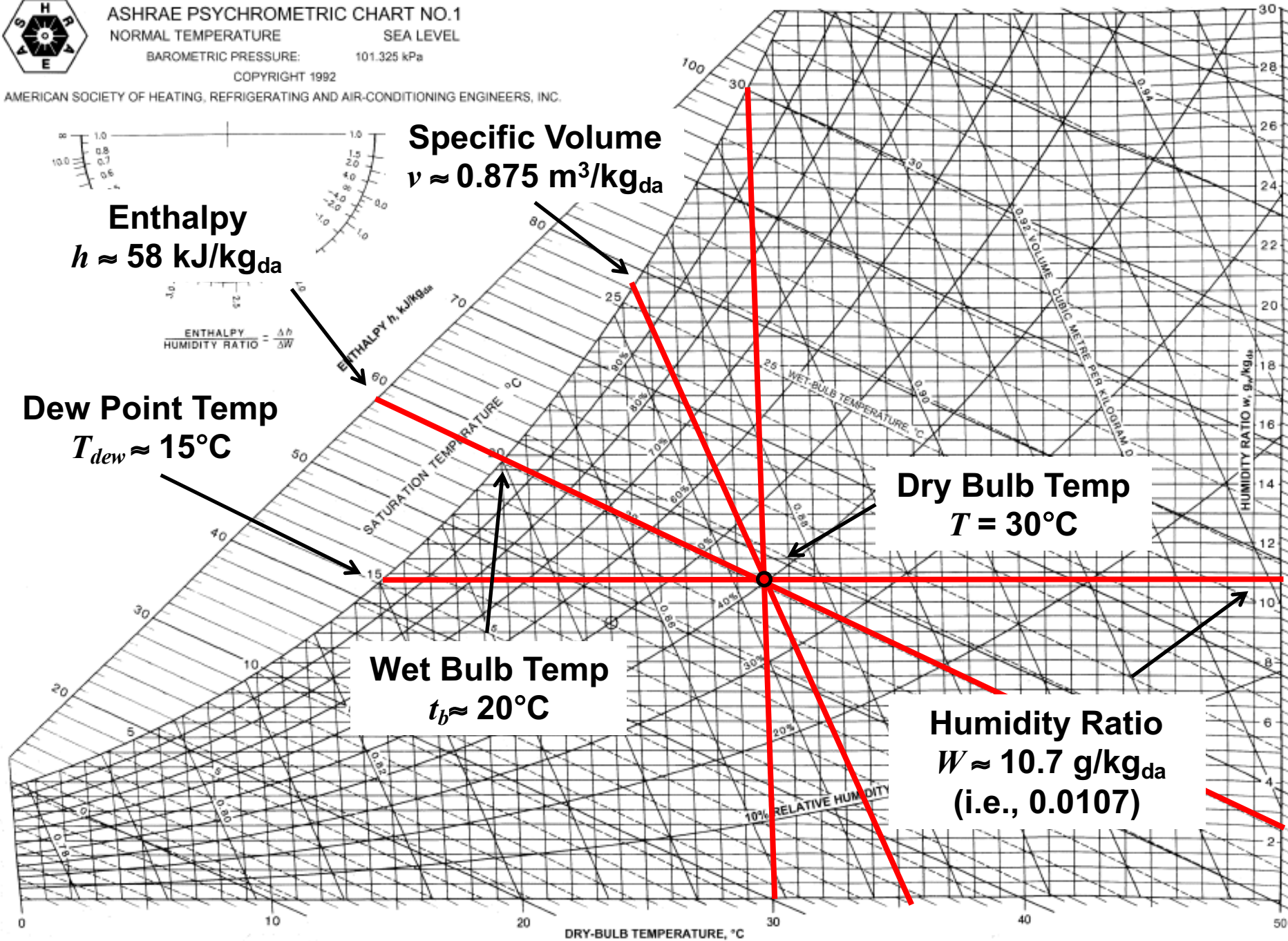
Humidity Ratio
 $W \approx 10.7 \text{ g/kg}_{da}$
 (i.e., 0.0107)





ASHRAE PSYCHROMETRIC CHART NO.1
NORMAL TEMPERATURE
SEA LEVEL
BAROMETRIC PRESSURE: 101.325 kPa
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Dew Point Temp
 $T_{\text{dew}} \approx 15^{\circ}\text{C}$

Specific Volume
 $v \approx 0.875 \text{ m}^3/\text{kg}_{\text{da}}$

Enthalpy
 $h \approx 58 \text{ kJ/kg}_{\text{da}}$

Dry Bulb Temp
 $T = 30^{\circ}\text{C}$

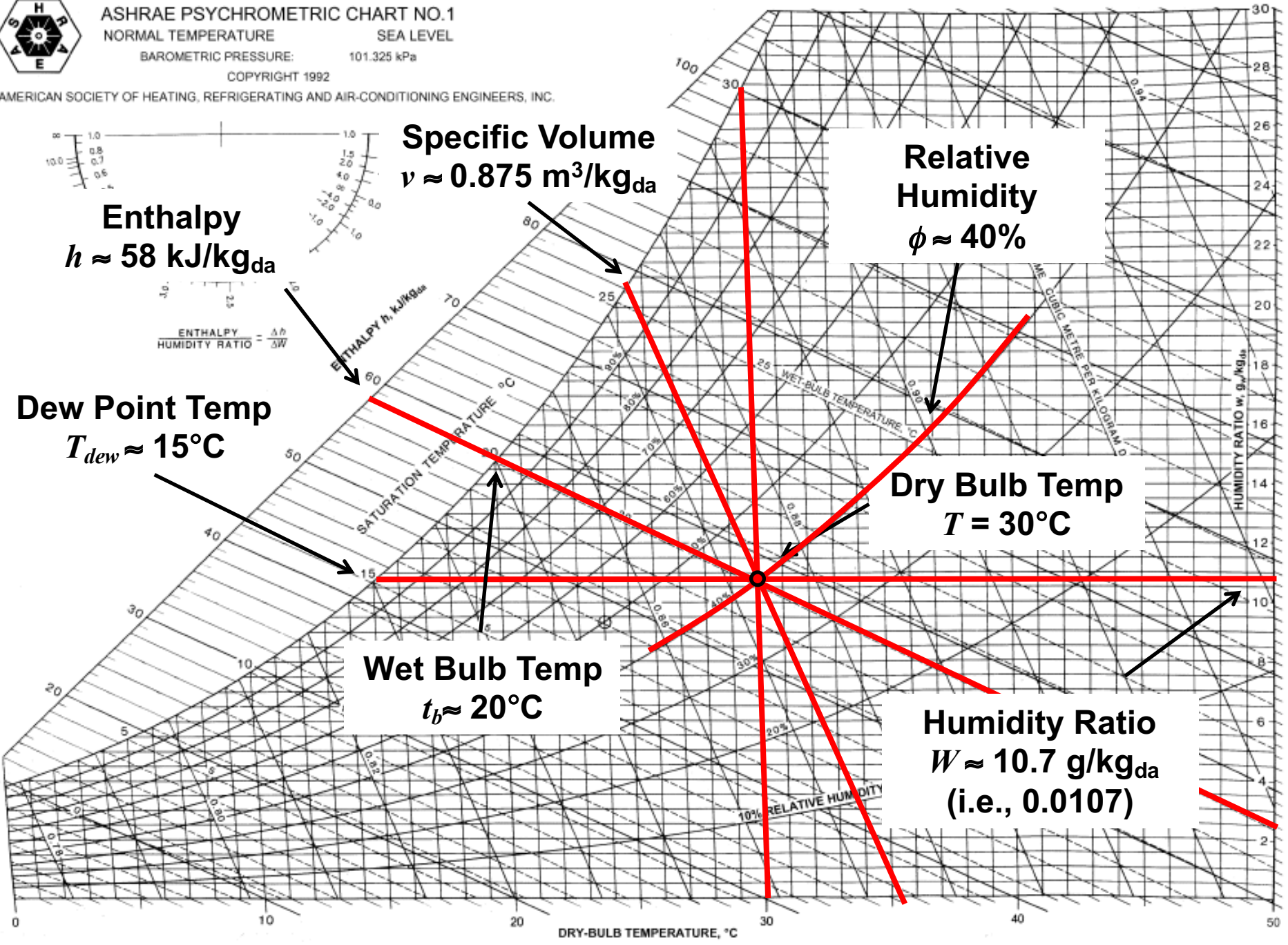
Wet Bulb Temp
 $t_b \approx 20^{\circ}\text{C}$

Humidity Ratio
 $W \approx 10.7 \text{ g/kg}_{\text{da}}$
(i.e., 0.0107)

$$\frac{\text{ENTHALPY}}{\text{HUMIDITY RATIO}} = \frac{\Delta h}{\Delta W}$$

DRY-BULB TEMPERATURE, $^{\circ}\text{C}$

HUMIDITY RATIO w , $\text{g}/\text{kg}_{\text{da}}$



Dew Point Temp
 $T_{dew} \approx 15^{\circ}\text{C}$

Enthalpy
 $h \approx 58 \text{ kJ/kg}_{da}$

Specific Volume
 $\nu \approx 0.875 \text{ m}^3/\text{kg}_{da}$

Relative Humidity
 $\phi \approx 40\%$

Dry Bulb Temp
 $T = 30^{\circ}\text{C}$

Wet Bulb Temp
 $t_b \approx 20^{\circ}\text{C}$

Humidity Ratio
 $W \approx 10.7 \text{ g/kg}_{da}$
 (i.e., 0.0107)

$$\frac{\text{ENTHALPY}}{\text{HUMIDITY RATIO}} = \frac{\Delta h}{\Delta W}$$

DRY-BULB TEMPERATURE, °C

HUMIDITY RATIO w , g/kg_{da}

SATURATION TEMPERATURE °C

WET-BULB TEMPERATURE °C

10% RELATIVE HUMIDITY

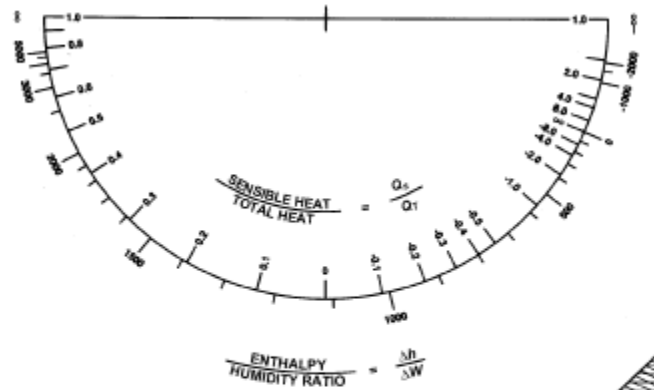
ONE CUBIC METRE PER KILOGRAM

Psychrometrics: IP units example

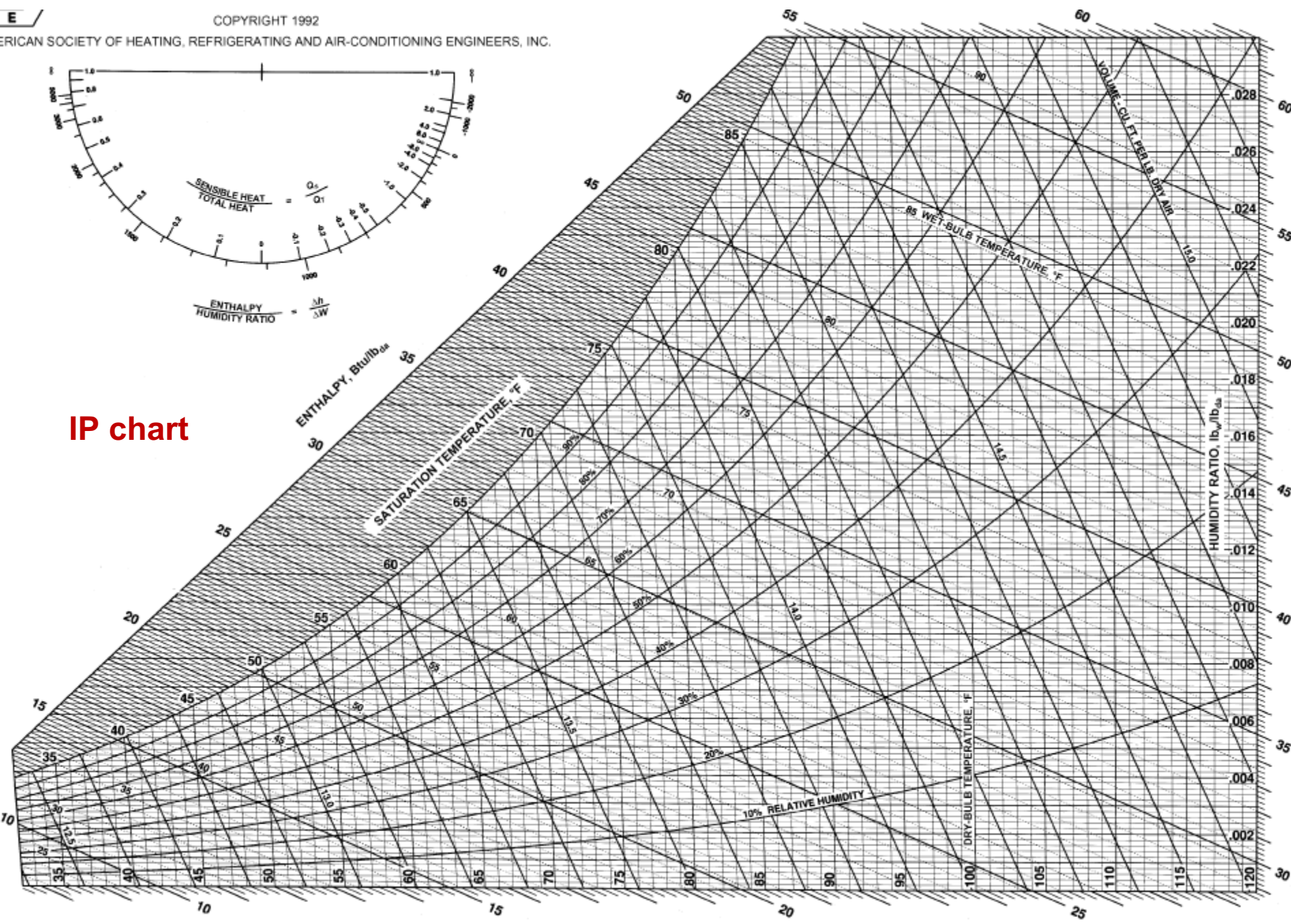
- Moist air exists at 100°F dry bulb, 65°F wet bulb and 14.696 psia

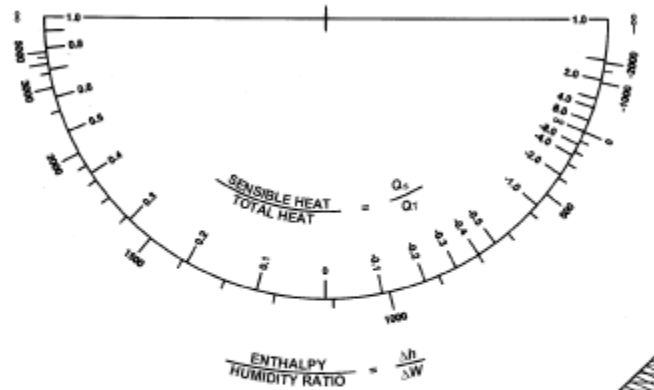
Find:

- a) Humidity ratio
- b) Enthalpy
- c) Dew-point temperature
- d) Relative humidity
- e) Specific volume

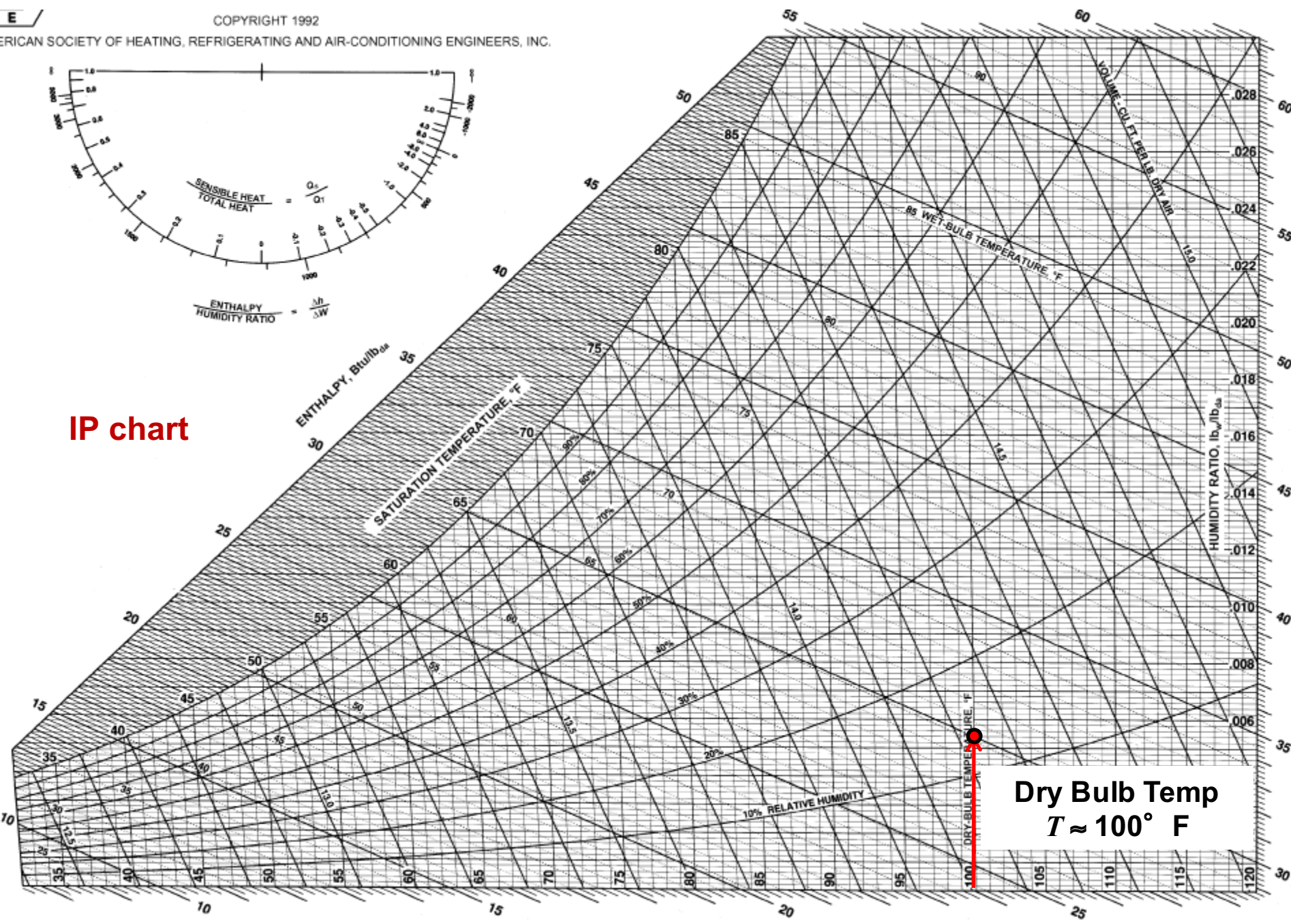


IP chart

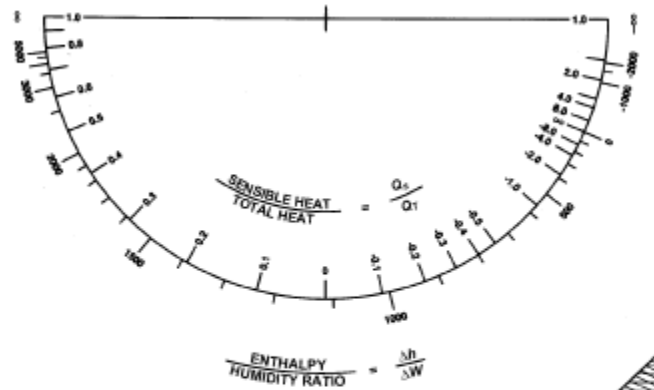




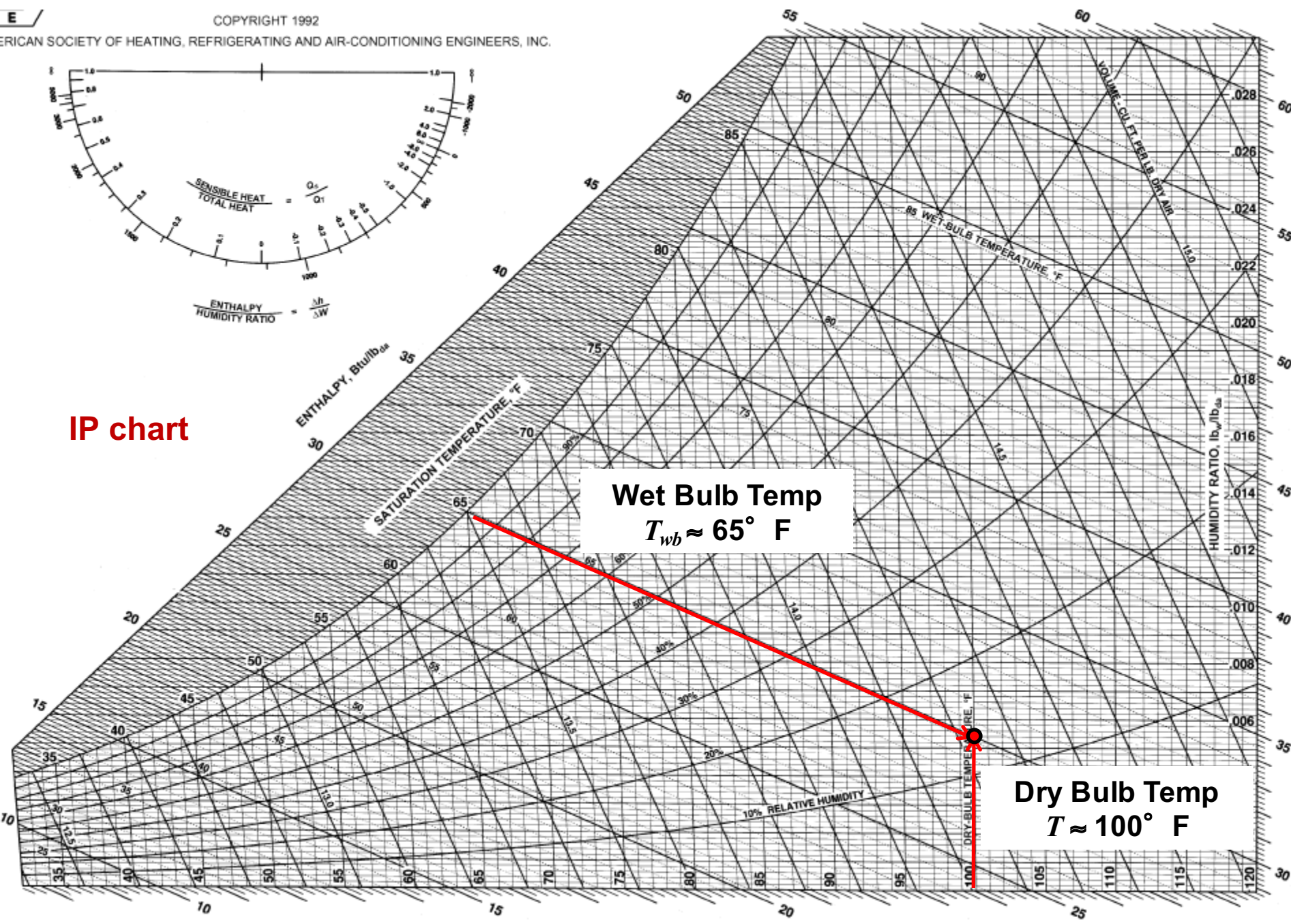
IP chart



Dry Bulb Temp
 $T \approx 100^\circ \text{ F}$

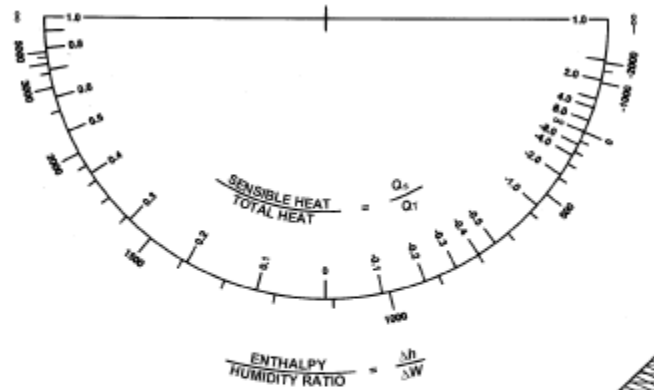


IP chart



Wet Bulb Temp
 $T_{wb} \approx 65^\circ \text{ F}$

Dry Bulb Temp
 $T \approx 100^\circ \text{ F}$

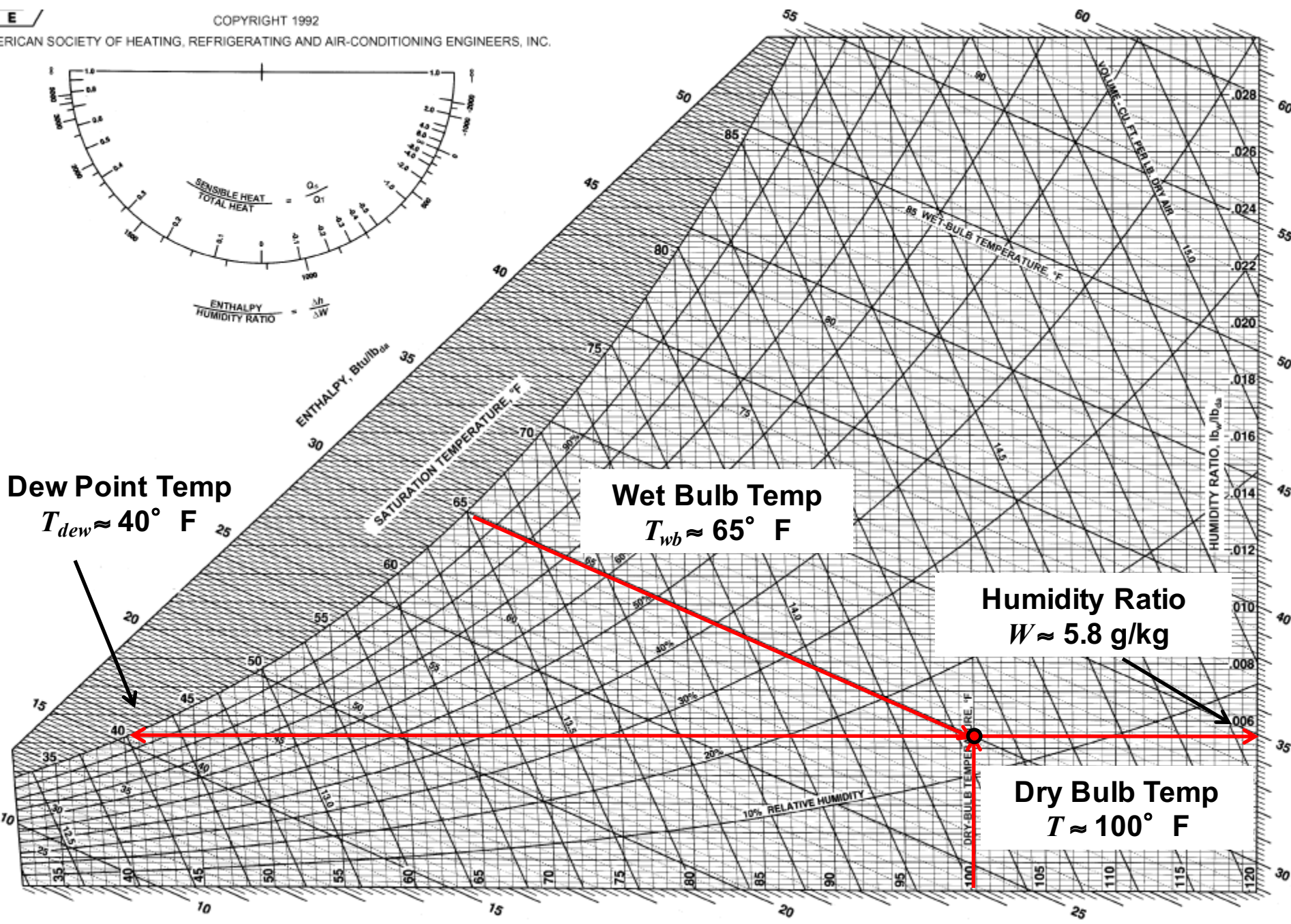


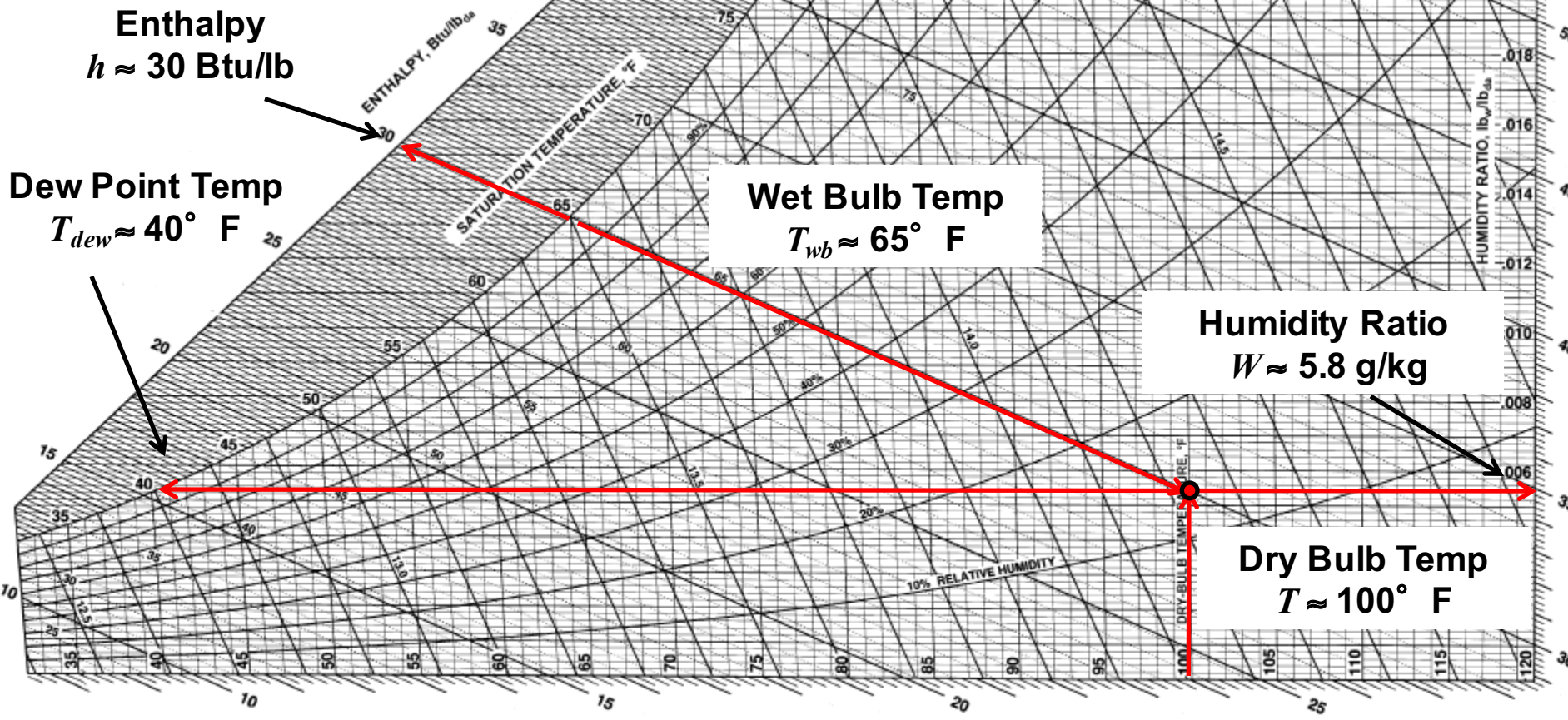
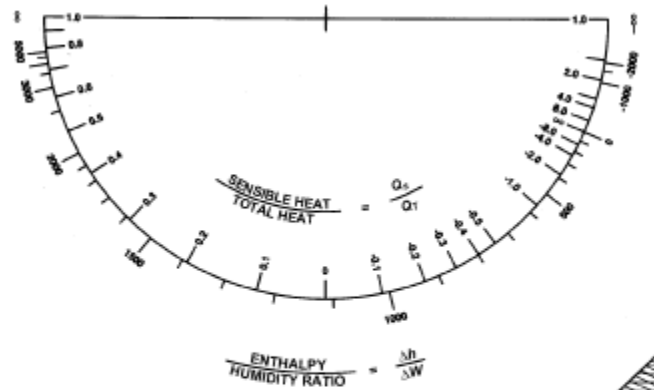
Dew Point Temp
 $T_{dew} \approx 40^\circ \text{ F}$

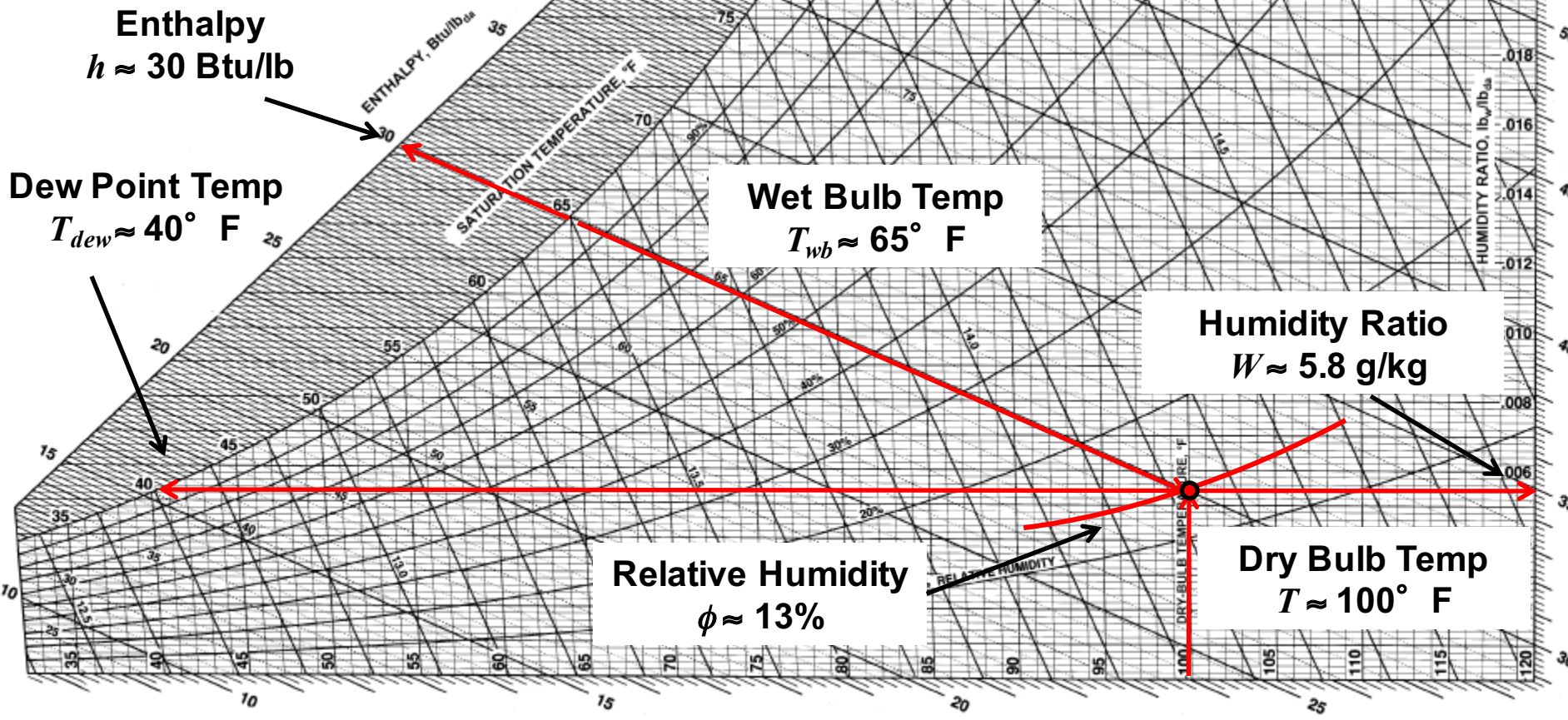
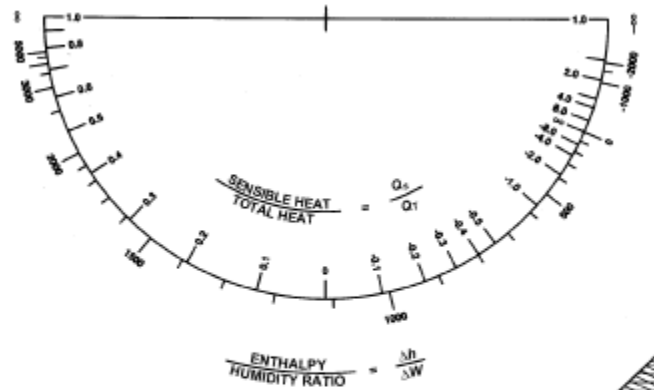
Wet Bulb Temp
 $T_{wb} \approx 65^\circ \text{ F}$

Humidity Ratio
 $W \approx 5.8 \text{ g/kg}$

Dry Bulb Temp
 $T \approx 100^\circ \text{ F}$







Enthalpy
 $h \approx 30$ Btu/lb

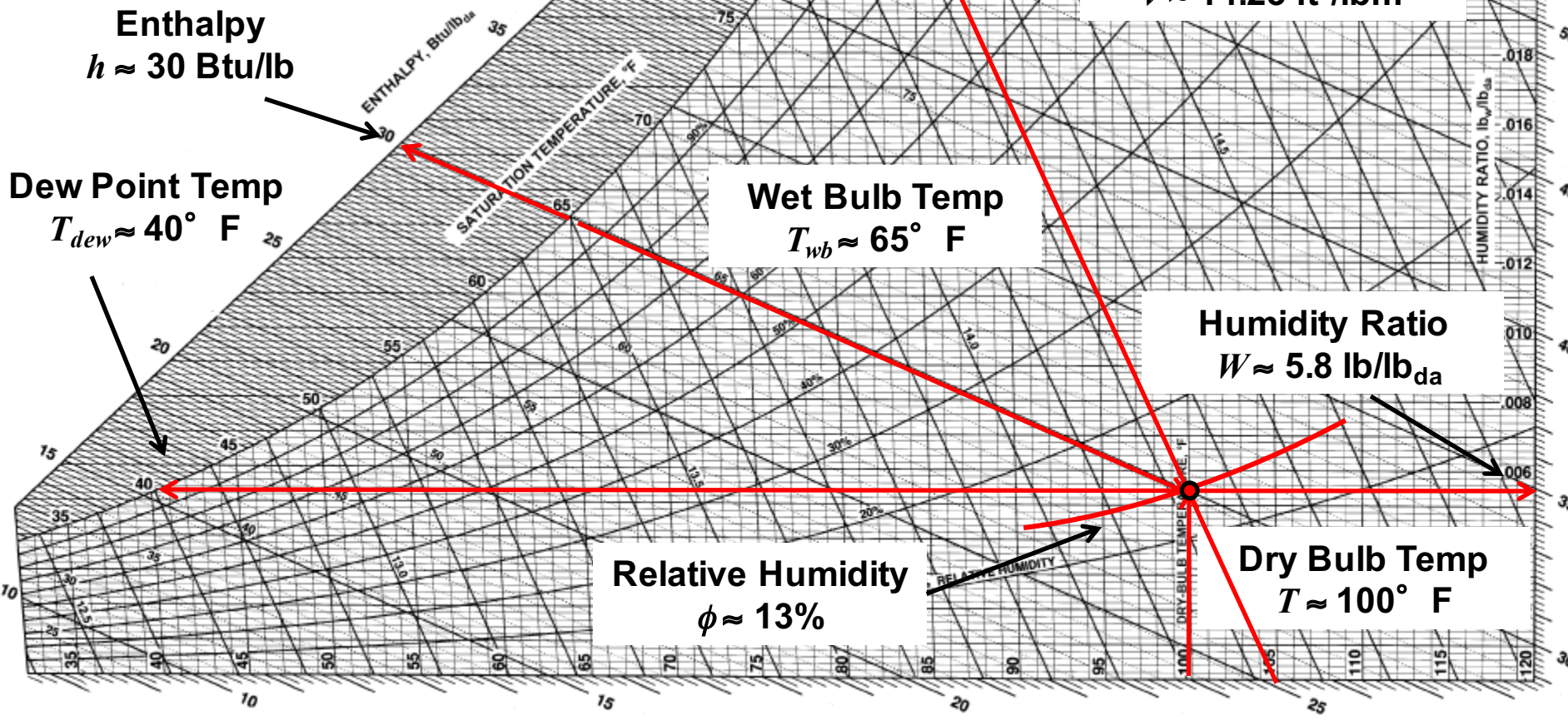
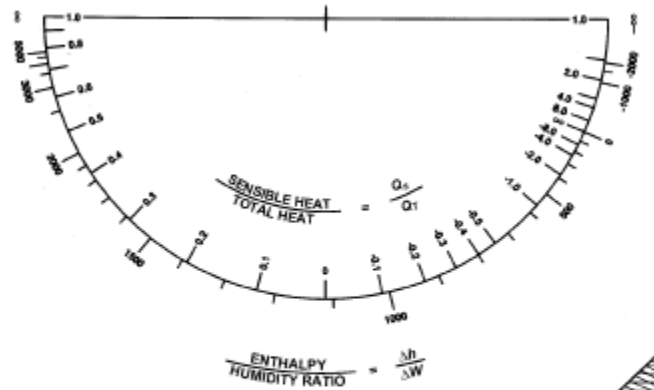
Dew Point Temp
 $T_{dew} \approx 40^\circ$ F

Wet Bulb Temp
 $T_{wb} \approx 65^\circ$ F

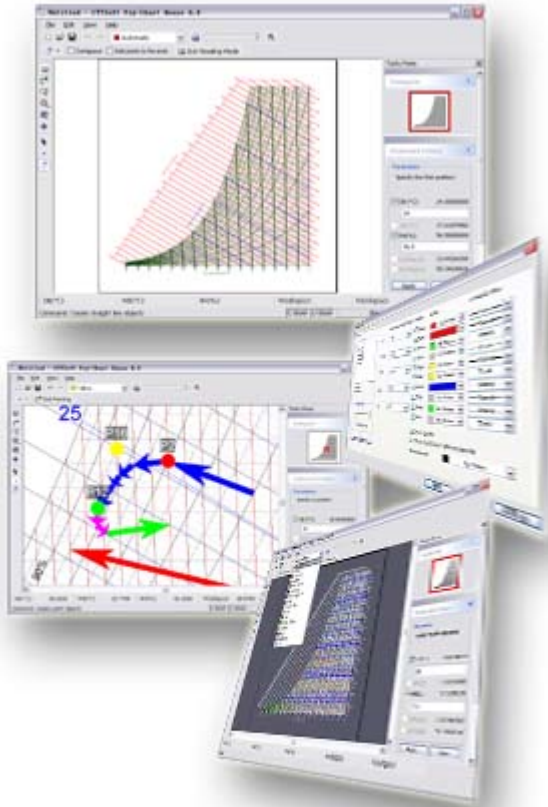
Humidity Ratio
 $W \approx 5.8$ g/kg

Relative Humidity
 $\phi \approx 13\%$

Dry Bulb Temp
 $T \approx 100^\circ$ F



Applying psychrometrics



- We can also use psychrometric charts or software
 - Psych and Psychpro
 - Very popular psych chart and analysis software
 - I think at least one of these is in the AM 217 lab
- There are a bunch of online calculators as well
 - <http://www.psychrometric-calculator.com>
 - <http://www.sugartech.co.za/psychro/>
 - <http://www.wolframalpha.com/examples/Psychrometrics.html>
- And smart phone apps too
- You can also make your own (i.e., in Excel)
 - You will have a HW problem where you have to do this

Psychrometrics also involves learning how to use and combine those quantities to determine things like sensible and latent heating and cooling loads (i.e., **processes**) (covered in a future lecture)